

#### Draft

# **Feasibility Study**

# Site 11 - School of Music Plating Shop

# Naval Amphibious Base Little Creek Virginia Beach, Virginia

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Prepared by



# **Executive Summary**

This report presents the Feasibility Study (FS) for Site 11, the former School of Music Plating Shop at Naval Amphibious Base (NAB) Little Creek. This FS is prepared by CH2M HILL under the Naval Facilities Engineering Command (NAVFAC), Atlantic Division Comprehensive Long-Term Environmental Action Navy II (CLEAN II) Contract N62470-95-D-6007, Contract Task Order (CTO) 0159, for submittal to NAVFAC, the United States Environmental Protection Agency (USEPA), and the Virginia Department of Environmental Quality (VDEQ).

Contamination at Site 11 consists of a volatile organic compound (VOC) groundwater plume including a residual source area (sorbed mass and aqueous phase contaminants) and a down-gradient plume. This site is located in the eastern portion of the base, near the intersection of Seventh and E Streets and consisted of the plating shop (Building 3651), an in-ground concrete tank used to neutralize plating solutions, and its associated piping. The tank and associated soil and piping have been removed.

This FS summarizes the nature and extent of the contaminated groundwater at Site 11, defines the remedial action objective (RAO), evaluates remedial action alternatives for the RAO, and identifies the applicable or relevant and appropriate requirements (ARARs). Following screening of groundwater treatment technologies of the source and plume area, the three alternatives retained for detailed evaluation and comparative analysis include:

Alternative 1 – No Action
Alternative 2 – Enhanced Reductive Dechlorination (ERD)
Alternative 3 – Electrical Resistance Heating (ERH) and Enhanced Reductive Dechlorination (ERD)

This FS provides a detailed analysis of each alternative against the National Oil and Hazardous Substance Pollution Contingency Plan (NCP) criteria followed by the comparative analysis of the remedial alternatives against one another. Alternative 1 is required by the NCP as a baseline. It does not meet the statutory requirements of the NCP and is not a viable remedial action for this site. In comparison to Alternative 2, Alternative 3 is more difficult and more costly to implement and has lower short-term effectiveness. Alternative 2, which would enhance the active biological degradation of site VOCs, meets the NCP criteria and was selected as the preferred alternative for Site 11.

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# **Acronyms and Abbreviations**

ARAR applicable or relevant and appropriate requirement

bgs below ground surface

°C degrees Celsius CD cyclodextrin

CERCLA Comprehensive Environmental Response, Compensation, and Liability Act

CFR Code of Federal Regulations

CLEAN II Comprehensive Long-Term Environmental Action Navy II

COPC contaminant of potential concern

CT central tendency CTO Contract Task Order

DCA dichloroethane DCE dichloroethene

DNAPL dense non-aqueous phase liquid

DPT Direct Push Technology

ERA ecological risk assessment ERH electrical resistance heating

ERD enhanced reductive dechlorination

ESTCP Environmental Security Technology Certification Program

FS Feasibility Study

ft feet

FWES Foster Wheeler Environmental Services

g gram

HHRA human health risk assessment HRC® Hydrogen Release Compound®

H<sub>2</sub> hydrogen

IAS Initial Assessment Study
IDW Investigation Derived Waste
IR Installation Restoration

IRI Interim Remedial Investigation

kg kilogram

L liter

LUCs land use controls

MCL maximum contaminant level

μg micrograms mg milligrams

MIP Membrane Interface Probe

NAB Naval Amphibious Base

NACIP Navy Assessment and Control of Installation Pollutants

NAVFAC Naval Facilities Engineering Command

NCP National Oil and Hazardous Substance Pollution Contingency Plan

O&M operations and maintenance

OMB Office of Management and Budget

PCP pentachlorophenol

PRG Preliminary Remediation Goals

PVC polyvinyl chloride

RAO remedial action objective RGH Rogers, Golden, and Halpern

RI Remedial Investigation ROD Record of Decision ROI radius of influence

RVS Round 1 Verification Step

SARA Superfund Amendments and Reauthorization Act

SERA screening ecological risk assessment
SRI Supplemental Remedial Investigation
SVOC semi-volatile organic compound
SWMU Solid Waste Management Unit

TBC to-be-considered
TCA trichloroethane
TCE trichloroethene
TOC total organic carbon
TOD total oxidant demand

USEPA United States Environmental Protection Agency

VC vinyl chloride

VDEQ Virginia Department of Environmental Quality

VFAs volatile fatty acids

VOC volatile organic compound

yr year

# Introduction and Background

This report presents the Feasibility Study (FS) for Site 11, the former School of Music Plating Shop at Naval Amphibious Base (NAB) Little Creek. This FS report is prepared by CH2M HILL under the Naval Facilities Engineering Command (NAVFAC), Atlantic Division Comprehensive Long-Term Environmental Action Navy II (CLEAN II) Contract N62470-95-D-6007, Contract Task Order (CTO) 0159, for submittal to NAVFAC, the United States Environmental Protection Agency (USEPA), and the Virginia Department of Environmental Quality (VDEQ). The FS is prepared in accordance with the process outlined in the Navy's Installation Restoration (IR) Program, which is consistent with the National Oil and Hazardous Substance Pollution Contingency Plan (NCP) and Section 120 of the Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA) as amended by the Superfund Amendments and Reauthorization Act (SARA).

Previous investigations have identified a groundwater plume containing volatile organic compounds (VOCs) associated with the former School of Music Plating Shop and neutralization tank. The nature and extent of contamination and human health risk assessment (HHRA) are documented in the Site 11 Supplemental Remedial Investigation (SRI) report (CH2M HILL, June 2004) and the Site 11 Revised HHRA, SRI Addendum (CH2M HILL, January 2006). There are no unacceptable ecological risks identified at Site 11 (CH2M HILL, June 2000). Additional soil and groundwater sampling and analyses were completed in 2005 as part of development of this FS; results of the 2005 investigations are documented herein.

The objectives of this FS are to evaluate remedial alternatives to prevent unacceptable risk exposure to groundwater and reduce the concentration of VOCs in groundwater to levels that allow for unlimited use and unrestricted exposure at Site 11. The FS develops and evaluates remedial alternatives to meet the remedial action objective (RAO) and identifies applicable or relevant and appropriate requirements (ARARs) and to-be-considered (TBC) criteria.

This FS report is composed of the following sections:

**Executive Summary** 

Section 1.0 - Introduction and Background

Section 2.0 - Remedial Action Objective and Applicable or Relevant and Appropriate Requirements

Section 3.0 -Screening of Remedial Technologies and Identification of Remedial Alternatives

Section 4.0 - Evaluation of Remedial Alternatives

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Section 6.0 - References

Figures and tables referenced within the text are provided at the end of the text. Appendices are provided at the end of the report.

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# 1.1 Site Description and History

NAB Little Creek is primarily an industrial facility and provides logistic facilities and support services for local commands, organizations, home-ported ships, and other units to meet the amphibious warfare training requirements of the Armed Forces of the United States. In addition to industrial land-use, NAB Little Creek is also used for recreational, commercial, and residential purposes. The location of NAB Little Creek is shown in Figure 1-1.

The area surrounding the 2,215-acre NAB is low lying and relatively flat with several fresh water lakes. Chubb Lake, Lake Bradford, Little Creek Reservoir/Lake Smith, and Lake Whitehurst are located on, or adjacent to, the base. Little Creek Reservoir/Lake Smith, located south of the base, serves as a secondary drinking water supply for parts of the city of Norfolk. NAB Little Creek is bordered by three saltwater bodies: Little Creek Cove, Desert Cove, and Little Creek Channel, which connects the coves with the Chesapeake Bay. The Chesapeake Bay borders the facility to the north.

#### 1.1.1 Site History

Site 11 is located in the eastern portion of the base, near the intersection of Seventh and E Streets (Figure 1-2). The site consisted of the plating shop (Building 3651), an in-ground concrete tank used to neutralize plating solutions, and its associated piping. The tank was approximately 10 feet (ft) east of the south corner of Building 3651. Use of the neutralization tank took place between 1964 and 1974. Small quantities of plating baths, acids, and lacquer strippers were disposed of down the sink in the plating shop which drains into the neutralization tank and eventually into the storm sewer system. Reportedly, 10 gallons of plating solutions were disposed in the shop sinks each year. There are no records of chlorinated solvents such as trichloroethene (TCE) being used at Site 11, however degreasing solvents such as TCE and 1,1,1-trichloroethane (TCA) have historically been associated with operations at similar plating shops.

The neutralization tank, piping, and surrounding soil were excavated in 1996 (Figure 1-2). Subsurface soil samples were taken from the excavation and groundwater samples were collected from the three existing monitoring wells to confirm the effectiveness of the removal action (Figure 1-3). Four VOCs were detected in groundwater above the maximum contaminant level (MCL). The maximum detected concentration of each VOC exceeding the MCL is:  $490 \text{ microgram } (\mu g)/\text{liter } (L) \text{ TCE}$ ,  $340 \text{ } \mu g/\text{L 1,1-dichloroethene}$  (DCE), and  $17 \text{ } \mu g/\text{L 1,1-dichloroethane}$  (DCA).

#### 1.1.2 Site Characteristics

The hydrogeologic setting at Site 11 includes the unconfined coastal plain sands and silts of the Columbia Aquifer that extends approximately 20 to 25 ft below ground surface (bgs). The water table ranges in depth from 5 to 7 ft bgs. The hydrogeology is depicted in cross-section on Figure 1-4. The Columbia Aquifer is underlain by a clay-confining unit (Yorktown Confining Unit) that ranges in thickness from 30 to 40 ft. The confined Yorktown Aquifer underlies the confining clay and extends to a depth of 280 ft in the area of NAB Little Creek (Meng and Harsh 1988). As evidence by the general absence of VOCs detected in the Yorktown Aquifer and the low vertical permeability of the confining clay (between

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 $1.56 \times 10^{-8}$  and  $3.0 \times 10^{-7}$  centimeters/second) there is little risk of contamination moving from the Columbia Aquifer to the Yorktown Aquifer.

Groundwater flow in the Columbia Aquifer near Site 11 is generally east to west, but is locally influenced by a sanitary sewer system paralleling Gator Boulevard (Figure 1-2), where groundwater flow immediately north of the sewer line is to the south and flow direction immediately south of the sewer line is to the north (CH2M HILL, June 2004). Groundwater gradients are relatively flat. The average groundwater flow velocity in the Columbia Aquifer at Site 11 has been calculated to be approximately 110 ft/year (yr). Groundwater flow in the Yorktown Aquifer is to the northwest, toward the Chesapeake Bay (CH2M HILL, June 2004).

# 1.2 Previous Investigations

A summary of previous investigations at NAB Little Creek is provided in Table 1-1. NAB Little Creek initiated environmental investigation efforts under the Navy Assessment and Control of Installation Pollutants (NACIP) Program by conducting an Initial Assessment Study (IAS) in 1984 followed by a Round 1 Verification Step (RVS) in 1986.

An Interim Remedial Investigation (IRI) was completed in 1991 and a Remedial Investigation (RI)/FS report was completed in 1993. Subsequent to the RI/FS, a decision document was issued in November 1994 (FWES, November 1994a), proposing removal of the neutralization tank, associated piping, and neighboring surface and subsurface soil. The neutralization tank, piping, and surrounding soil were excavated in 1996. An Interim Removal Action closeout report was completed in 1996 (IT Corporation, May 1996). The results of post-removal action sampling are documented in Final Groundwater Monitoring Report, Sites 5 and 11, Naval Amphibious Base Little Creek, Virginia Beach, Virginia (CH2M HILL, February 1998). Additional groundwater sampling was recommended to further define the extent of VOCs in groundwater.

#### **Ecological Risk Assessment (ERA) 2000**

A screening ecological risk assessment (SERA) for Site 11 was completed in June 2000 (CH2M HILL, June 2000). The SERA concluded potential ecological risks at Site 11 are negligible based on the lack of complete and significant exposure pathways, and no further action was recommended for ecological resources.

#### **Delineation Investigations 2001-2003**

A Membrane Interface Probe (MIP) investigation was conducted in 2001 to further characterize the extent of VOCs in groundwater. Direct-push samples for off-site laboratory analysis were collected to confirm the MIP results. The results indicated that there had not been significant degradation of TCE (CH2M HILL, June 2004).

An Environmental Security Technology Certification Program (ESTCP) funded pilot test was conducted at Site 11 in 2002 to evaluate the *in situ* removal of organic contaminants from groundwater through the injection and extraction of a cyclodextrin (CD) solution (Boving et al., 2003). Six wells were installed for this study and follow-up groundwater sampling was completed in January 2003. A second MIP investigation was conducted in September 2003 to further assess the efficacy of the CD solution on the groundwater at the

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site. The field activities and findings associated with these 2003 investigations are documented in Technical Memorandums "Summary of Site 11 Cyclodextrin Pilot Study Post-Treatment Groundwater Sampling", NAB Little Creek, Virginia Beach Virginia (CH2M HILL, July 2003), and "NAB Little Creek Sites 11, 11a, and 13 Membrane Interface Probe Investigation and Confirmation Sampling" (CH2M HILL, November 2003).

#### Supplemental Remedial Investigation (SRI) 2004

A SRI was completed in 2004 that incorporated data from 1996 through 2001. The SRI identified three inorganic contaminants of potential concern (COPCs) in surface soil (iron, manganese, and thallium) and two inorganic COPCs in groundwater (iron and chromium). Additionally, one semi-volatile organic compound (SVOC) [pentachlorophenol (PCP)] and two chlorinated VOCs (TCE and 1,1-DCE) were identified as COPCs in groundwater. The SRI concluded that VOCs in groundwater are limited to the lower portion of the Columbia Aquifer in the area of the former plating shop neutralization tank and extend south beneath the School of Music building to Gator Boulevard.

The SRI HHRA was completed for Site 11 based on data collected in 1998 and 1999. Groundwater samples collected following the CD pilot study (2003 through 2005) indicated degradation of parent VOCs. To effectively evaluate remedial action alternatives in this FS, human health risks associated with exposure to VOCs in groundwater were reevaluated and are documented in the *Site 11 Revised HHRA SRI Addendum* (CH2M HILL, January 2006). Based on background concentrations and the calculated potential risk from central tendency (CT) exposures, the Navy in partnership with the VDEQ and USEPA determined there were no unacceptable human health risk associated with exposure to inorganic constituents in groundwater and soil at Site 11 (CH2M HILL, January 2006). PCP was retained as an SVOC COPC. A summary of the VOC COPCs posing potential unacceptable risk to be addressed by this FS are summarized in Table 1-2.

#### **Vapor Intrusion Investigation 2005**

To address potential vapor intrusion of VOCs from groundwater into the School of Music (Building 3602), a site visit was conducted and groundwater samples from the top of the water table aquifer and a water sample from the basement sump for VOC analyses were collected in May 2005. This effort concluded that there are limited pathways for soil gas to intrude into the building as the first floor was under a positive pressure relative to the basement mechanical room, and there were no VOCs detected in six of the eight shallow groundwater samples. Only chloromethane (1.7  $\mu$ g/L) and TCE (6.3  $\mu$ g/L) were detected at very low concentrations. There were no VOCs detected in the sample collected from the basement sump. VOC concentrations at the top of the water table are well below risk screening levels determined using the Johnson and Ettinger model. The vapor intrusion assessment concluded even in the event of conditions promoting vapor intrusion, concentrations of VOCs in groundwater will not represent unacceptable human health risks from vapor intrusion inside the School of Music building. Results are presented in *Vapor Intrusion Assessment*, *Site 11, Naval Amphibious Base Little Creek* (CH2M HILL, 2005; Appendix A).

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#### **Pre-Feasibility Study Investigations 2005**

Groundwater sampling for VOC analysis was conducted at Site 11 in March 2005 to support evaluation of remedial action alternatives. Additional soil and groundwater sampling was completed in October 2005 to further support analysis of remedial action alternatives for the FS. The sampling protocol and results are provided in Appendix B.

Total VOC concentrations in groundwater samples collected from the Columbia Aquifer exceeded 100,000  $\mu$ g/L in the area of the former neutralization tank. Figure 1-5 illustrates the total VOC concentrations in groundwater and identifies highest concentration source area. Although DNPL was not identified, individual VOC concentrations in soils in the source area exceeded 10,000  $\mu$ g/kilogram (kg) at the top of the Yorktown Confining Unit; the greatest concentration was 25,000  $\mu$ g/kg of TCE. In the lower portion of the Columbia Aquifer the maximum individual VOC concentration in soil was 600  $\mu$ g/kg of cis-1,2-DCE. In the upper portion of the Columbia Aquifer the maximum individual VOC concentration in soil was 55  $\mu$ g/kg of TCE.

Microbial analysis verified the presence of healthy microbial populations capable of biodegradation of chlorinated VOCs. Total oxidant demand (TOD) was analyzed using sodium persulfate as the oxidant. Results ranged from 1.9 to 3.7 gram (g)/kg of sodium persulfate in the lower portion of the Columbia Aquifer and 11 to greater than 19.5 g/kg of sodium persulfate in the Yorktown Confining Unit. These values were not unexpected based on the elevated concentrations of total organic carbon (TOC) in soil (Appendix B). Geotechnical analysis including soil characterization, grain size, moisture content, and porosity was also completed; the results are provided in Appendix B.

#### 1.3 Nature and Extent of Contamination

The former Plating Shop neutralization tank was the source of VOCs in groundwater. The neutralization tank, associated piping, and surrounding soil have been removed. VOCs released from the former neutralization tank migrated into the subsurface and were further transported through the groundwater system via dissolution, advection, and dispersion. Groundwater flow is towards the south and southeast, and is influenced by a leaking sanitary sewer line along Gator Boulevard. The current groundwater infiltration rate is approximately 10 gallons per minute (CH2M HILL, June 2004). A detailed evaluation of the site conceptual model, including nature and extent, and contaminant fate and transport, is documented in the *Supplemental Remedial Investigation for Site 11* (CH2M HILL, June 2004).

Groundwater contamination at Site 11 includes a residual source area (sorbed mass and aqueous phase contaminants) and a down-gradient plume consisting predominantly of dissolved-phase contaminants (Figure 1-5). Residual dense non-aqueous phase liquid (DNAPL) may be indicated if dissolved phase concentrations are greater than or equal to 1% of the maximum aqueous solubility. Since the CD pilot test, no TCE or 1,1,1-TCA concentrations have been detected in groundwater samples above 1% of its maximum aqueous solubility. Because cis-1,2-DCE was never used at the site in pure form and has only been detected at concentrations of 1% of its maximum aqueous solubility following the CD pilot study, its presence is likely due to the degradation of TCE and not the presence of DNAPL.

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TCE was detected at a concentration above its 1% of maximum solubility level (11,000  $\mu$ g/L) in well LS11-MW5D once in 1998. Subsequently in 1999 and later, TCE concentrations in groundwater samples from this well were significantly below 11,000  $\mu$ g/L. This suggests that only a dissolved phase plume is present in this area (i.e., no DNAPL). Based on these groundwater data and the site history, the area selected for source area remedial action consideration includes the area treated during the CD pilot test. This area is shown in Figure 1-5.

The target depth interval for the remedial action is the bottom 3 to 5 ft of the shallow surficial aquifer, just above the clay layer present at the site at approximately 21 to 23 ft bgs. Previous groundwater sampling has shown that the groundwater contamination is highly stratified, with the interval just above the clay containing the greatest concentrations of contaminants. Shallower groundwater contains much lower VOC concentrations.

Prior to implementation of a remedial action, the sanitary sewer intercepting groundwater flow will be repaired. Following repair of the sanitary sewer, remedial actions will not be implemented until the aquifer has re-equilibrated and an additional round of groundwater level gauging is conducted to verify groundwater velocity and direction. Based on these observations, the number and alignment of injection and monitoring wells may be modified.

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SECTION 2

# Remedial Action Objective (RAO) and Applicable or Relevant and Appropriate Requirements (ARARs)

This section discusses the NCP and CERCLA objectives, identifies the Site 11 RAO and ARARs for the remedial actions considered in this FS.

# 2.1 NCP and CERCLA Objectives

The NCP requires that the selected remedy meet the following:

- Each remedial action selected shall be protective of human health and the environment [40 Code of Federal Regulations (CFR) 300.430 (f)(ii)(A)].
- Onsite remedial actions that are selected must attain those ARARs that are identified at the time of the Record of Decision (ROD) signature [40 CFR 300.430 (f)(ii)(B)].
- Each remedial action selected shall be cost-effective, provided that it first satisfies the threshold criteria set forth in §300.430(f)(1)(ii)(A) and (B). A remedy shall be cost-effective if its costs are proportional to its overall effectiveness.
- Each remedial action shall use permanent solutions and alternative treatment technologies or resource-recovery technologies to the maximum extent practicable {40 CFR 300.430 (f)(ii)(E)].

The statutory scope of CERCLA was amended by SARA to include the following general objectives for remedial action at all CERCLA sites:

- Remedial actions "shall attain a degree of cleanup of hazardous substances, pollutants, and contaminants released into the environment and of control of further releases at a minimum which assures protection of human health and the environment" (Section 121(d)(1)).
- Remedial actions in which treatment that "permanently and significantly reduces the
  volume, toxicity, or mobility of the hazardous substances, pollutants, and contaminants
  is a principal element are to be preferred" (Section 121(b)(1)). If the treatment or
  recovery technologies selected are not a permanent solution, an explanation must be
  published (Section 121 (b)(1)(G)).
- The least-favored remedial actions are those that include "offsite transport and disposal
  of hazardous substances or contaminated materials without treatment" where
  practicable treatment technologies are available (Section 121(b)(1)).

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The selected remedy must comply with or attain the level of any standard, requirement, criteria, or limitation under Federal environmental law or any promulgated standard, requirement, criteria, or limitation under a State environmental or facility siting law that is more stringent than any Federal standard, requirement, criteria, or limitation (Section 121(d)(2)(A)).

# 2.2 Remedial Action Objective

The only media of concern at Site 11 is groundwater. No unacceptable ecological risks are identified at Site 11. Remedial actions are developed for consideration to ensure protection of human health and to cost-effectively minimize disruption to the Base Mission and existing facility operations.

The RAO for the protection of human health and the environment for Site 11 groundwater is:

 Reduce concentrations in groundwater to the maximum extent practicable and maintain land use controls until concentrations allow for unlimited use and unrestricted exposure at Site 11.

#### 2.2.1 Development of Risk-Based Preliminary Remediation Goals (PRGs)

Preliminary Remediation Goals (PRGs) were developed for constituents with concentrations contributing appreciably to unacceptable risks and hazards from exposure to groundwater within Site 11. Based on the SRI HHRA (CH2M HILL, June 2004) and the Revised HHRA, SRI Addendum (CH2M HILL, January 2006), COPCs were identified as those constituents with cancer risks exceeding 10<sup>-4</sup>, or hazard index exceeding 1. The COPCs include one SVOC (PCP) and 13 VOCs, and are identified in Table 1-2.

To achieve remedial action objectives for unlimited use and unrestricted exposure, remediation goals are established as the MCL to the extent practicable. Because there is no established MCL for 1,1-DCA, a PRG was calculated using the same exposure assumptions used in the human health risk assessment and equations from the *Risk Assessment Guidance for Superfund Volume 1, Part B* (USEPA, December 1991) (Appendix C). To achieve remedial action objectives for unlimited use and unrestricted exposure, remediation goals are established as the MCL to the extent practicable.

The remediation goals for Site 11 groundwater are presented in Table 2-1.

# 2.3 Applicable or Relevant and Appropriate Requirements (ARARs)

As required by Section 121 of CERCLA, remedial actions carried out under Section 104 or secured under Section 106 must attain the levels of standards of control for hazardous substances, pollutants, or contaminants specified by the ARARs of federal and state environmental laws and state facility-siting laws, unless waivers are obtained. According to USEPA guidance, remedial actions should also be based on non-promulgated To-beconsidered (TBC) criteria or guidelines if the ARARs do not address a particular situation.

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ARARs are identified by the USEPA as either being applicable to a situation or relevant and appropriate to it.

"Applicable requirements" are standards and other environmental protection requirements of federal or state law dealing with a hazardous substance, pollutant, contaminant, action being taken, location, or other circumstance at a CERCLA site.

"Relevant and appropriate requirements" are standards and environmental protection criteria of federal or state law that, although not "applicable" to a hazardous substance, pollutant, contaminant, action being taken, location, or other circumstance, address problems or situations sufficiently similar to those encountered at the CERCLA site that their use is well suited to the particular site. A requirement that is relevant and appropriate must be met as if it were applicable. TBC criteria are non-promulgated advisories or guidance issued by federal or state government that are not legally binding, and do not have the status of potential ARARs. TBCs are evaluated along with ARARs and may be implemented by USEPA when ARARs are not fully protective of human health and the environment.

Onsite CERCLA response actions must meet substantive requirements but not administrative requirements. Substantive requirements are those dealing directly with actions or with conditions in the environment. Administrative requirements implement the substantive requirements by prescribing procedures such as fees, permitting, and inspection that make substantive requirements effective. This distinction applies to onsite actions only; offsite response actions are subject to all applicable standards and regulations, including administrative requirements such as permits.

Three classifications of requirements are defined by USEPA in the ARAR determination process: chemical-specific, location-specific, and action-specific. These classifications are described below. The remedial action alternatives developed in this FS were analyzed for compliance with the potential Federal and State ARARs, and are provided in Appendix D.

Chemical-specific ARARs are health or risk management-based numbers or methodologies that result in the establishment of numerical values for a given medium that would meet the NCP "threshold criterion" of overall protection of human health and the environment. These requirements generally set protective cleanup concentrations for the chemicals of concern in the designated media, or set safe concentrations of discharge for response activity. Federal and Commonwealth of Virginia chemical-specific regulations that have been reviewed are summarized in Appendix D.

Location-specific ARARs restrict response activities and media concentrations based on the characteristics of the surrounding environments. Location-specific ARARs may include restrictions on response actions within wetlands or floodplains, near locations of known endangered species, or on protected waterways. Federal and Commonwealth of Virginia location-specific regulations that have been reviewed are summarized in Appendix D.

Action-specific ARARs are usually technology- or activity-based requirements or limitations on actions taken with respect to hazardous substances. Federal and Commonwealth of Virginia action-specific ARARs that may affect the development and conceptual arrangement of response alternatives are summarized in Appendix D.

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SECTION 3

# Screening of Remedial Technologies and Identification of Remedial Alternatives

General response actions are broad responses, remedies, or technologies developed to meet site-specific RAO(s) and address COPCs, migration pathways, and exposure routes. The general response actions listed below have been identified for the remediation of Site 11:

- No Action
- In situ Treatment
- Land Use Controls
- Monitoring

The *No Action* response is included in accordance with the NCP to serve as a baseline for evaluation of the remedial actions.

*In situ Treatment* response actions are *in situ* methods of reducing the toxicity, mobility, or volume of contaminants in groundwater. Treatment technologies include biological and physical processes.

Land Use Controls (LUCs) consist of a number of alternatives that can be used alone or as part of another response action. LUCs include activities such as restricting groundwater use though land-use restrictions, deed restrictions, or access restrictions.

The *Monitoring* response action includes a groundwater sampling program to assess the behavior of contaminants over time, natural processes attenuating the contaminants, and performance of an active remediation.

Prior to implementing any alternative, the sanitary sewer line located south and east of Site 11 would be repaired. Following this repair, another round of groundwater samples, including water levels should be performed to confirm the extent of the plume, existing geochemical groundwater quality, baseline data and groundwater velocity and direction.

## 3.1 Screening of Remedial Technologies

Remediation of COPCs in groundwater at Site 11 is required to address potential unacceptable risks. Groundwater contamination to be addressed by the remedial alternatives consists of the "source" and the "plume." The source area at Site 11 is characterized by the highest groundwater concentrations and sorbed phase constituents. The plume area includes the entire area of groundwater contamination that consists predominantly of dissolved-phase constituents. The source and plume area at Site 11 are illustrated in Figure 1-5. The technologies were screened separately for the source and the plume to allow for the selection of the most appropriate technology for each area.

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An initial review of the available technologies was completed; technologies that were considered unsuitable for the remedial action at Site 11 were screened out early in the process. This screening process incorporated the Navy's preference to select a remedy that would minimize impacts to current land use, and minimize use of technologies requiring the construction and prolonged (greater than one year) operation of *ex-situ* systems. The technologies excluded from further consideration include pump and treat, soil vapor extraction, and air sparging. Based on the effectiveness of the CD pilot study, further consideration was not given to co-solvent flushing. Technologies that would not effectively treat all COPCs (e.g., zero valent iron) were also excluded from further consideration. Furthermore, *in situ* chemical oxidation was screened out due to cost and technical impracticability associated with delivering enough oxidant to meet the elevated site TOD. The assessment conducted in 2005 (Appendix A) verified there is no concern for potential vapor intrusion at Site 11. Consequently, vapor intrusion is not anticipated to be a concern with the implementation of the alternatives provided in this FS, and associated vapor mitigation and monitoring was not evaluated.

Technologies that were retained for further consideration included those that compliment the existing reducing conditions and the naturally occurring biodegradation of VOCs. Enhanced reductive dechlorination (ERD) was selected for further evaluation for treatment of both the source and the plume. Electrical resistance heating (ERH) was also selected for further evaluation. However ERH was only evaluated as a treatment technology for the source area since it is not considered a cost effective technology for the treatment of dissolved phase VOC plumes.

## 3.2 Description of Remedial Alternatives

Three remedial alternatives were developed from the technologies retained following the screening process. These are:

Alternative 1 - No Action

Alternative 2 - Enhanced Reductive Dechlorination

Alternative 3 - Electrical Resistance Heating & Enhanced Reductive Dechlorination

With the exception of Alternative 1 (no action), each of the remedial alternatives evaluated requires groundwater monitoring and the implementation of LUCs to prevent unacceptable risk exposure. Monitoring and LUCs would be maintained until groundwater concentrations allow for unlimited use and unrestricted exposure, with five-year statutory reviews to ensure protection of human health and the environment.

#### 3.2.1 Alternative 1 – No Action

Alternative 1 is the no-action alternative. Under this scenario, no remedial actions are taken at Site 11 and contaminants would remain in the groundwater at Site 11.

#### 3.2.2 Alternative 2 – Enhanced Reductive Dechlorination (ERD)

Biological reductive dechlorination is a naturally-occurring, microbially-mediated, anaerobic process in which chlorine atoms on a parent VOC molecule are sequentially

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replaced with hydrogen. In the reductive dechlorination process, electrons are transferred from an electron donor source to the VOC compound, which functions as the electron receptor. Therefore, an external electron donor source is required for the reaction to occur. Potential electron donor sources include biodegradable organic co-contaminants, native organic matter, or substrates intentionally added to the subsurface. Deeply anaerobic (reducing) conditions are required for reductive dechlorination of many VOCs, and competing electron acceptors such as dissolved oxygen, nitrate, nitrite, manganese(IV), ferrous iron, and sulfate must be depleted.

The predominant parent COPCs at Site 11 are TCE and 1,1,1-TCA. The principal anaerobic biodegradation pathway for TCE is:

TCE 
$$\rightarrow$$
 cis-1,2-DCE  $\rightarrow$  vinyl chloride (VC)  $\rightarrow$  ethene  $\rightarrow$  ethane

The transformation rate for each step varies but tends to become slower with progress along the breakdown sequence, often resulting in accumulation of 1,2-DCE and VC. Further breakdown from 1,2-DCE and VC to ethene varies and is based on site specific conditions.

1,1,1-TCA degrades biotically to 1,1-DCA and abiotically to 1,1-DCE. Following this step, the principal anaerobic biodegradation pathway is:

1.1-DCA  $\rightarrow$  chloroethane  $\rightarrow$  ethane

1,1-DCE  $\rightarrow$  VC  $\rightarrow$ ethene  $\rightarrow$  ethane

Complete dechlorination of TCE and 1,1,1-TCA has been occurring and is expected to continue at Site 11.

Enhanced anaerobic bioremediation of VOCs is implemented by adding a suitable substrate (soluble or insoluble) to the subsurface. The introduced substrate serves multiple purposes: depletion of competing electron acceptors, creating strongly reducing conditions, and producing an electron donor source for reductive dechlorination.

The most commonly used insoluble substrates are Hydrogen Release Compound® (HRC®) and vegetable oil. Vegetable oil is injected as an emulsified liquid. Linoleic and other long chain fatty acids in the vegetable oil slowly solubilize in water over time and are broken down by native microorganisms to lower molecular weight fatty acids such as pyruvate and propionate. Ultimately, the oil degrades to form acetic acid and hydrogen. The hydrogen and dissolved organic carbon from the acetic acid are then available to support reductive dechlorination of chlorinated solvents. Vendors estimate that vegetable oil may serve as an electron donor for at least a year and as much as three years depending on site specific conditions, and are typically applied via direct push technology (DPT) points.

Soluble substrates include benzoate, lactate, acetate, propionate, butyrate, methanol, ethanol, sucrose, molasses, and hydrogen (H<sub>2</sub>). These substrates are water soluble, degrade rapidly, and are transported with groundwater flow. Since these substrates degrade rapidly, they typically require more frequent injections than insoluble substrates and therefore are generally dispensed via permanent injection wells.

For the purpose of this FS conceptual design and cost estimate, sodium lactate, a widelyused and effective soluble substrate, was selected. Sodium lactate is available in 55 gallon

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drums or approximately 2000 gallon totes. It is typically delivered as a 60 % solution. The cost estimate was prepared assuming multiple injections over time to maintain the electron donor available for use by dechlorinating bacteria. The repeated substrate injections throughout each year of substrate injection also serves to increase subsurface mixing thereby enhancing substrate distribution which subsequently allows for increased degradation of COPCs. To minimize disruption of current land use, the use of a slow-release organic substrate (e.g., vegetable oil), which requires less frequent injections, may be substituted for lactate at the onset of the remedial action. However, reducing the number of injections minimizes subsurface mixing and may consequently increase the length of time the remedial action is implemented. If COPC degradation is not sufficient, use of the slow-release organic substrate should be replaced by use of a soluble substrate.

#### **Source Treatment**

For treating the source area, an injection well array, with wells spaced on no more than approximately 15 ft centers, was selected. The existing injection wells (LS11-MW23D, LS11-MW24D, LS11-MW25D, LS11-MW26D, LS11-MW27D, LS11-MW28D, LS11-MW29D, and LS11-MW30D), which were installed as part of the CD injection pilot test, are suitable for substrate injection. These existing wells are constructed of 4 inch-diameter polyvinyl chloride (PVC) with 5 ft of well screens. Because these wells provide adequate coverage of the target source area, no new injection wells are considered necessary.

As shown on Figure 3-1, two new monitoring wells are proposed to monitor the performance of the ERD process in the source area. One well is located within the target injection zone and will measure changes in groundwater quality that occur within the injection zone. The other performance monitoring well is located slightly downgradient of the injection zone to measure changes in groundwater quality migrating from the source zone. In addition to these monitoring wells, periodic monitoring of wells LS11-MW5S and LS11-MW5D is recommended.

#### **Plume Treatment**

TCE concentrations collected from monitoring well LS11-MW10D are less than  $500 \,\mu g/L$ , but are greater than concentrations detected in other portions of the downgradient plume. Therefore, additional treatment in this location was selected to expedite cleanup of this localized area. To target this area, two injection wells will be installed approximately 10 ft upgradient of well LS11-MW10D and well LS11-MW10D will be monitored to evaluate the performance of the ERD process in this area.

A biobarrier, consisting of 23 injection wells spaced at approximately 15 ft centers and located near the downgradient edge of the plume, was selected based on current groundwater flow conditions (Figure 3-1). Following repair of the leaky sanitary sewer and stabilization of groundwater flow, groundwater flow will be re-evaluated for effective placement of the biobarrier. It is anticipated the injection wells in the biobarrier will be installed at least 15 to 20 ft from the parking lot and road. This will allow space to install three downgradient performance monitoring wells in the unpaved area. In addition to these new wells, well LS11-MW09 will be monitored to evaluate system performance.

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#### **Well Construction**

New monitoring wells will be constructed of 2 inch-diameter PVC with 5 ft well screens, whereas new injection wells will be constructed of 2 inch diameter PVC, with 5 ft continuous slot (wire-wrapped) well screens. The wells should be constructed to the top of the Yorktown Confining Unit which is located approximately 23 ft bgs. Soil cores will be collected via DPT along the proposed biobarrier alignment to ensure that the correct depth is established prior to injection well installation. To substantiate the proposed biobarrier location and width, groundwater samples will be collected from DPT locations and analyzed to confirm the presence of VOCs

#### Substrate Injection and Performance Monitoring

For the source area and biobarrier injection wells, the target volume of injectate for each injection event is the amount necessary to achieve a radius of influence (ROI) equal to half of the distance between each well. For a 5 ft well screen, a target radius of influence of 7.5 ft, and an assumed effective porosity of 0.20, the target injectate volume per well is approximately 1,320 gallons.

The injectate solution should initially have a lactate concentration of 1% (10,000 milligrams (mg)/L). As the treatment progresses and the ERD system matures, this concentration may increase or decrease based on the system response and frequency of injection. It is difficult to estimate the treatment time required to achieve adequate reduction in VOCs to allow active remediation to cease. For the purpose of this FS it was assumed that the source and plume would receive six substrate injections per year during year zero of the remedial action. During years one through seven, the source would receive four injections per year, while the plume would receive three injections per year. It is assumed that the source will be adequately treated after year seven and during years eight through 14, only the plume would require treatment (at a frequency of 3 injections per year).

Sampling and analysis of the ERD process is important to ensure that effective and optimal conditions are established for the microorganisms. A proposed performance monitoring schedule and analyte list is provided in Table 3-1. Additionally, groundwater monitoring will be required to continue after active remediation ceases if VOC concentrations in groundwater continue to exceed MCLs.

# 3.2.3 Alternative 3 – Electrical Resistance Heating (ERH) & Enhanced Reductive Dechlorination (ERD)

In situ thermal treatment (electrical resistive heating or conductive heating) is an applicable technology for treatment of high concentrations of dissolved- and sorbed-phase VOCs. This technology involves the active heating of the subsurface to force volatile contaminants into the vapor phase where they can vent to the ground surface or be removed by an active vapor extraction system for ex-situ treatment. Thermal treatments also typically vaporize some or all of the pore water within an aquifer to steam, which either carries or flushes contaminants to a vapor extraction point. In addition to the physical destruction of VOCs, thermal treatment increases microbial activity of dechlorinating bacteria, which enhance the naturally occurring biological degradation of VOCs.

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ERH involves the placement of a network of electrodes in the subsurface and the application of current through the subsurface. Resistance to current flow within the subsurface produces heat. ERH is typically used to raise subsurface temperatures to the boiling point of the contaminant, causing partial vaporization of the contaminant within the treatment zone. Steam generated by this process acts as a contaminant carrier and migrates upward to the vadose zone, where co-located vapor extraction wells remove the steam for further treatment at an aboveground treatment system. Because this process relies on elevating the temperature of water, ERH is only capable of volatilizing constituents with boiling points of 100 °C or less.

Conductive heating involves the application of a network of direct-heating probes installed within subsurface wells. Heat from the probes, typically installed within a well also used for vapor extraction, is transmitted through the subsurface by conductance. Conductive heating is typically used to raise subsurface temperatures significantly above the water boiling point, forcing the complete vaporization of all pore water near the heating probes. Vaporized steam can then be extracted at depth without requiring steam to migrate to the vadose zone.

For the purpose of the FS, it is assumed that the source area is to be treated using ERH followed by polishing and plume treatment with ERD (Figure 3-2). With the exception of PCP, all site COPCs have a boiling point of less than 100 °C and can be treated via ERH. Although PCP has a boiling point of greater than 100 °C, PCP was detected only once in a sample collected in 1999 from monitoring well LS11-MW04D and this sample location is not the area to be remediated by thermal treatment. Additionally, PCP is reductively dechlorinated by anaerobic bacteria, and hence will be treated by ERD.

The cost estimate for the ERH system and operation was provided by a vendor that specializes in the construction of ERH systems. However, if ERH is selected as the source treatment technology for this site, the ultimate design of the system will be completed by the vendor awarded the work. The cost estimate for Alternative 3 was based on the following assumptions: the cost estimate prepared by the vendor was representative of the cost to implement ERH in the designated source area; the source area and plume area are comparable in size and location as those designated for Alternative 2; and the remainder of the plume, including the elevated concentrations at monitoring well LS11-MW10D would be treated and monitored using the approach described for Alternative 2 (e.g., biobarriers, injection, and monitoring wells).

Because of the thermal stress imposed in the source area, the existing PVC wells located within the thermal treatment area will need to be abandoned and replaced with stainless steel wells. The new well should be placed as close as possible to the existing wells and their screened intervals should be the same as the existing wells so that the analytical data from these new wells is comparable to the previous sampling data. Also, the new wells should be constructed with continuous slot well screens since they will be used as injection wells for the ERD polishing after ERH is completed.

For cost estimating purposes, it was assumed that the ERH system would operate for four months. However, residual CD remaining in the subsurface from the CD pilot study may reduce the rate at which COPCs volatilize, thereby resulting in a longer ERH operating

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period. During the period in which the ERH system is operating, groundwater sampling would be completed after the second and third month of operation, and twice during the fourth month of operation to monitor the effectiveness of the remedy. If groundwater concentrations are sufficiently reduced, operation may cease. However, if COPC concentrations remain elevated or rebound the ERH system will continue to operate. Based on the effectiveness of this remedy and previous experience it is not anticipated that this system will be required to operate for greater than nine months.

Since ERH may not reduce concentrations to MCLs, ERD will be used as a polishing step in the source area. For cost estimating purposes, it was assumed that the ERH system would operate for four months followed by two subsequent source area ERD injections in year zero. It was assumed that source area polishing would be necessary for three more years (at four injections per year). Similar to Alternative 2, it was assumed that the plume would receive six injections during year zero and three injections per year during years one through 14. It was assumed that monitoring would be completed as described for Alternative 2. However to meet the ERH performance monitoring schedule, one additional sampling round would be completed for VOCs, TOC, methane, ethane, and volatile fatty acids (VFAs) during year zero. The monitoring schedule for Alternative 3 is provided in Table 3-2.

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# **Evaluation of Remedial Alternatives**

Remedial alternatives, including the no action alternative, were developed for Site 11 to reduce concentrations of VOCs in groundwater to meet the RAO.

#### 4.1 Evaluation Criteria

The remedial alternatives that have been developed for Site 11 are evaluated based on nine NCP criteria. Each alternative is evaluated and with respect to each NCP criterion and one another. The purpose of this comparative analysis is to identify the relative advantages and disadvantages of each alternative and identify the best balance of trade offs for remedy selection. The Navy developed this FS in partnership with the EPA and VDEQ, and therefore concurs with the comparative analysis and selection of a preferred remedial alternative. Community acceptance for selection of a preferred remedial alternative will be addressed in the ROD for Site 11. The nine NCP criteria are:

- Overall protection of human health and the environment
- Compliance with ARARs
- Long-term effectiveness and permanence
- Reduction of toxicity, mobility, and volume
- Short-term effectiveness
- Implementability
- Cost
- State acceptance
- Community acceptance

For the cost analysis, the expenditures required to complete each remedial action are estimated in terms of both capital and annual operations and maintenance (O&M) costs. All expenditures for Year 0 were included as capital costs. Assumptions associated with present worth calculations include a discount rate of 3.1 percent (Office of Management and Budget (OMB), January 2005), cost estimates in the planning years in constant dollars, and a period of performance that would vary depending on the activity, but would not exceed 30 years.

The cost estimate for each alternative is provided as an order of magnitude cost estimate and were estimated from comparable projects (e.g., engineering experience) and quotations. The estimate has been prepared without equipment specifications, layout, design, or engineering calculations. The expected level of accuracy is +50 percent to -30 percent. The cost estimates are in 2005 dollars and are based on the current conceptual design. Cost estimates for Alternatives 2 and 3 are provided in Appendix E.

## 4.2 Detailed Analysis of Remedial Alternatives

A summary of the detailed analysis of each remedial alternative is presented below and summarized in Table 4-1.

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#### 4.2.1 Alternative 1—No Action

Evaluating a "no action" alternative is required by the NCP. Under this alternative, no further effort or resources would be expended to remediate contaminated groundwater at Site 11. Because contaminated media would be left on the site, a review of site conditions would be required every 5 years. Alternative 1 serves as the baseline against which the other alternatives are judged.

#### Overall Protection of Human Health and the Environment

The No Action alternative is not protective of human health and the environment. This alternative does not provide any means to prevent exposure to contaminated groundwater or measures to reduce contamination to acceptable levels that would allow unlimited use and unrestricted exposure.

#### Compliance with ARARs

VOC concentrations in groundwater exceed MCLs. The No Action Alternative does not include measures to reduce VOC concentrations; therefore Alternative 1 does not comply with chemical-specific ARARs. There are no location- or action-specific ARARs for this alternative because no remedial actions would be undertaken.

#### Long-Term Effectiveness and Permanence

Although groundwater sampling at Site 11 indicates VOCs are undergoing reductive dechlorination, with no action to enhance this process it is uncertain if the natural dechlorination could reduce concentrations to levels that would allow unlimited use and unrestricted exposure, and the time frame for natural dechlorination is considered unacceptable. Furthermore, LUCs would not be in place to prevent exposure to COPCs. Therefore the adequacy and reliability of this alternative is very low rendering Alternative 1 ineffective over the long term.

#### Reduction of Toxicity, Mobility, and Volume through Treatment

Alternative 1 has no feature that would act to reduce toxicity, mobility, or volume through treatment. Because no remedial actions would be undertaken, reduction of toxicity, mobility, and volume would only gradually occur as a result of natural processes.

#### **Short-Term Effectiveness**

There is no construction associated with this Alternative 1, so there are no adverse short-term impacts on workers, the community, or the environment.

#### **Implementability**

There are no issues concerning technical implementation of No Action.

#### Cost

Taking no action would require no capital expenditure.

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#### 4.2.2 Alternative 2: ERD

Alternative 2 involves implementation of ERD technology for treatment within the source and plume areas, post-treatment groundwater monitoring, and LUCs in the form of land and groundwater use restrictions.

#### Overall Protection of Human Health and the Environment

Alternative 2 is protective of human health and the environment. This alternative would actively treat COPCs and prevent human exposure through the use of LUCs during the implementation of the remedy until the RAO is met. The use of the biobarrier would also prevent or minimize the migration of COPC concentrations exceeding MCLs into currently unaffected media.

#### Compliance with ARARs

Alternative 2 would comply with chemical-, location-, and action-specific ARARs. Injection of substrate would enhance naturally occurring biological degradation processes to reduce VOC concentrations in groundwater, and is expected to comply with chemical-specific ARARs. The substantive requirements associated with injection and the storage, analysis, and disposal of waste generated during implementation of this alternative would be met. Therefore this alternative is expected to comply with location- and action-specific ARARs. Appendix D contains a detailed evaluation of ARARs for Alternative 2.

#### Long-Term Effectiveness and Permanence

Alternative 2 would effectively reduce concentrations of VOCs in groundwater to unlimited use and unrestricted exposure. Naturally occurring degradation processes would be accelerated by injecting a fermentable organic substrate to stimulate native microbes to degrade chlorinated solvents. In addition to source treatment, a biobarrier would be installed along the downgradient edge of the plume. Following the termination of the substrate injection activities, the aquifer would be conditioned for continued degradation of VOCs. Consequently, once adequately treated, VOC concentrations would remain below MCLs assuming that any source material is removed and no external source area is present.

LUCs and 5-year reviews would be implemented until levels allow for unlimited use and unrestricted exposure. VOCs would be removed through source treatment, plume treatment, and the installation of a downgradient biobarrier thereby reducing risk associated with migration of groundwater.

LUCs are expected to be adequate and reliable, and a groundwater monitoring program would be implemented to substantiate the effectiveness of the remedial action through tracking groundwater quality COPC over time.

#### Reduction of Toxicity, Mobility, and Volume through Treatment

Implementation of ERD would reduce the toxicity, mobility and volume of the VOCs in the source area and plume. Natural processes are expected to occur at an accelerated rate to degrade the remaining dissolved phase COPCs.

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#### **Short-Term Effectiveness**

Alternative 2 requires the initial installation of injection and monitoring wells and regularly scheduled injections and groundwater monitoring throughout the life of this remedial action. Investigation derived waste (IDW) requiring disposal would be generated during well installation and during groundwater monitoring. IDW would be containerized and temporarily stored on-site prior to disposal at an approved facility. NAB Little Creek maintains a temporary storage area sufficient to accommodate the small volume of waste generated during the implementation of this alternative. Health and safety precautions would be required to protect workers and the community during drilling, transport and storage of IDW, and throughout subsequent substrate injections. Since ERD is an *in situ* technology, impacts to the community, workers and the environment are minimized.

#### Implementability

#### Technical Implementability and Availability of Services and Materials

ERD is a proven technology in which the addition of substrate to the subsurface provides the necessary conditions for dechlorinating bacteria to degrade VOCs. The previous investigations confirmed abundant populations of dechlorinating bacteria (Appendix B), thereby reducing uncertainty associated with this alternative.

The installation of the injection and monitoring wells is straightforward and can be accomplished by an experienced environmental drilling firm. The subsequent substrate injections follow a basic procedure that can be accomplished with relative ease. Well locations have been selected to maximize the effectiveness of this alternative while minimizing disturbance to the site. Nevertheless, disruption to existing land use (parking lot, access to Building 3602, and landscaped areas) would occur as the wells are installed and during injection and groundwater monitoring. The management of IDW generated during well installation, substrate injection, and groundwater monitoring is routinely and easily implemented.

The effectiveness of ERD would be monitored by analyzing groundwater geochemistry, the decrease in parent compounds, and presence of daughter products caused by biological degradation of the parent compound. Groundwater samples collected from up-gradient, mid-gradient, down-gradient, and side-gradient wells would provide data needed to monitor changes in VOC concentrations and plume size and location.

#### Administrative Implementability

Long-term administrative resources for implementation of LUCs and annual reviews would also be required, and can easily be implemented throughout the duration of this alternative, which is assumed to be 30 years.

#### Cost

The present value cost for Alternative 2 is \$2,399,000 (Appendix E). The capital cost associated with Alternative 2 is \$499,000 and includes well installation, the first year (Year 0) of substrate injection, sampling, reporting, and the cost incurred for the implementation of LUCs. Annual operating costs include substrate injection, sampling, annual site inspections, and associated reporting. These costs are expected to be incurred through Year 14. Annual costs are greater during Year 1-7 due to an additional injection (per year) in the source zone.

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The cost associated with the 5-year reviews is presented as periodic costs incurred every 5 years. Long term monitoring costs include sampling to monitor the effectiveness of the remedy, annual site inspections, and the reporting associated with these activities. Long-term monitoring costs are provided through year 30.

As described in Section 3.2, this FS assumes sodium lactate would be used as the injectate for this alternative. Accordingly, the cost for this alternative is estimated using the cost to purchase sodium lactate as well as the cost for the injection procedures and schedules associated with the use of sodium lactate. However, a variety of other substrates are available and the actual cost to implement ERD would be dependent upon the substrate cost, the number of annual injections, and the effectiveness of the substrate that is ultimately selected.

Because of the uncertainty associated with the time required to reach the RAO, a conservative number of years for injection was used to estimate the cost of this alternative. The sampling scheme associated with the implementation of this alternative would provide a mechanism to evaluate the effectiveness of the remedy. An extension of the injection schedule may be required if the VOC concentrations are not adequately reduced, thereby causing a higher cost for the implementation of this alternative. Conversely, the number of injection may be reduced if VOCs continue to attenuate at an acceptable rate without the further addition of substrate.

#### 4.2.3 Alternative 3: ERH and ERD

Alternative 3 involves implementation of ERH for treatment of the source area, ERD for a polishing step following ERH source treatment, and ERD treatment for the plume. The ERH system is targeted to accelerate mass reduction of VOCs in the source zone, and ERH and ERD are expected to increase naturally occurring biological degradation processes to further reduce VOC concentrations in groundwater. This alternative includes post treatment groundwater monitoring and LUCs in the form of land and groundwater use restrictions.

#### Overall Protection of Human Health and the Environment

Alternative 3 is protective of human health and the environment. This alternative would actively treat COPCs and prevent human exposure through the use of LUCs during the implementation of the remedy until the RAO is met ensuring protection of human health and the environment. The use of ERH would provide expedited mass reduction in the source area. The use of the biobarrier would also minimize the migration of COPCs in groundwater into unaffected media.

#### Compliance with ARARs

Alternative 3 is expected to comply with chemical-, location-, and action-specific ARARs. A system would be constructed to treat vapors generated during ERH operation. Alternative 3 is expected to comply with chemical-specific ARARs. The substantive requirements associated with injection and the storage, analysis, and disposal of IDW generated during implementation of this alternative would be met. Therefore this alternative is expected to comply with location- and action-specific ARARs. Appendix D contains a detailed evaluation of ARARs for Alternative 3.

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#### **Long-Term Effectiveness and Permanence**

Alternative 3 would effectively reduce concentrations of VOCs in groundwater to unlimited use and unrestricted exposure. The increase in subsurface temperature caused by the operation of the ERH system and the injection of substrate would stimulate the native dechlorinating microbial populations thereby accelerating the naturally occurring degradation processes. Additionally, the installation of the biobarrier along the downgradient edge of the plume would serve as a mechanism to promote continued degradation of the plume. In conjunction with the implementation of this alternative, the anaerobic conditions present at Site 11 would provide for continued degradation of VOCs following the completion of the ERH operation and substrate injections. Therefore, once adequately treated, VOC concentrations would remain below MCLs assuming that any source material is removed and no external source area is present.

LUCs and 5-year reviews would be implemented until levels allow for unlimited use and unrestricted exposure. ERH would expedite mass reduction of VOCs in the source area and the injection of substrate would enhance biological degradation of VOCs in the plume thereby mitigating risk associated with the migration of VOCs to currently unaffected media.

LUCs are expected to be adequate and reliable, and a groundwater monitoring program would be implemented to substantiate the effectiveness of the remedial action through tracking groundwater quality and COPC concentrations over time.

#### Reduction of Toxicity, Mobility, and Volume through Treatment

Alternative 3 would provide a reduction of VOC toxicity, mobility, and volume in the source and plume through enhanced (physical and biological) degradation of VOC COPCs.

#### Short-Term Effectiveness

Construction activities for the implementation of Alternative 3 include the abandonment of monitoring wells in the source zone, the initial installation of injection and monitoring wells in the source zone and plume, and the installation of an ERH system in the source area. Additionally, this alternative requires the maintenance of the ERH system during ERH operation and regularly scheduled injections and groundwater monitoring events. IDW requiring disposal would be generated during well abandonment, well installation, ERH installation, and during groundwater monitoring. IDW would be containerized and temporarily stored on-site prior to disposal at an approved facility.

Health and safety precautions would be required to protect workers and the community during drilling, ERH operation, transport and storage of IDW, and throughout subsequent substrate injections. Since ERD is an *in situ* technology, impacts to the community, works and the environment are minimized. However, ERH contains *ex-situ* components including the power control system, vapor recovery and treatment system, and electrodes. Precautions would be necessary to minimize impacts to the community, environment, and the operation of the facility. Additionally, engineering controls would be constructed to prevent exposure to high voltages.

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#### Implementability

#### Technical Implementability and Availability of Services and Materials

ERH is a proven technology capable of providing expedited mass reduction. As a result of the increase in subsurface temperatures, VOCs are physically degraded and populations of dechlorinating bacteria are stimulated, providing biological degradation of VOCs. The CD remaining in the subsurface following the pilot study contributes to the anaerobic conditions, thereby providing a suitable habit for dechlorinating bacteria. However, it is unknown how the CD would affect the volatilization of COPCs and the ERH system may require a longer than anticipated period of operation. ERD is also a proven technology in which the addition of substrate to the subsurface provides the necessary conditions for dechlorinating bacteria to degrade VOCs. Results from microbial analysis confirmed abundant populations of dechlorinating bacteria (Appendix B), thereby reducing uncertainty associated with this alternative.

The use of an ERH system requires PVC wells within the ERH treatment area be abandoned and replaced with stainless steel wells capable of withstanding the heat generated during operation. The design and construction of the system should be completed by an experienced vendor familiar with this type of thermal treatment. Following construction, the ERH system is anticipated to operate for approximately four months. During this time, the system would require monitoring and upkeep of the vapor recovery system. Since this system includes *ex-situ* components, impact to daily use of the site can not be avoided during system operation. For instance engineering controls would be required to prevent exposure to high voltages and to the power control system. However, to the greatest extent possible, the system would be designed with the intent to minimize impacts to the use of the site.

The abandonment and installation of the injection and monitoring wells is straightforward and can be accomplished by an experienced environmental drilling firm. The subsequent substrate injections follow a basic procedure that can accomplished with relative ease. Well locations have been selected to maximize the effectiveness of this alternative while minimizing disturbance to the site. Nevertheless, the use of some areas of the site would be temporarily impeded during well abandonment and installation, substrate injection, and groundwater monitoring.

IDW generated during well installation, substrate injection, and groundwater monitoring would be containerized and temporarily stored on-site prior to disposal in an approved facility. NAB Little Creek routinely manages IDW and maintains a temporary storage area sufficient to accommodate the small volume of waste generated during the implementation of this alternative.

The effectiveness of ERH can be measured by the overall decrease in COCs in the source area, which can be determined by groundwater monitoring throughout and subsequent to the operation of the ERH system. The effectiveness of ERD can be monitored by analyzing groundwater geochemistry, the decrease in parent compounds, and presence of daughter products caused by biological degradation of the parent compound. Groundwater samples collected from up-gradient, mid-gradient, down-gradient, and side-gradient wells would provide data needed to monitor changes in VOC concentrations and plume size and location.

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#### Administrative Implementability

Long-term administrative resources for implementation of LUCs and annual reviews would also be required throughout the assumed 30-year duration of this alternative.

#### Cost

The present value cost for Alternative 3 is \$2,841,000 (Appendix E). The capital cost associated with Alternative 2 is \$1,047,000 and includes PVC well abandonment in the source zone followed by stainless steel well installation in the source zone, PVC well installation in the plume, ERH construction and operation, the first year (Year 0) of substrate injection, sampling, reporting, and the cost incurred for the implementation of LUCs. Annual operating costs include substrate injection, sampling, annual site inspections, and associated reporting. These costs are expected to be incurred through Year 14. Annual costs are greater during Year 1-3 due to an additional injection (per year) in the source zone. The cost associated with the 5-year reviews is presented as periodic costs incurred every 5 years. Long term monitoring costs include sampling to monitor the effectiveness of the remedy, annual site inspections, and the reporting associated with these activities. Long-term monitoring costs are provided through year 30.

A vendor quote was used to estimate the cost to construct and operate the ERH system. This cost may vary based on the actual vendor selected. As described in Section 3.2, this FS assumes sodium lactate would be used as the injectate for the ERD portion of this alternative. Accordingly, the cost for ERD is estimated using the cost to purchase sodium lactate as well as the cost for the injection procedures and schedules associated with the use of sodium lactate. However, a variety of other substrates are available and the actual cost to implement ERD would be dependent upon the substrate cost, the number of annual injections, and the effectiveness of the substrate that is ultimately selected.

Because of the uncertainty associated with the time required to reach the RAO, a conservative number of years for injection was used to estimate the cost of this alternative. The sampling scheme associated with the implementation of this alternative would provide a mechanism to evaluate the effectiveness of the remedy. An extension of the injection schedule may be required if the VOC concentrations are not adequately reduced, thereby causing a higher cost for the implementation of this alternative. Conversely, the number of injection may be reduced if VOCs continue to attenuate at an acceptable rate without the further addition of substrate.

# 4.3 Comparative Analysis of Remedial Alternatives

This section provides a comparison analysis to evaluate the relative performance of each alternative in relation to each other and the NCP criteria. The purpose of this analysis is to identify the advantages and disadvantages of each alternative relative to one another. A qualitative comparative analysis employed a ranking system of 1, 3, and 5, with 1 being the lowest valued metric and 5 being the highest. The results of the ranking for each alternative are included in Table 4-2.

Alternative 1, no action, is easily implemented, with no concerns for short term effectiveness and no associate cost. However, Alternative 1 does not provide protection of human health and the environment, does not comply with ARARs, is not effective in the long term, and

does not reduce toxicity, mobility or volume through treatment. Alternative 1 is serves only as a baseline for the comparative analysis.

#### 4.3.1 Overall Protection of Human Health and the Environment

The overall assessment of protection draws on the assessments of long-term effectiveness and permanence, short-term effectiveness, and compliance with ARARs. Alternatives 2 and 3 are both protective of human health and the environment and are comparable in their evaluation against long-term effectiveness and permanence and compliance with ARARs. However, as a result of the additional construction associated with the ERH and the *ex-situ* component of Alternative 3, Alternative 2 has greater short term effectiveness, and is therefore the most protective alternative evaluated.

#### 4.3.2 Compliance with ARARs

Alternative 2 and 3 are expected to comply with ARARs. Since vapors extracted during ERH operation would require treatment, Alternative 3 includes additional ARARs. However, a vapor recovery system is incorporated in to the design and cost of this alternative, and would comply with ARARs. Appendix D contains a detailed evaluation of ARARs for Alternatives 2 and 3.

#### 4.3.3 Long-Term Effectiveness and Permanence

Alternatives 2 and 3 would effectively reduce concentrations of VOCs in groundwater to unlimited use and unrestricted exposure. The use of ERH in the source zone is expected to increase the rate of mass reduction in the source zone, thereby decreasing the number of substrate injections in the source zone. Plume treatment with ERD, monitoring, and LUCs, are similar for Alternatives 2 and 3. Therefore, these alternatives are considered equally effective in achieving long-term effectiveness.

#### 4.3.4 Reduction of Toxicity, Mobility, and Volume through Treatment

Alternatives 2 and 3 are similar in their use of substrate to enhance naturally occurring reductive dechlorination. However, Alternative 3 is most effective in achieving this criterion since it includes the use of ERH as source treatment to expedite mass reduction of VOCs in the source area.

#### 4.3.5 Short-Term Effectiveness

The short-term effectiveness associated with Alternatives 2 and 3 are similar with respect to the implementation of ERD. Alternative 3 however, requires the replacement of wells and the construction and maintenance of an ERH system. Additionally, since there is an *ex-situ* component associated with ERH, there is additional potential for worker, community, and environmental exposure. Therefore Alternative 2 provides the greatest short-term effectiveness.

#### 4.3.6 Implementability

The implementability associated with Alternative 2 and the ERD component of Alternative 3 is similar, with the exception that Alternative 3 requires wells located in the source area are replaced. Since Alternative 3 includes the construction, operation, and maintenance of an

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ERH system, disruption of existing land use (parking, building access, and landscaped areas) in the vicinity of the School of Music, and difficulty of implementation is greater with Alternative 3. Alternative 2 is therefore easier to implement than Alternative 3.

#### 4.3.7 Cost

The cost estimate for Alternatives 2 and 3 are provided in Appendix E. The use of ERH reduces the number of substrate injections in the source area, and therefore reduces annual injection costs during years three through seven. Nonetheless, the cost associated with replacing wells in the source area and the construction, operation, and maintenance of the ERH system for Alternative 3, renders Alternative 2 the most cost effective alternative.

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# Rationale for the Preferred Alternative

The detailed evaluation (Section 4.2) followed by the comparative analysis (Section 4.3) of the remedial alternatives provided the basis for identifying the preferred alternative. Alternative 1 does not meet the statutory requirements of the NCP and is not a viable remedial action for this site. While the ERH component of Alternative 3 allows for expedited mass removal of the source area, this benefit does not outweigh the cost, greater difficulty associated with implementation, and lower short-term effectiveness associated with increased exposure to workers, the community, and the environment during construction, operation, and maintenance of the ERH system. The balance of trade-offs in the comparative analysis is illustrated in Table 4-2 and identifies Alternative 2 as the preferred alternative. In comparison to Alternative 3, Alternative 2 is protective of human health and the environment, complies with ARARs, is effective in the short- and long-term, reduces toxicity, mobility, and volume through treatment, is readily implemented, and is cost effective. Since, this site contains the anaerobic conditions necessary for reductive dechlorination the implementation of Alternative 2 would serve to enhance the biological degradation COPCs that is actively occurring at Site 11.

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6-2

**Tables** 

#### Table 1-1 History of Environmental Investigations Site 11, NAB Little Creek Virginia Beach, Virginia

Virginia Beach, Virginia					
Date of Report	Author	Report Title	Purpose of Investigation		
December 1984	Rogers, Golden and Halpern	Initial Assessment Study	To identify and assess sites posing a potential threat to human health or to the environment due to contamination from past hazardous materials operations.		
October 1986	CH2M HILL	Final Progress Report Round 1 Verification Step	To present the results of the Verification Step, Round 1 sampling at Site 11 performed under the NACIP program. 8 soil samples and 3 groundwater samples were collected for pollutant VOCs and acid extractables, Chromium III & VI, and cyanide.		
November 1991	Ebasco Environmental	Interim Remedial Investigation	To determine whether or not further characterization activities or remedial action is warranted at Site 11. 3 groundwater samples, 1 tank liquid sample, and 2 tank solid samples were collected and analyzed for VOCs, acid extractables, and total TAL metals.		
	Foster Wheeler Environmental	Final Remedial Investigation/Feasibility Study	To fill information gaps and collect additional site-specific data necessary to fully evaluate site conditions, determine potential risks posed by each site, and develop and evaluate remedial action alternatives to mitigate any risks found. 3 groundwater samples were collected and analyzed for VOCs and total and dissolved metals. 10 surface soil samples were analyzed for VOCs and 5 surface soils were analyzed for metals.		
May 1996	IT Corporation	Draft Final Closeout Report for Site 11	Document the soil conditions after the removal of the neutralization tank and piping 18 subsurface soil samples were collected and analyzed for VOCs and 8 metals.		
February 1998	CH2M HILL	February 1998. Final Groundwater Monitoring Report, Sites 5 and 11	Monitor and document the groundwater conditions after the removal of the neutralization tank and piping. 3 groundwater samples were collected in March and December for VOCs and		
June 2004	CH2M HILL	Supplemental Remedial Investigation for Site 11	DPT and MIP sampling to establish the horizontal and vertical extent of the VOC and PCP plumes, the VOC source area, and aquifer (shallow and deep) characteristics, to conduct a Human Health Risk Assessment, and to evaluate the integrity of the sanitary sewer. Soil, subsurface soil, and groundwater collected and analyzed for VOCs, SVOCs, and/or metals.		
June 2000	CH2M HILL	Screening Ecological Risk Assessment, IR Sites 5, 7, 9, 10, 11, 12, 13, & 16, and SWMU 3	To confirm the absence of potential ecological risks.		
	ESTCP. Boving et al.	Draft Cyclodextrin Enhanced In-situ Removal of Organic Contaminants from Groundwater at Department of Defense Sites	Evaluate the in-situ removal of organic contaminants from groundwater using a cyclodextrin solution.		
July 2003	CH2M HILL	Summary of Site 11 Cyclodextrin Pilot Study Post-Treatment Groundwater Sampling.	To assess the impact of the cyclodextrin solution on the groundwater. Groundwater samples were collected and analyzed for VOCs		
November 2003	CH2M HILL	NAB Little Creek Sites 11, 11a, and 13 Membrane Interface Probe Investigation and Associated Confirmation Sampling	MIP investigation and groundwater sampling to further assess the impact of the cyclodextrin solution on the groundwater.		

#### Table 1-2 Summary of VOC RME Cancer Risks and Hazard Indices Site 11, NAB Little Creek Virginia Beach, Virginia

		· · · · · · · · · · · · · · · · · · ·	<u> </u>	CORCe with Concer	T	T
Pacantar	Media	Exposure Route	Cancer Risk	COPCs with Cancer Risks >10 <sup>-4</sup>	Hazard Index	COPCs with HI > 1
Receptor	meuia	Exposure Route	Cancer Risk	RISKS > 10	nazaro inoex	COPCS With HI > 1
		Ingestion	NA NA		5.5E+02	1,1,1-Trichloroethane, 1,1- Dichloroethene, 1,2-Dichloroethene (total), Carbon tetrachloride, Trichloroethene, Vinyl chloride,cis-1,2- Dichloroethene
	l i	•		Carbon tetrachloride,	<del></del>	
Future Resident Adult	Groundwater	Inholotion	1 25 02	Trichloroethene, Vinyl chloride, 1,1,2- Trichloroethane, 1,2- Dichloroethane, Chloroform, Methylene	1.45.02	1,1-Dichloroethane, 1,1- Dichloroethene, 1,2-Dichloropropane, Vinyl chloride, trans-1,2- Dichloroethene, Carbon tetrachloride
		Inhalation	1.3E-02	chloride	1.1E+02	and Methylene chloride  1,2-Dichloroethene (total), Carbon
		Dermal Contact	NA NA		8.3E+01	tetrachloride, Vinyl chloride,cis-1,2- Dichloroethene
		Total	1.3E-02		7.5E+02	
		Receptor Total	1.3E-02		7.5E+02	
Future Resident Child	Groundwater	Ingestion	NA		1.3E+03	1,1,1-Trichloroethane, 1,1- Dichloroethane, 1,1-Dichloroethene, 1,2-Dichloroethene (total), Carbon tetrachloride, Methylene chloride, Trichloroethene, Vinyl chloride,cis-1,2- Dichloroethene, trans-1,2- Dichloroethene
		Inhalation	NA		NA	
		Dermal Contact Total	NA NA		1.9E+02 1.5E+03	1,2-Dichloroethene (total), Carbon tetrachloride, Trichloroethene, Vinyl chloride,cis-1,2-Dichloroethene
		Receptor Total	NA NA		1.5E+03	
		Ingestion	6.0E-02	Carbon tetrachloride, Methylene chloride, Trichloroethene, Vinyl chloride	NA NA	
Future Resident Child/Adult	Groundwater	Inhalation	NA		NA	
Offinar Addit		Dermal Contact Total	3.5E-03 6.3E-02	Carbon tetrachloride, Vinyl chloride	NA NA	
		Receptor Total	6.3E-02		NA NA	
Future Industrial Worker	Groundwater	Ingestion	6.1E-04	Carbon tetrachloride, Vinyl chloride	2.0E+02	1,2-Dichloroethene (total), Carbon tetrachloride, Trichloroethene, Vinyl chloride,cis-1,2-Dichloroethene
r uture muustriai vvorker		Inhalation Dermal Contact	NA NA		NA NA	
		Total	6.1E-04		2.0E+02	
		Receptor Total	6.1E-04		2.0E+02	
		Ingestion	NA 0.55.05		NA COE+00	
Future Construction Worker	Groundwater	Inhalation  Dermal Contact	3.5E-05 8.3E-05		6.9E+00 5.5E+01	1,1-Dichloroethane, Vinyl chloride 1,2-Dichloroethene (total), Carbon tetrachloride, Vinyl chloride
		Total	1.2E-04		6.2E+01	The second of th
		Receptor Total	1.2E-04		5.5E+01	

Note: Pentachlorophenol exceeds acceptable risks associated with residential dermal exposure to groundwater and will be addressed with this FS.

# Table 2-1 Summary of Groundwater PRGs Site 11, NAB Little Creek Virginia Beach, Virginia

	PRG	4.
COPC	(UG/L)	SOURCE
1,1,1-Trichloroethane	200	MCL
1,1,2-Trichloroethane	5	MCL
1,1-Dichloroethane	2,900	Calculated PRG*
1,1-Dichloroethene	7	MCL
1,2-Dichloroethane	5	MCL
1,2-Dichloropropane	5	MCL
Carbon tetrachloride	5	MCL
Chloroform	80	MCL
Methylene chloride	5	MCL
Trichloroethene	5	MCL
Vinyl chloride	2	MCL
cis-1,2-Dichloroethene	70	MCL
trans-1,2-Dichloroethene	100	MCL

<sup>\*</sup>PRG Calculation is provided in Appendix C and is based on USEPA Guidance Document. USEPA. December 1991. Risk Assessment Guidance for Superfund, Volume 1 -Human Health Evaluation Manual (Part B, Development of Risk-based Preliminary Remediation Goals). EPA/540/R-92/003.

#### Table 3-1

# Alternative 2 - ERD

# **Proposed Performance Monitoring Schedule**

# Site 11, NAB Little Creek

#### Virginia Beach, Virginia

	virginia beach, virginia		
Parameter	First Year	Year 1-14	Year 15+
Microbiological parameters:			
(Dehalococcoides, Dehalobacter, BAV-1*, and			
phospholipid fatty acids)	Semi-annually	NA	NA
priorphic facty dollar)	Communication		
Field parameters:	Monthly for first 6 months		
(pH, temperature, dissolved oxygen, oxidation-	Bimonthly rest of year	Quarterly	
reduction potential, specific conductance)	(9 events)	(4 events per year)	Annually
	(3.3.1.1.1.1)	( v a v a v v a p a v y a a v y	
	Monthly for first 6 months		
Total organic carbon, methane, ethane, ethene,	Bimonthly rest of year	Quarterly	
volatile organic compounds	(9 events)	(4 events per year)	Annually
0			
Geochemical parameters:	,		
(dissolved iron, dissolved manganese, sulfate,			
sulfide)	Bimonthly	Semi-annually	Annually
	Monthly for first 6 months		
	Bimonthly rest of year		
Volatile fatty acids	(9 events)	Annually	Annually
voidine ratty dolus	(a evenus)	Aimually	Allitually

<sup>\*</sup> Analysis of functional gene for strain BAV-1, which is associated with the reductive dechlorination of vinyl chloride

# Table 3-2

# Alternative 3 - ERH & ERD

# **Proposed Performance Monitoring Schedule**

# Site 11, NAB Little Creek

# Virginia Beach, Virginia

Parameter	First Year	Year 1-14	Year 15+
Microbiological parameters:			
(Dehalococcoides, Dehalobacter, BAV-1*, and			
phospholipid fatty acids)	Semi-annually	NA	NA
	Monthly for first 6 months, with 2		
Field parameters:	events in month four.		
(pH, temperature, dissolved oxygen, oxidation-	Bimonthly rest of year	Quarterly	
reduction potential, specific conductance)	(10 events)	(4 events per year)	Annually
	Monthly for first 6 months, with 2		
	events in month four.		
Total organic carbon, methane, ethane, ethene,	Bimonthly rest of year	Quarterly	
volatile organic compounds	(10 events)	(4 events per year)	Annually
Geochemical parameters:			
(dissolved iron, dissolved manganese, sulfate,			
sulfide)	Bimonthly	Semi-annually	Annually
	Monthly for first 6 months, with 2		
	events in month four.		
	Bimonthly rest of year		
Volatile fatty acids	(10 events)	Annually	Annually

<sup>\*</sup> Analysis of functional gene for strain BAV-1, which is associated with the reductive dechlorination of vinyl chloride

# Table 4-1 Detailed Evaluation of Remedial Alternatives Site 11 FS, NAB Little Creek Virginia Beach, Virginia

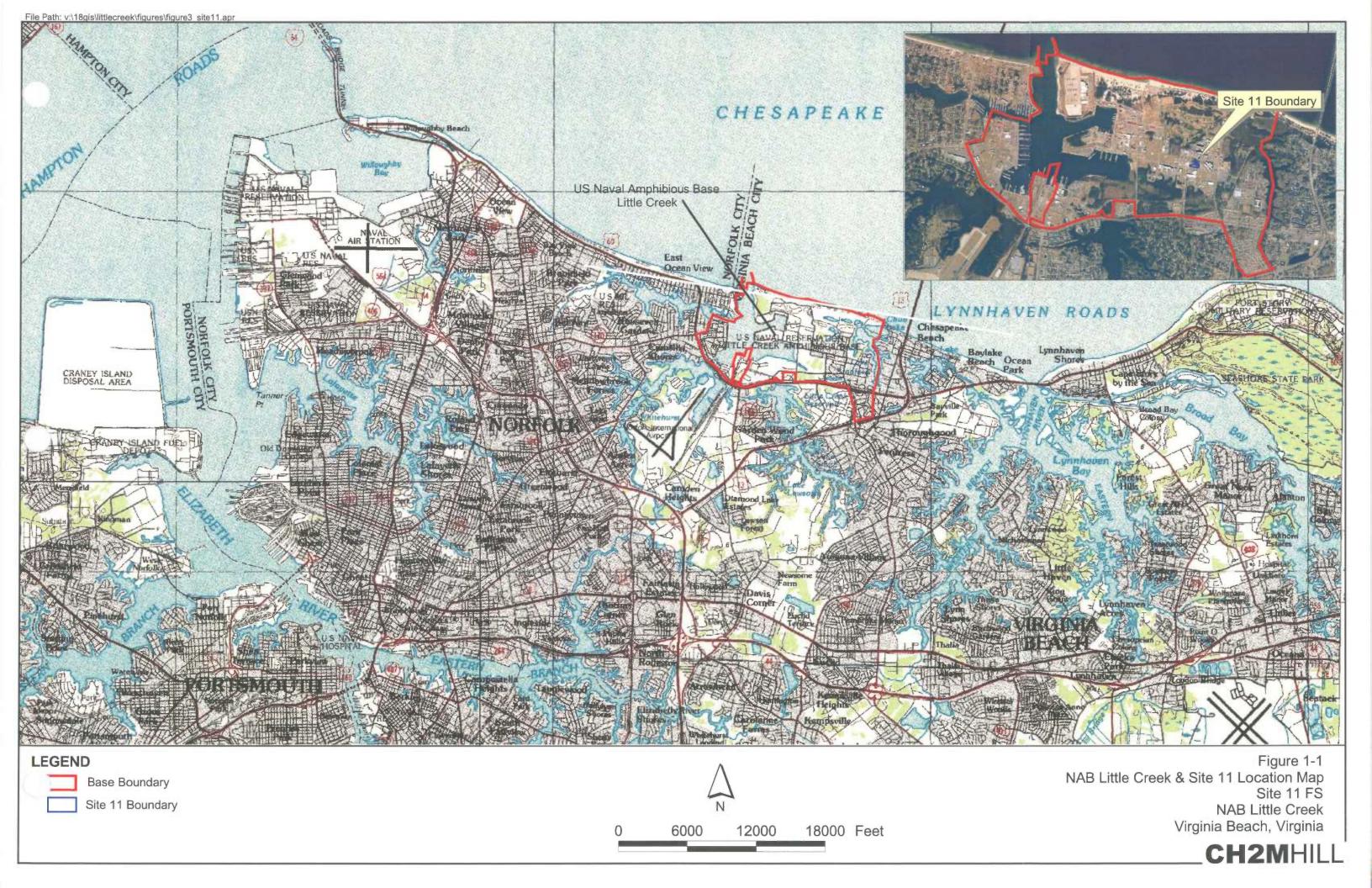
Alternative 1 No Action  of profective of human health and e environment.  oes not achieve ARARs.  oes not provide long-term fectiveness and permanance.  oes no provides reduction of toxicity, obility, and volume through	Alternative 2 ERD  Adequate protection of human health and the environment through groundwater treatment. LUCs and a groundwater monitoring will enusre protection is maintained.  Alternative 2 will comply with chemical-, location-, and action-specific ARARs.  This alternative would provide long-term effectiveness and permanance.  Alternative 2 is expected to reduce toxicity, mobility, and	Alternative 3 ERH & ERD  Adequate protection of human health and the environment through groundwater treatment. LUCs and a groundwater monitoring will enusre protection is maintained.  Alternative 3 will comply with chemical-, location-, and action-specific ARARs.  This alternative would provide long-term effectiveness and permanance.
e environment.  Des not achieve ARARs.  Des not provide long-term fectiveness and permanance.  Des no provides reduction of toxicity, obility, and volume through	through groundwater treatment. LUCs and a groundwater monitoring will enusre protection is maintained.  Alternative 2 will comply with chemical-, location-, and action-specific ARARs.  This alternative would provide long-term effectiveness and permanance.	groundwater treatment. LUCs and a groundwater monitoring will enusre protection is maintained.  Alternative 3 will comply with chemical-, location-, and action-specific ARARs.
pes not provide long-term fectiveness and permanance.  Does no provides reduction of toxicity, obility, and volume through	specific ARARs.  This alternative would provide long-term effectiveness and permanance.	ARARs.
fectiveness and permanance.  oes no provides reduction of toxicity, obility, and volume through	permanance.	This alternative would provide long-term effectiveness and permanance.
obility, and volume through	Alternative 2 is expected to reduce toxicity, mobility, and	
eatment.	volume through treatment via substrate injection to enhance biological degradation of VOCs in the source area, the area proximal to LS11-MW10D, and as a biobarrier in the downgradient plume.	Alternative 3 is expected to reduce toxicity, mobility, and volume through treatment via ERH in the source zone and ERD in the source area, the area proximal to LS11-MW10D, and as a biobarrier in the downgradient plume.
o concerns for short term fectiveness.	Alternative 2 requires the installation of monitoring and injection wells and regularly scheduled injections and monitoring events throughout the life of the project. Potential exposures associated with Alternative 2 would be minimized with appropriate protective equipment.	Alternative 3 requires well abandonment, well installation, construction and operation of the ERH system, and regularly scheduled injections and monitoring events throughout the life of the project. Potential exposures associated with Alternative 3 would be minimized with appropriate protective equipment. Since ERH includes an ex situ component, engineering controls will be required.
o action is easily implemented.	ERD is a proven technology, Wells can be installed by an experienced environmental drilling company. Disruption to current land use will occur during well installation, injection, and groundwater monitoring events.  This alternative is administratively feasible.	ERH and ERD are proven technologies. Well abandonment and installation can be completed by an experienced environmental drilling company. Disruption to current land use will occur during well installation, injection, and groundwater monitoring events. Additionally, since the ERH system has an ex situ component daily use of the site will be impeded during operation and construction and engineering controls will be required. This alternative is administratively feasible.
	\$2,399,000	\$2,841,000
	action is easily implemented.	exposures associated with Alternative 2 would be minimized with appropriate protective equipment.  ERD is a proven technology. Wells can be installed by an experienced environmental drilling company. Disruption to current land use will occur during well installation, injection, and groundwater monitoring events.  This alternative is administratively feasible.

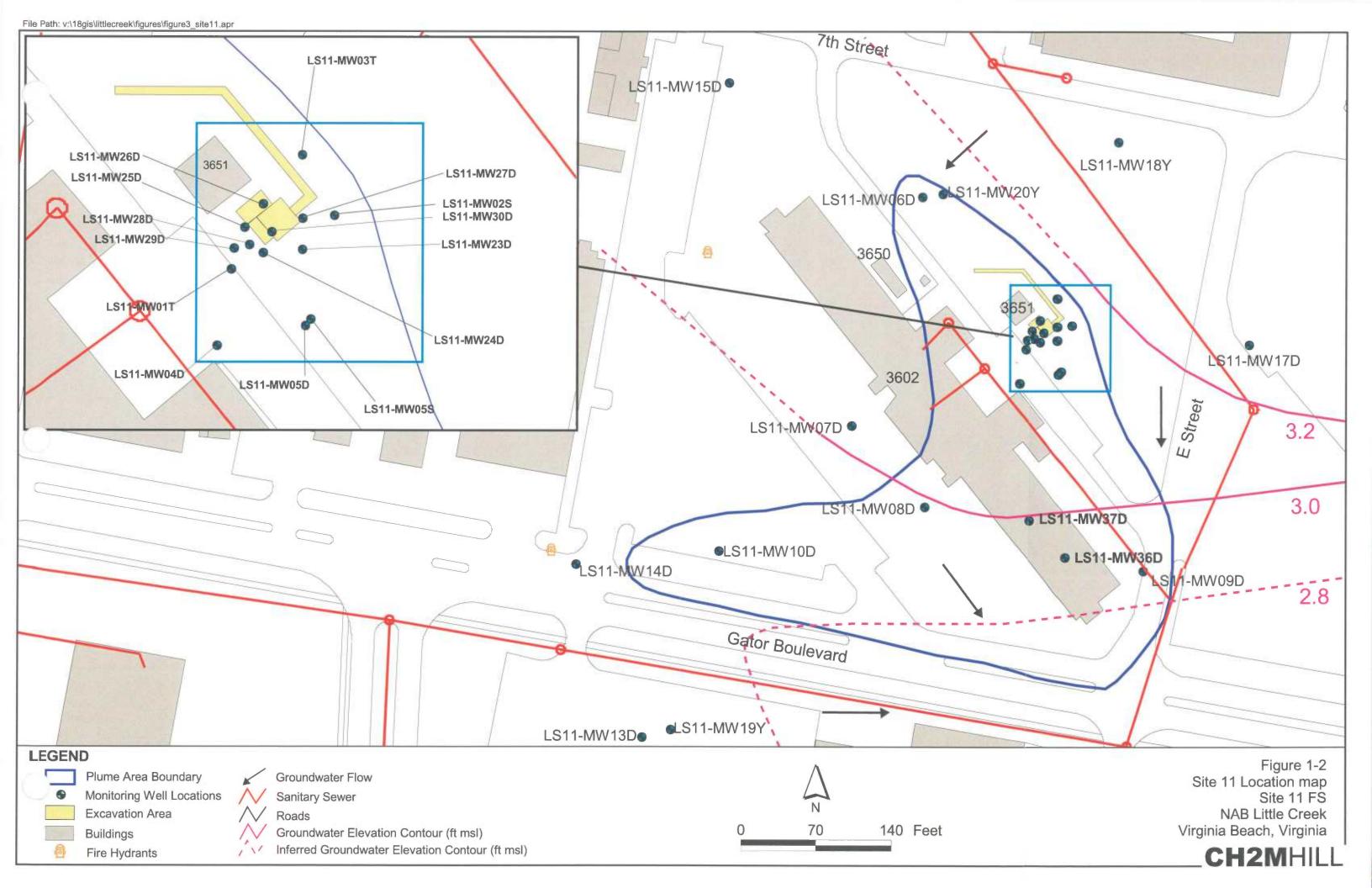
# Table 4-2 Comparative Analysis of Remedial Alternatives Site 11 FS, NAB Little Creek Virginia Beach, Virginia

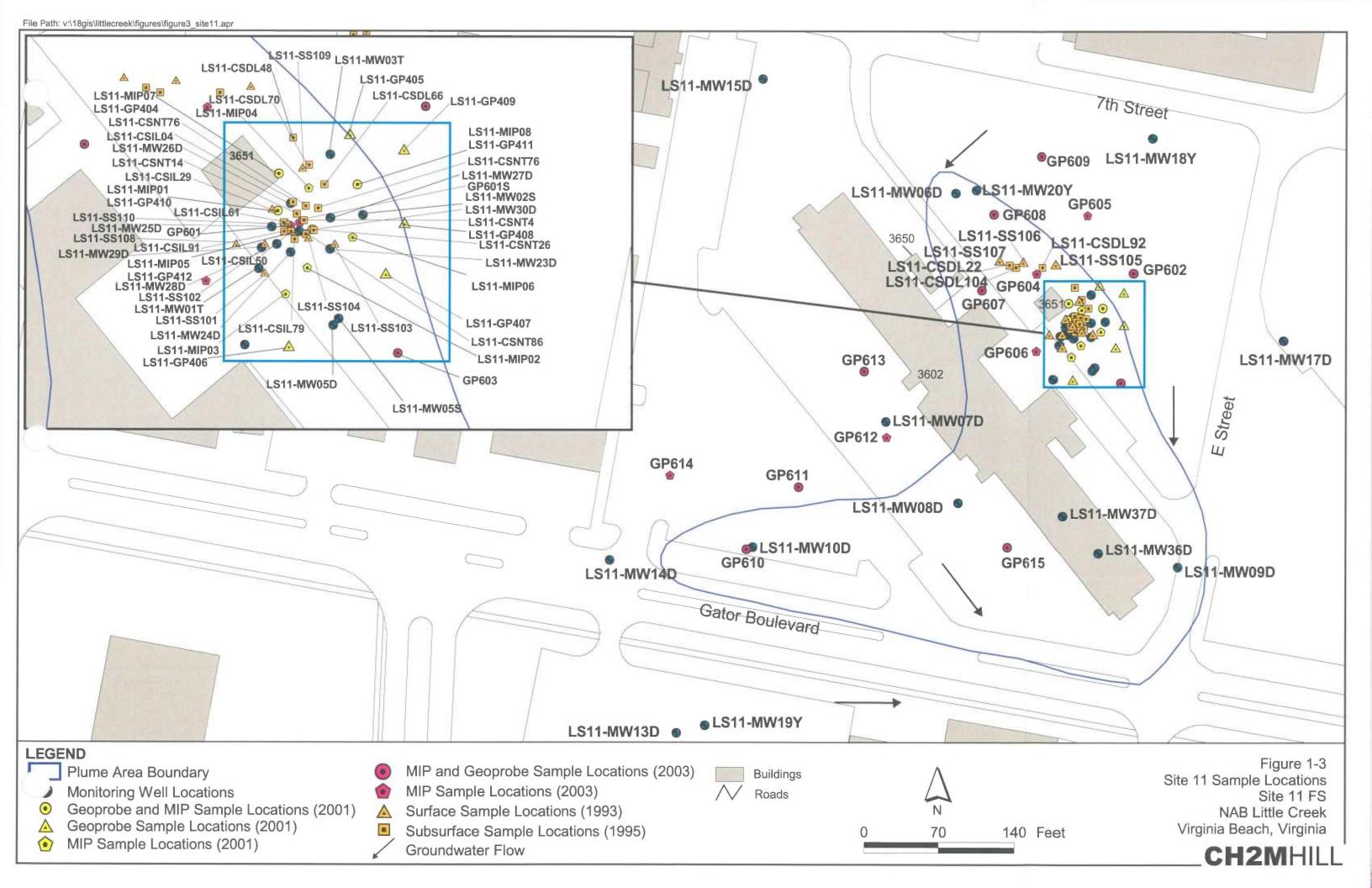
Evaluation Criteria	Alternative 1 No Action	Alternative 2 ERD	Alternative 3 ERH & ERD
Overall Protection of Human Health and the Environment	1	5	3
Compliance with ARARs	1	5	5
Long-Term Effectiveness and Permanence	1	5	5
Reduction of Toxicity, Mobility, and Volume Through Treatment	1	3	5
Short Term Effectiveness	5	3	1
Implementability	5	3	1
Cost	5	3	1 -
Total	19	27	21

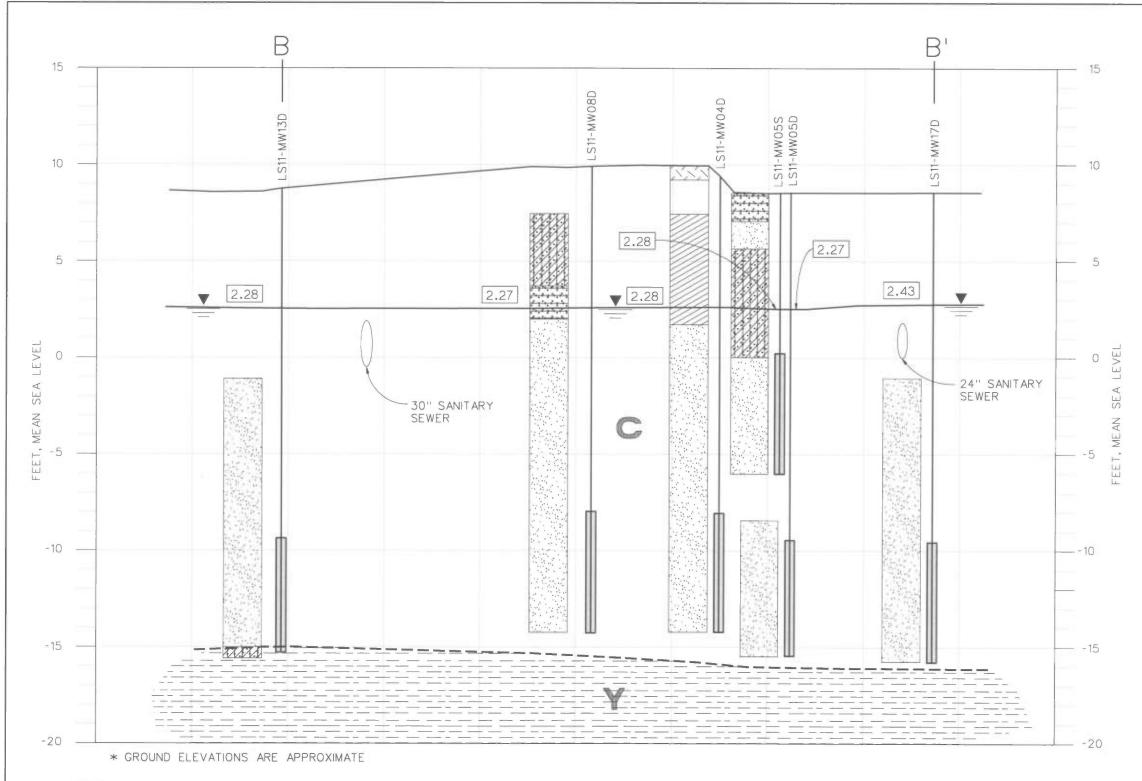
Qualitative comparative analysis of alternatives using a rating scale of 1, 3, and 5 (1 = lowest score, 5 = highest score) Shading designates the preferred alternative.

**Figures** 









NOTE: THIS CROSS SECTION IS INTERPRETIVE AND WAS PREPARED BY INTERPOLATION BETWEEN BORING LOCATIONS. ACTUAL CONDITIONS BETWEEN BORINGS MAY DIFFER FROM THOSE SHOWN HERE.

# LEGEND



WELL CONSTRUCTION DIAGRAM INCLUDING SCREENED INTERVAL



\_ WATER TABLE SURFACE: (NOVEMBER 3, 2000)



2.27 WATER LEVEL MEASURED IN FT ABOVE MSL NOVEMBER 3, 2000

# STRATIGRAPHY

- GROUND SURFACE



-- INFERRED CONTACTS BETWEEN HYDROSTRATIGRAPHIC UNITS



CONFINING UNIT



AQUIFER UNIT



COLUMBIA AQUIFER: UNDIFFERENTIATED SILTY SANDS/ CLAYEY SANDS/SANDS/GRAVEL/ INTERBEDDED CLAYS



YORKTOWN CONFINING UNIT: SILTS/CLAYS/SILTY CLAYS WITH SOME FINE SAND/SHELLS AND

# LITHOLOGY/USCS DESCRIPTION

FILL (ASPHALT, WOOD, ETC.)



SM: SILTY SAND



SP: WELL SORTED SAND (POORLY GRADED)



CLE CLAY



MHE CLAYEY, SANDY SILT

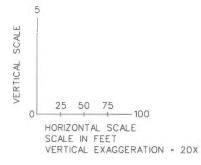
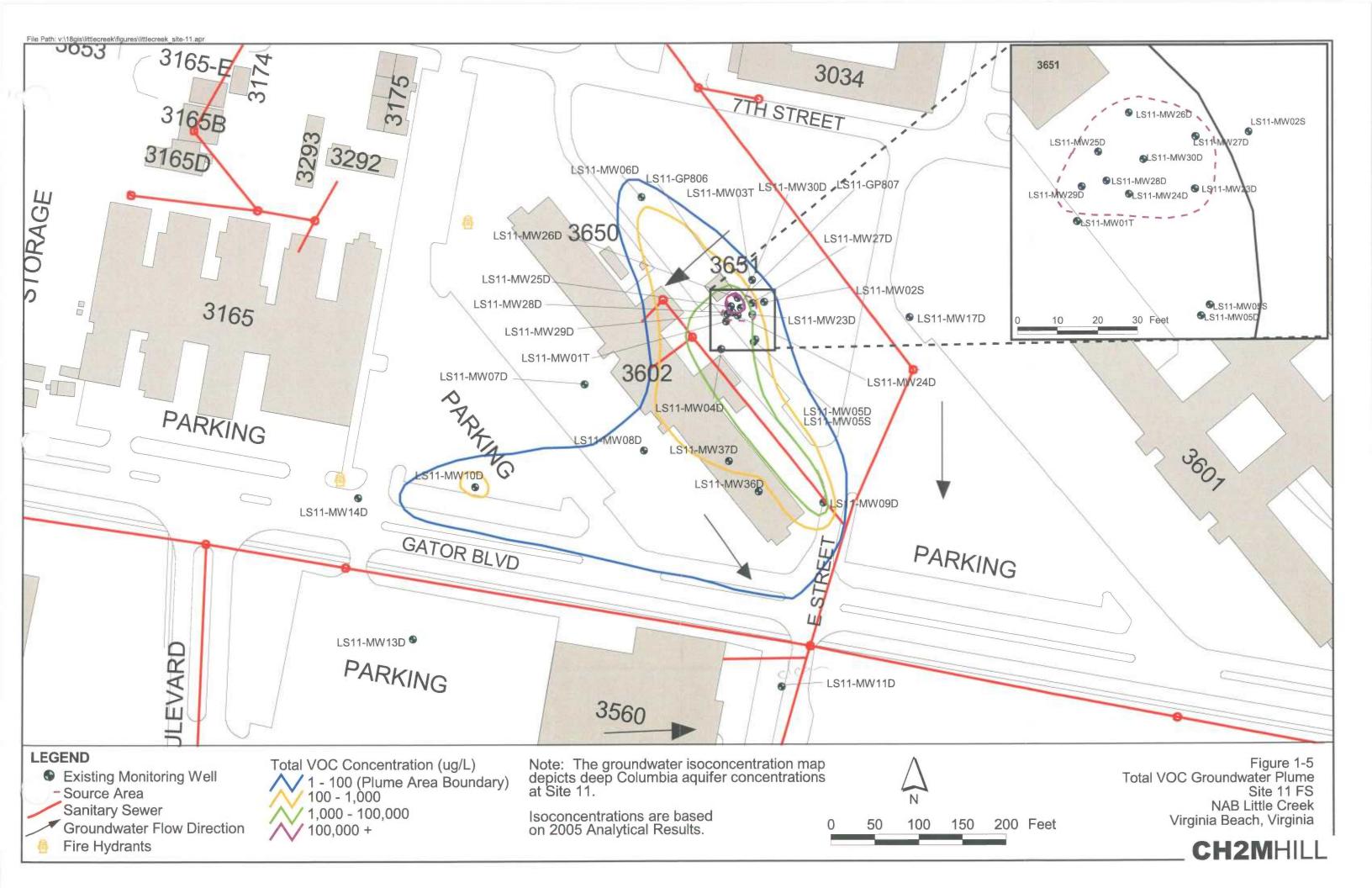
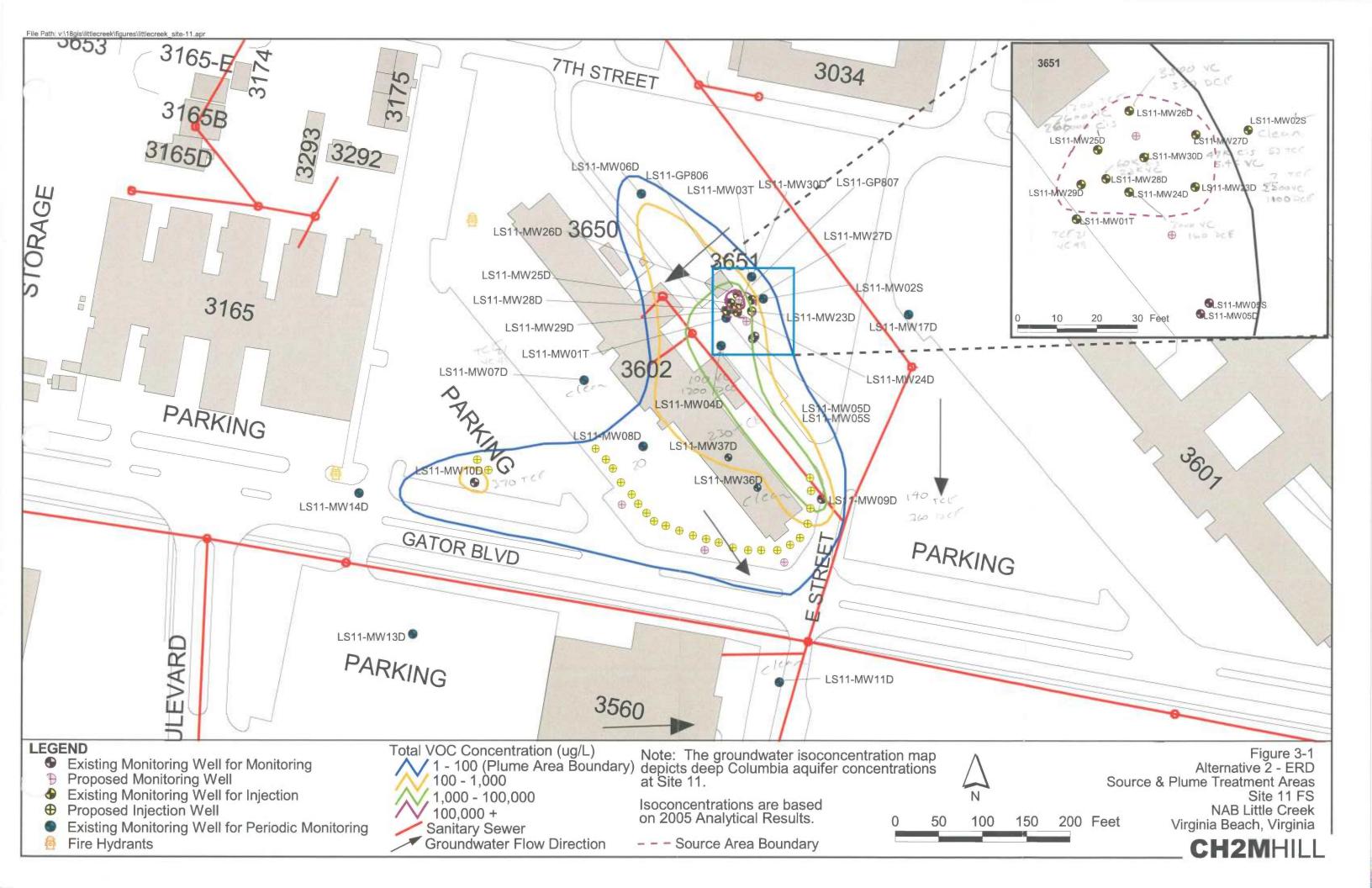
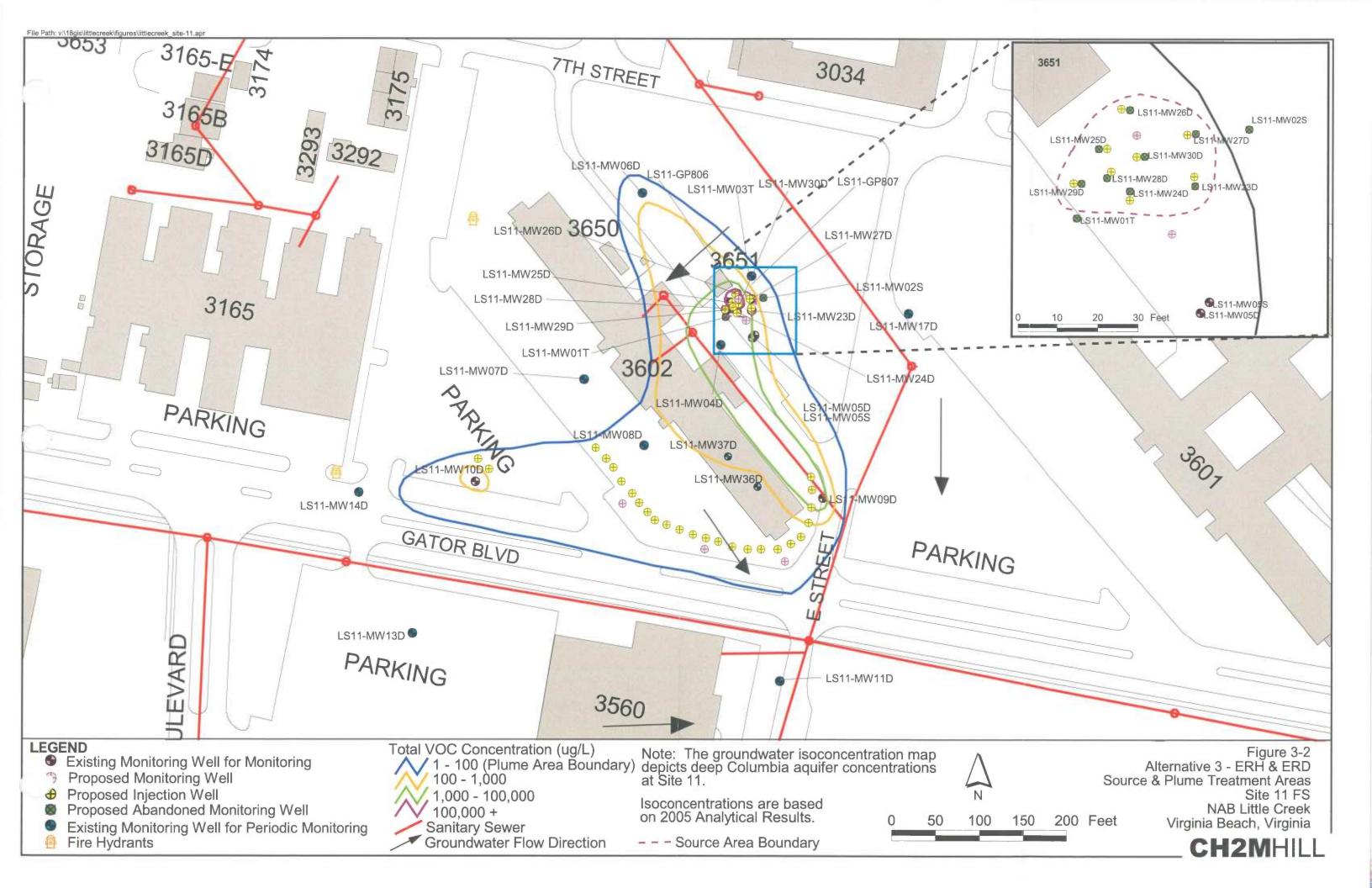


FIGURE 1-4 HYDROSTRATIGRAPHIC CROSS-SECTION B-B' WATER TABLE SURFACE (NOVEMBER 3, 2000) SITE 11 FS NAB LITTLE CREEK VIRGINIA BEACH, VIRGINIA







Appendix A Vapor Intrusion Assessment, Site 11

# Vapor Intrusion Assessment, Site 11, Naval Amphibious Base Little Creek

PREPARED FOR:

NAB Little Creek Partnering Team

PREPARED BY:

CH2M HILL

**COPIES:** 

DATE:

September 8, 2005

# Introduction

This technical memorandum summarizes the results of the vapor intrusion assessment of Building 3602, overlying Installation Restoration Site 11 at Naval Amphibious Base (NAB) Little Creek, Virginia Beach, Virginia. Site 11 is a chlorinated volatile organic compound (VOC) plume in groundwater underlying the School of Music Building 3602. To identify potential vapor intrusion pathways, a site visit was conducted in May 2005 to assist the Navy with evaluating whether the groundwater contamination poses a risk to building occupants. Based on the conclusions of the site visit, a sampling plan to further evaluate the indoor vapor intrusion pathway and potential human health risk was developed and agreed upon by the NAB Little Creek Tier I Partnering Team. The objectives for the field effort were to:

- Determine chlorinated VOC concentrations in groundwater in the upper portion of the shallow aquifer around the building through the collection of eight grab groundwater samples.
- Collect a grab sample of water from a sump in the basement of Building 3602 for VOC analysis to evaluate if a potential vapor intrusion pathway is present in the basement mechanical room.

Results from the shallow groundwater sampling, completed on June 27 and 28, 2005, were used to identify and assess human health risk due to potential vapor intrusion pathways. Site background, sampling methods, analytical results, risk assessment and the conclusions from the vapor intrusion risk assessment are summarized below.

# Site Background

Site 11 is located in the eastern portion of NAB Little Creek, near the intersection of 7th and E Streets (Figure 1). Site 11 consists of a VOC groundwater plume from a former in-ground concrete tank and associated piping used to neutralize plating solutions. The approximate extent of groundwater contamination is shown on Figure 1. The groundwater plume underlies Building 3602 (the School of Music) and Building 3651 (the former School of Music Plating Shop). Building 3602 is a rectangular 24,000 square foot building (approximately

The site visit to Building 3602 was conducted during normal business hours under sunny skies and with temperatures in the mid- to high-60 degree F range<sup>2</sup>. The survey included an inspection of the heating, ventilating and air conditioning (HVAC) system to identify the potential for depressurization relative to ambient conditions. Depressurization can potentially create advective flow of soil gas (and volatile contaminants in soil gas) into indoor air. The basement and first floor were inspected for potential vapor intrusion entry points and to evaluate ventilation characteristics in the inhabited areas. Additionally, a limited review of the building plans and chemical inventory for the building was completed to identify potential indoor sources for the constituents of interest in the subsurface (TCE) was also complete. Photographs taken during the site visit are presented in Attachment A.

### **Building Envelope**

Observations made during the site visit indicate that there are limited pathways for soil gas to intrude into the building. According to 1954 structural as-built construction drawings, the building was constructed on fill, approximately 2.5′ to 3′ above grade. The slab is approximately 6″ thick concrete overlain with 2 layers of vinyl floor tile. There were few penetrations through the first floor slab indicated on the drawings. The building was originally constructed with large open areas for crew quarters and bunks; the first floor had been renovated to create several small music practice rooms. Rain leaders from the roof do penetrate the slab at approximately 12 locations; however, in the subsequent renovations, the leaders appear to have been encased inside wall cavities. It was noted that exterior wall penetrations for steam piping for corridor radiators occurred at first floor level, which is approximately 3 feet above grade.

The below-grade mechanical room originally housed a water cooled chiller. In the 1988 renovation, the chiller was moved to an outdoor location. Currently, the room is used for storage. Based on the 1954 drawings and site observations, the mechanical room slab appears to be approximately 8 to 10 inches thick. Cracks in the slab were not observed; one penetration in the floor of the mechanical room, approximately 4" in diameter, was noted. There were small areas of dampness in the room and staining on the walls. Personnel have reported that the mechanical room had been flooded during a recent rainfall event.

A floor sump for the collection of steam condensate is located in the northeast corner of the below-grade mechanical room. The drawings indicate that the sump is fully lined with concrete. Based on a review of the drawings and water levels for the nearby monitoring wells, the sump and a portion of the mechanical room are below the water table.

Water seeps into the below-grade mechanical room during rain events at several openings around steam conduit pipes that enter the room at ground surface. Smoke testing indicated that the first floor was under a positive pressure relative to the mechanical room (smoke moved from the first floor to the mechanical room). Smoke moved upwards through one pipe penetration between the mechanical room and the first floor room containing air handling unit labeled AHU-3, indicating air from the basement does move into that air handling unit room.

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<sup>&</sup>lt;sup>2</sup> http://www.wunderground.com/history/airport/KORF/2005/5/23/MonthlyHistory.html#calendar

In addition to the eight DPT samples collected from the top of the aquifer adjacent to Building 3602, a grab sample was collected from the floor steam condensate sump in the mechanical room (basement) of Building 3602. This sample was collected using a peristaltic pump with low-flow sampling protocol. Samples were sent to an off-site laboratory (Mitkem Corporation, Warwick, RI) for analysis of Target Compound List (TCL) Low Concentration (LC) VOCs by Contract Laboratory Program (CLP) method OLC03.

## **Quality Control**

Quality assurance (QA)/QC samples were collected during the field activities in order to evaluate field methodologies (duplicates), evaluate whether cross contamination had occurred during sampling or shipping (equipment and trip blanks), establish field ambient conditions (field blanks), and measure sample-specific interference due to sample matrix (matrix spike (MS)/matrix spike duplicates (MSD). Two laboratory trip blanks and two equipment blanks were collected (one per day). One duplicate, one field blank, and one MS/MSD were collected during the sampling event.

For all samples, laboratory prepared, pre-preserved bottles for VOC analysis were filled completely with the aqueous sample to minimize aeration, and capped to prevent the entrapment of any air bubbles in the vial. All samples were labeled with the predetermined identification number. Samples were packed on ice for overnight shipment to an off-site laboratory. Temperature blanks were included in each cooler to confirm sample temperatures were less than 4 degrees Celsius when received by the laboratory.

Field samples and their corresponding analytical tests were recorded on chains-of custody (COC). Upon receipt of the samples by the laboratory, a comparison to the field information was made to determine if the samples, including the QA/AC samples, were documented correctly.

#### Decontamination and Investigation Derived Waste Procedures

All non-disposable sampling equipment, such as the direct push stainless steel rods and well screen, were decontaminated immediately after each use in accordance with applicable SOPs included with the Master Project Plans (MPP) Field Sampling Plan (FSP) checklist (CH2M HILL, August 2000). Investigation derived waste (IDW) generated during field activities, including purge water and decontamination fluids, were containerized in 55-gallon drums. The 55-gallon drums were properly labeled and are stored at a location designated by NAVFAC and NAB Little Creek pending disposal.

# **Analytical Results**

The location of the DPT groundwater samples and the basement sump sample is illustrated on Figure 3. A complete summary of the analytical results are provided in Table 1. There were no VOC detections in six of the eight DPT shallow groundwater samples. Only two VOCs were detected: chloromethane (1.7 micrograms per liter [ $\mu$ g/l] at GP704) and TCE (6.3  $\mu$ g/l at GP705). There were no VOCs detected in the grab sample collected from the basement sump.

required before the contaminant becomes available for volatilization into the overlying vadose zone. Once the volatilized contaminant reaches the building's zone of influence, convective air movement within the soil column transports the vapors through cracks between the foundation and the basement slab floor. This convective sweep effect is induced by a negative pressure within the structure caused by a combination of wind effects and stack effects due to building heating and mechanical ventilation.

The Johnson and Ettinger (1991) model was used to calculate risk-based screening levels in groundwater. These screening levels were based on standard default worker exposure assumptions (250 days/year exposure frequency, 25 years exposure duration). Use of these worker exposure assumptions probably overstates potential exposures and risks actually associated with activities in Building 3602, since most of the individuals in Building 3602 are active-duty personnel. The groundwater, building, and intake parameters used in the Johnson and Ettinger (1991) model are presented in Table 2.

Building parameters were obtained from the site visit and construction plans, while the soil and groundwater values were taken from boring logs collected during the current and previous site investigations. Some inputs are default values specified in the User's Guide for the Johnson and Ettinger model (USEPA 2004).

# **Key Assumptions**

Key assumptions that were to develop conservative screening levels include the following:

- The model assumes a potential pathway could be present via intrusion through foundation cracks, which may not be consistent with conditions at Building 3602.
- The model assumes that indoor air mixing is restricted to the first floor only. However, the air most likely mixes between floors, which would result in lower indoor air concentrations resulting from vapor intrusion.
- The VOC concentration in groundwater is assumed to be uniform under the building footprint. Most of the samples collected around the building did not detect VOCs in shallow groundwater.
- The model assumes the building is uniformly negatively pressurized relative to underlying soil gas. Actual building conditions observed during the site visit suggest that much of the building may be positively pressurized relative to outdoors.
- The worker receptor chosen in the model is assumed to spend 25 years in the building for 250 days per year--a conservative representation of the population in the building considering the normally short tour length of active duty military personnel (typically not more than five years).

The screening level for TCE was calculated using the cancer slope factor developed by the California Environmental Protection Agency (Cal-EPA, 2005). Updated cancer slope factors for TCE developed by USEPA currently are under review by the National Academy of Sciences (NAS, 2005). NAS is expected to issue a report on USEPA's risk assessment of TCE in June 2006. Pending the outcome of the NAS review, potential risks associated with TCE are being evaluated using toxicity values developed by Cal-EPA.

# References

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**Tables** 

Table 1
Shallow Groundwater and Basement Sump Analytical Results
Site 11 Vapor Intrusion Assessment
NAB Little Creek
Virginia Beach, Virginia

Sample ID	LS11-BSW1-05C	LS11-GP701-05C	LS11-GP702-05C	LS11-GP703-05C	LS11-GP704-05C	LS11-GP704P-05C	LS11-GP705-05C	LS11-GP706-05C	LS11-GP707-05C	LS11-GP708-05C
Sample Date	7/7/05	7/7/05	7/7/05	7/7/05	7/7/05	7/7/05	7/7/05	7/8/05	7/7/05	7/8/05
Chemical Name										770100
VOC- (1104)										
VOCs (UG/L) 1,1,1-Trichloroethane										
11,1,2,2-Tetrachloroethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,1,2,7-retrachioroethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,1,2-Trichloroethane	0.5 U 0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,1-Dichloroethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 ∪	0.5 U	0.5 U	0.5 U
1,1-Dichloroethene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 ∪	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,2,3-Trichlorobenzene	0.5 U	0.5 U 0.5 U	0.5 U	0,5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,2,4-Trichlorobenzene	0.5 U	0.5 U	0,5 U	0.5 U	0.5 ∪	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,2-Dibromo-3-chloropropane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,2-Dibromoethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 ∪
1,2-Dichlorobenzene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,2-Dichloroethane	0.5 U	commence of the contract of th	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,2-Dichloropropane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,3-Dichlorobenzene	0.5 U	0.5 U 0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
1,4-Dichlorobenzene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 ∪	0.5 U	0.5 U	0.5 U
2-Butanone	5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
2-Hexanone	5 U	5 U	5 U	5 U	5 U	5 U	5 U	5 U	5 U	5 U
4-Methyl-2-pentanone	5 U	5 U	The second commence of the second	5 U	5 U	<u>5 U</u>	5 U	5 U (	5 U	5 U
Acetone	5 U	5 U	5 U 5 U	5 U	5 U	5 U	5 U	5 U	5 U	5 U
Benzene	0.5 U	0.5 U	The second control of the control of	5 U	5 U	5 U	5 U	5 U	5 U	5 U
Bromochloromethane	0.5 U	0.5 U	0.5 U 0.5 U	0.5 U 0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Bromodichloromethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 ∪
Bromoform	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Bromomethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U 0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Carbon disulfide	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	A CONTRACTOR OF THE PARTY OF TH	0.5 U	0.5 U	0.5 U	0.5 U
Carbon tetrachloride	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Chlorobenzene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Chloroethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Chloroform	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Chloromethane	0.5 U	0.5 U	0.5 U	0,5 U	No. of the contract of the con	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
cis-1,2-Dichloroethene	0.5 U	0.5 U	0.5 U	0.5 U	1.7 0.5 U	0.5 U	1.7	0.5 U	0.5 U	0.5 U
cis-1,3-Dichloropropene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Cyclohexane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U 0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Dibromochloromethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U 0.5 U	0.5 U	0.5 U	0.5 U
Dichlorodifluoromethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Ethylbenzene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0,5 U	0.5 U 0.5 U	0.5 U
Isopropylbenzene	0.5 Ü	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Methyl acetate	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Methyl tert-butyl ether	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U		0.5 U
Methylcyclohexane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0,5 U   0.5 U	0.5 U
Methylene chloride	0.81 B	0.79 B	0.53 B	0.5 U	0.99 B	1.6 B	0.5 U	0.45 BJ	0.5 U	0.5 U
Styrene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.45 B3	0.5 U	0.5 U
Tetrachloroethene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Toluene	0.5 ∪	0.5 U	0,5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
trans-1,2-Dichloroethene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 Ü	0.5 U
trans-1,3-Dichloropropene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Trichloroethene	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	6.3	0.5 U	0.5 U	0.5 U
Trichlorofluoromethane	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Vinyl chloride	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U
Xylenes (Total)	0.5 Ū	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0.5 U	0,5 U

Notes

VOCs- Volatile Organic Compounds

U- Analyte not detected

B- Blank Contamination

TABLE 2
Groundwater to Indoor Air Parameters Used to Calculate Preliminary Cleanup Goals for Indoor Air Scenario Using the Johnson and Ettinger (1991) Model Site 11, Naval Amphibious Base, Little Creek

Symbol	Parameter	Description	Selected Value	Units	Sources
Ts	Average Soil/ Groundwater Temperature		14	°C	USEPA, 2004
L <sub>F</sub>	Depth Below Grade to Bottom of Enclosed Space Floor	This is the depth from soil surface to the bottom of the floor in contact with soil	15	cm	Based on observations during the Site Visit
Lwt	Depth Below Grade to Water Table		152.4	cm	Based on boring log information.
h <sub>A</sub>	Thickness of Soil Stratum A		152.4	cm	Thickness of soil stratum A is assumed to be consistent with average depth to groundwater at combined on- and offsite locations.
h <sub>B</sub>	Thickness of Soil Stratum B		NA	cm	Not Used
h <sub>C</sub>	Thickness of Soil Stratum C		NA	cm	Not Used
	Soil Stratum Directly Above Water Table		Α	unitless	Consistent with the deepest stratum with a specified thickness ( $h_A$ ).
	SCS Soil Type Above Water Table		S	unitless	Soils are assumed to be sand based on grain size data from borings around the building.
	Soil Stratum A SCS Soil Type	Used to estimate soil vapor permeability	S	unitless	
k,	User-defined Soil Vapor Permeability	A parameter associated with convective transport of vapors within the zone of influence of a building. It is related to the size and shape of connected soil pores	1 x 01 <sup>-7</sup>	cm²	Value calculated within the model and is consistent with sand.
$\rho_b^A$	Stratum A Soil Dry Bulk Density		1.66	g/cm <sup>3</sup>	Default value for sand calculated in the model
n <sup>A</sup>	Stratum A Total Soil Porosity	Used with water-filled porosity to calculate air-filled porosity (see below)	0.375	unitless	Default value for sand calculated in the model
θ <sub>w</sub> ^	Stratum A Soil Water-filled porosity	Used with total porosity to calculate air-filled porosity (see below)	0.054	cm <sup>3</sup> /cm <sup>3</sup>	Default value for sand calculated in the model
$\rho_b^{\ B}$	Stratum B Soil Dry Bulk Density		NA	g/cm <sup>3</sup>	Not Used
n <sup>B</sup>	Stratum B Total Soil Porosity	Used with water-filled porosity to calculate air-filled porosity (see below)	NA	unitless	Not Used
$\theta_{\mathbf{w}}^{\;\;B}$	Stratum B Soil Water-filled porosity	Used with total porosity to calculate air-filled porosity (see below)	NA	cm <sup>3</sup> /cm <sup>3</sup>	Not Used
$\rho_b^{\ C}$	Stratum C Soil Dry Bulk Density		NA	g/cm <sup>3</sup>	Not Used

TABLE 3
Preliminary Cleanup Goals for Indoor Air Exposure Scenarios Calculated Using the Johnson and Ettinger (1991) Model Site 11, Naval Amphibious Base, Little Creek

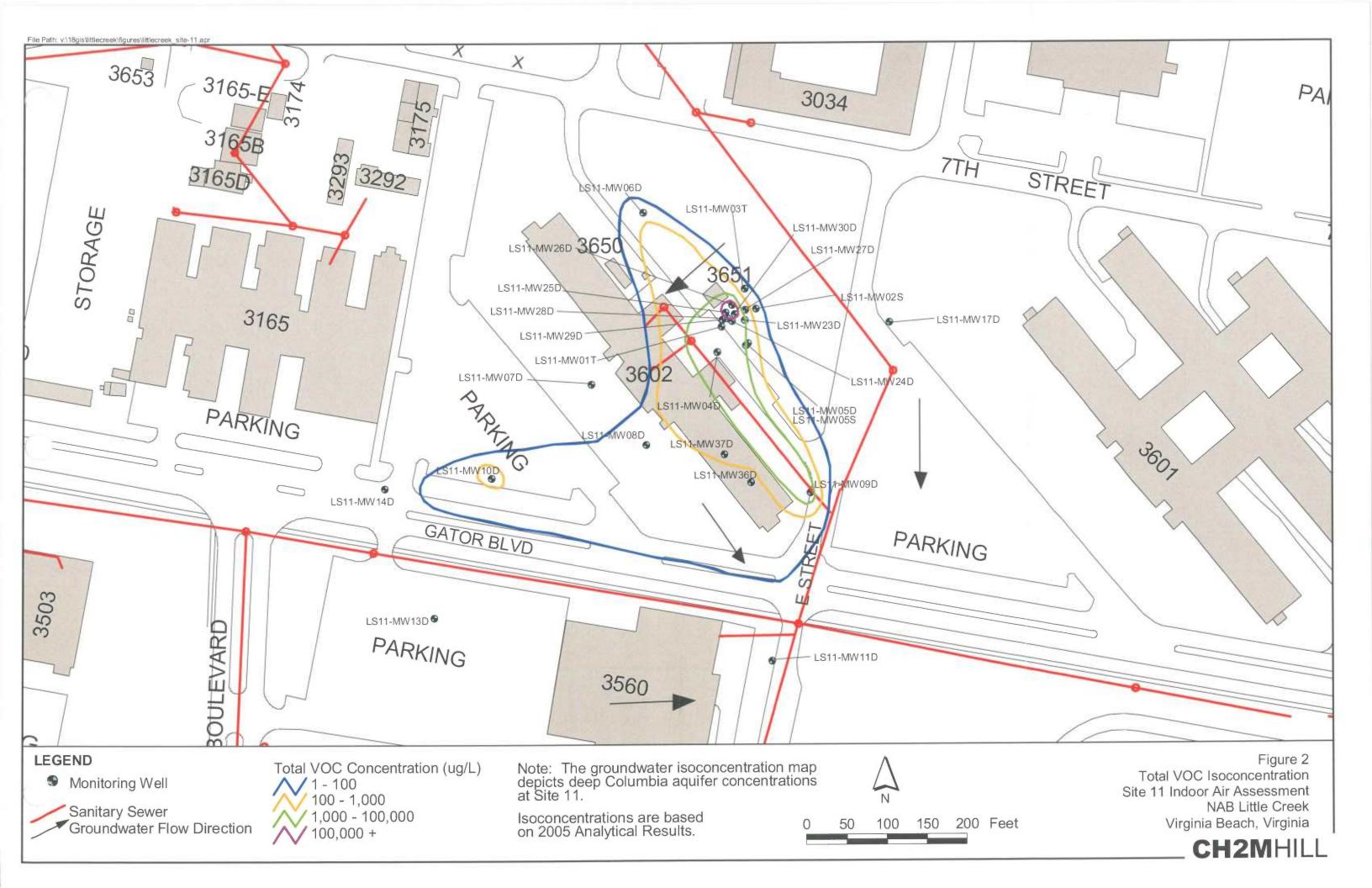
· _	Worker, Indoor Air				
Compound	Screening Value (μg/L)	Concentration in Groundwater (μg/L)			
TCE	29	6.3			
Chloromethane	41	1.7			

#### Notes:

Screening values in groundwater are based on 1 x 10<sup>-6</sup> excess lifetime cancer risk level.

TCE screening value calculated using Cal-EPA cancer slope factor.

**Figures** 



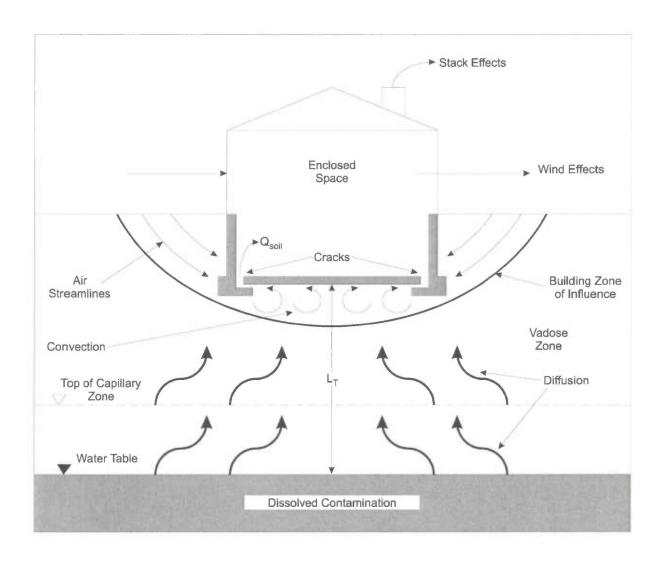


Figure 4
Conceptual Model of VOC Intrusion from Groundwater to Indoor Air
Vapor Intrusion Assessment
Site 11, NAB Little Creek
Virginia Beach, Virginia
CH2M HILL

Attachment A



Southern side of Building 3602 exterior. Building 3651 is in the foreground.



Entrance to the sub-grade mechanical room



Sump for condensate from steam pipe, which has an open drain leading into the sub-grade mechanical room. Mechanical room was flooded from rainfall collecting in this sump.



Sub-grade mechanical room, facing northeast. Sump is in the background.



Close up of sump.



Intrusion from mechanical room ceiling to first floor hallway. Smoke testing indicated that hallway is positively pressured relative to the mechanical room (smoke flowed from hallway to mechanical room).



Intrusion from mechanical room to first floor AHU-3. Smoke testing indicated that AHU-3 was negatively pressured relative to the mechanical room.



Intrusion from mechanical room to first floor hallway. Hallway is pressurized relative to mechanical room.



View from the roof, looking down on an air handling unit room (AHU-3) and outside air intake.



View of Building 3651 from roof of Building 3602.



Supply vent into first floor rehearsal room (approximately 4' by 8').



Return from first floor rehearsal room.

Appendix B Summary of 2005 Pre-Feasibility Study Investigations, Site 11

# Summary of 2005 Pre-Feasibility Study Investigations at Site 11 School of Music Plating Shop, Naval Amphibious Base Little Creek, Virginia Beach, Virginia

PREPARED FOR:

NAB Little Creek Partnering Team

PREPARED BY:

CH2M HILL

DATE:

February 9, 2006

### **Background**

This Technical Memorandum presents a summary of pre-Feasibility Study (FS) investigations conducted at Site 11 School of Music Plating Shop at Naval Amphibious Base (NAB) Little Creek, Virginia Beach, Virginia. Site 11 is a volatile organic compound (VOC) groundwater plume associated with a former Music Plating Shop (Building 3651) neutralization tank. A Supplemental Remedial Investigation (SRI) was completed in June 2004. As part of development of the Feasibility Study (FS) additional data needs were identified. In March 2005 a sampling event was completed to provide a current complete round of groundwater data. An additional sampling event was conducted in October 2005 to better define site characteristics associated with in situ remedial design technologies. This technical memorandum documents results of the pre-FS investigations at Site 11.

### **Objectives**

The objectives of the March 2005 investigation included:

- Installation of two directional surficial aquifer wells beneath School of Music Building,
- Collection of groundwater samples from 26 existing monitoring wells and two newly installed monitoring wells, and
- Collection of a complete round of groundwater levels at all site wells.

The objectives of the October 2005 investigation were:

- Collection of eight groundwater samples from seven existing monitoring wells and one Direct Push Technology (DPT) location and
- Collection of 17 soil samples at seven locations using DPT.

### **Methods**

### March 2005 and October 2005 Groundwater Sampling

In March 2005 groundwater samples were collected at 26 existing and two newly installed monitoring wells (Figure 1). In October of 2005 samples were collected at a total of seven monitoring wells and one Direct Push Technology (DPT) location (LS11-GP804) (Figure 1). Groundwater sampling was conducted using a peristaltic pump and low-flow purge method. Prior to sample collection, monitoring wells were purged until field water quality parameters (pH, conductivity, turbidity, dissolved oxygen (DO), temperature, oxidation/reduction potential (ORP), and salinity) stabilized. Results of the field water quality measurements were recorded in the field notebook and are provided in Table 1. The DPT groundwater sample was collected by advancing a stainless steel groundwater sampling tool to approximately 22 feet below ground surface (bgs), followed by purging to minimize turbidity to the maximum extent practical.

Groundwater samples were collected in laboratory prepared sample containers and analyzed at an offsite laboratory. The following parameters were analyzed in March and October: Target Compound List (TCL) volatile organic compounds (VOCs), total organic carbon (TOC), dissolved manganese, nitrate, nitrite, sulfate, methane, ethane, and ethene. Additionally, total and dissolved iron, total manganese, and sulfide were analyzed in March and chloride, alkalinity, and volatile fatty acids (VFAs) were analyzed in October. All dissolved inorganic samples were collected using an inline filter. Methane, ethane, ethene, and VOC bottles were filled completely to minimize aeration, and capped to prevent the entrapment of any air bubbles in the vial.

In addition to laboratory analysis, ferrous iron and sulfide were field analyzed using HACH test kits and DO was field analyzed using Chemets test kits during the October sampling event. Results were recorded in the field notebook.

#### **Quality Control Samples**

Quality assurance (QA)/quality control (QC) samples were collected during the field activities in order to evaluate field methodologies (duplicates), evaluate whether cross contamination had occurred during sampling or shipping (equipment and trip blanks), establish field ambient conditions (field blanks), and measure sample-specific interference due to sample matrix (matrix spike (MS)/matrix spike duplicates (MSD).

All samples were labeled with the predetermined identification number. Samples were packed on ice for overnight shipment to an off-site laboratory. Temperature blanks were included in each cooler to confirm sample temperatures were less than 4 degrees Celsius when received by the laboratory.

Field samples and their corresponding analytical tests were recorded on chains-of custody (COC). Upon receipt of the samples by the laboratory, a comparison to the field information was made to determine if the samples, including the QA/AC samples, were documented correctly.

#### **Decontamination and Investigation Derived Waste Procedures**

All non-disposable sampling equipment, such as the direct push stainless steel rods and well screen, were decontaminated immediately after each use in accordance with applicable standard operating procedures (SOPs) included with the Master Project Plans (MPP) Field Sampling Plan (FSP) checklist (CH2M HILL, August 2000). Investigation derived waste (IDW) generated during field activities, including purge water and decontamination fluids, was containerized in 55-gallon drums. The 55-gallon drums were properly labeled and were stored at a location designated by NAVFAC and NAB Little Creek prior to disposal at an approved facility.

### March 2005—Monitoring Well Installation

Two directional wells (LS11-MW36D and LS11-MW37D) were installed in the shallow aquifer using 4 ¼" hollow stem augers. The wells were installed around the perimeter of Building 3602 (Figure 1) at a 45° angle to better determine the extent of the VOC plume existing beneath Building 3602. Monitoring well construction diagrams and boring logs are provided in Attachment A.

Monitoring wells were constructed of 2-inch inner diameter, Schedule 40, polyvinyl chloride (PVC) screen and riser. Well screens consisted of machine slotted (0.01-inch) PVC, were prepacked with filter sand, and were placed in the entire length of the water column and set at the top of the Yorktown Confining Unit. A 2 foot bentonite layer was placed at the top of the sand pack. Following hydration of the bentonite layer, a cement-bentonite grout was placed in the remaining annular space. The monitoring wells were completed flush to ground surface with a watertight steel cover. A locking watertight cap was placed on the PVC pipe and the monitoring wells were marked with an identification numbers.

The newly installed wells were developed using a combination of surging throughout the well screen and pumping until the physical and chemical parameters of the discharge water met the requirements discussed in the Groundwater Sampling Section above.

### March 2005—Water Level Survey

To obtain the most consistent water level measurements, all water levels were taken concurrently on the last day of the investigation (April 1, 2005). Prior to taking water level measurements, the well cap was opened and the well was allowed to re-equilibrate. Top of casing elevations were used in conjunction with depth to water information to compute water table elevations. The station identification (ID) and depth to water below top of the PVC well casing were recorded in the field book. The results of the water level survey are summarized in Figure 2, Table 2.

### October 2005—Soil Sampling

Using DPT, 19 soil samples were collected at various depths from seven locations (Figure 1, Table 3). Soil samples were collected continuously to the depth of the Yorktown Confining Unit (approximately 23 feet below ground surface (bgs)) using clean, 4-foot, disposable acetate sleeves. Soil boring logs are provided in Attachment A. Soil descriptions including grain size, Unified Soil Classification System (USCS) group symbol, color (according to the Munsell Soil Color Chart), moisture content, density, and hardness were recorded in the

field notebook. A photoionization detector (PID) was used to field screen soils. Head space PID readings were collected from each two-foot sample interval in order to better determine the interval that contained the highest PID reading. A composite sample from each two-foot interval was placed in a 1-quart Ziploc bag. After a period of 5 to 10 minutes, a small opening large enough to accommodate the PID probe was made in the seal of the bag in order to measure concentrations while minimizing escape of volatiles. PID readings were recorded in the field notebook, and samples with the highest PID readings were submitted for laboratory analysis. Samples were collected from the following three depth intervals:

- Upper Columbia (UC) These samples were collected from the shallow portion of the Columbia Aquifer just below the water table or within the two-foot interval at approximately 8 ft to 16 ft bgs.
- Lower Columbia (LC) These samples were collected from the two-foot interval within the lower five-feet of the Columbia Aquifer just above the Yorktown Confining Unit.
- Yorktown Confining Unit (YC) These samples were collected in the first one to two feet of the Yorktown Confining Unit.

Sample IDs included UC, LC, or YC to designate the depth at which the sample was collected.

Humidity associated with rainy conditions during the sampling event limited the use of the PID. When the interval of soil containing the highest VOC concentrations could not be established using the PID, the soil sample was collected from the bottom two feet of the specified depth interval.

Soil samples were collected for the following analysis: VOCs, TOC, total oxidant demand (TOD), and microbial analysis [*Dehalococcoides* (DHC), *Dehalobacter* (DHB), a functional gene associated with DHC strain BAV1 (BAV1 R-Dase), and phospholipid fatty acids (PLFAs)]. In order to preserve organic materials in soils, grab samples were collected for VOC and TOD analysis. The interval of soil collected for TOC and microbial analysis was homogenized in a stainless steel bowl prior to placement in sample jars. Soil samples collected for geotechnical analysis (grain size, bulk density, and porosity) were collected in acetate sleeves, capped on the ends, and sent to the laboratory. Soil samples collected for laboratory analysis were stored, transported, and tracked using the same procedures described above for groundwater samples.

### **Analytical Results**

The overall objective of these investigations is to provide the information necessary to select the remedial alternatives identified in the FS. The results of the investigations are presented below.

Analytical groundwater results are consistent between the March and October sampling events. A summary of the data is provided in Table 4 and a complete data set is provided in Attachment B. The generally low concentrations of DO, ORP, and nitrate, and the generally elevated concentrations of ferrous iron suggest the aquifer is a reduced environment (Tables 1 and 4). Constituents with concentrations exceeding the maximum contaminant level

(MCL) include the following VOCs: 1,1,1-trichloroethane (TCA), 1,1,2-TCA, 1,1-dichloroethene (DCE), 1,2-dichloroethane (DCA), 1,2-DCE (cis and trans), 1,2-dichloropropane, carbon tetrachloride, methylene chloride, trichloroethene (TCE), and vinyl chloride (VC) (Table 5). Total VOC concentrations exceed 100,000 µg/L (micrograms/liter) in the source area. Isoconcentration lines for total VOCs are shown in Figure 3.

Soil analytical results are provided in Attachment B and are summarized in Table 5. VOC concentrations in soils collected from the upper portion of the Columbia Aquifer were less than 30  $\mu g/L$ , with the exception of LS11-SB805 which had a TCE concentration of 55  $\mu g/L$ . From those samples collected in the lower portion of the Columbia Aquifer, the greatest VOC concentration was 600  $\mu g/L$  for cis-1,2-DCE from sample location LS11-SB801. The greatest VOC concentrations were found in the upper portion of the Yorktown Confining Unit and exceeded 10,000  $\mu g/L$ , with the highest concentration found at sample location LS11-SB802 (25,000  $\mu g/L$  TCE). In the samples collected from the upper portion of the Yorktown Confining Unit, TOC concentrations were equal to or greater than 10,000 mg/kg (milligram/kilogram). Although TOC concentrations are greater than would be expected given the soil characteristics, these elevated concentrations can be attributed to the in situ use of cyclodextrin during the 2002 pilot study.

Microbial Insights conducted analysis of VFAs in groundwater and DHC, DHB, BAV1 R-Dase, and PLFAs in soil to evaluate microbial activity at the site. The results are provided in Attachment B. The presences of VFAs, DHC, DHB, BAV1 R-Dase, and bacterial biomass exceeding 1 × 106 cells/milliliter (mL) indicates the presence of healthy bacterial populations including those capable of reductive dechlorination.

TOD analysis was completed by Redox Tech using sodium persulfate as the oxidant. The results of the analysis are presented in Table 6. Oxidant demands for the lower portion of the Columbia Aquifer ranged from 1.9 to 3.7 grams/kilogram (g/kg) of sodium persulfate. The upper portion of the Yorktown Confining Unit has oxidant demands ranging from 11 to greater than 19.5 g/kg of sodium persulfate. These values are not unexpected based on the elevated TOC concentrations in soil. Consequently, treatment with a technology such as *in situ* chemical oxidation would require an excessive amount of oxidant to overcome the site TOD.

Geotechnical analysis was conducted on two samples from the lower portion of the Columbia Aquifer and two samples from the upper portion of the Yorktown Confining Unit. Complete results, including the grain size distribution figures, are provided in Attachment B. The samples from the lower portion of the Columbia Aquifer were classified as silty sand (SM) and sand (SP), with a moisture content of 19.6 percent and 18.5 percent, and a porosity of 49.4 percent and 38.0 percent, respectively. The samples from the upper portion of the Yorktown Confining Unit were classified as sandy silt (ML) and silty clay (CH), with a moisture content of 43.0 percent and 55.9 percent, and a porosity of 52.0 percent and 60.6 percent, respectively.

**Tables** 

### Table 1 Water Quality Field Parameters (March and October 2005) Pre-Feasibility Study Investigations NAB Little Creek Virginia Beach, Virginia

Station ID	LS11-MW01T	LS11-MW02S	LS11-MW03T	LS11-N	1W04D	LS11-M	W05D	LS11-MW05S	LS11-MW06D	LS11-MW07D	LS11-MW08D	LS11-MW09D
Sample ID Sample Date	LS11-MW01T-05A 04/01/2005	LS11-MW02S-05A 04/01/2005	LS11-MW03T-05A 03/31/2005	LS11-MW04D-05A 03/30/2005	LS11-MW04D-05D 10/10/2005	LS11-MW05D-05A 03/30/2005	LS11-MW05D-05D 10/10/2005	LS11-MW05S-05A 03/30/2005	LS11-MW06D-05A 03/30/2005	LS11-MW07D-05A 03/30/2005	LS11-MW08D-05A 03/30/2005	LS11-MW09D-05A 03/30/2005
Field Parameters	0 110 112000	0.000										00/00/2000
Dissolved Oxygen (mg/L)	0.5	1.4	0.4	0.5	0.8	0.5	0	0.5	5.1	1	1.9	1.8
Depth to Water (ft)	5.35	3.83	3.29	6.35	6.42	5.27	5.43	4.87	3.75	5.85	6.96	6.11
ORP (mV)	2	101	-11	-68	-136	-183	-215	-60	163	229	287	173
Flow Rate (GPM)	0.052	0.066	0.052	0.066	0.053	0.066	0.079	0.066	0.066	0.066	0.066	0.066
Gallons Purged (GAL)	NA	NA	NA	2	2	2.5	4	1	1.5	2	2	. 2
Salinity (%)	0	0	0	0	0.03	0	0.02	0	0	0	0	0
pH	5.84	5.66	5.7	5.96	6.18	6.79	6.88	6.05	7.67	5.78	5.48	5.65
Specific Conductance (ms/cm)	0.374	0.361	0.306	0.629	0.639	0.529	0.515	0.472	0.371	0.176	0.169	0.326
Temperature (C)	14.7	12.8	13.9	17.6	21.97	15.6	20.06	13.4	16.8	16.8	15	15.8
Turbidity (NTU)	3.7	2.1	0	0	4.5	1.6	-0.1	42.1	0	0.4	4.7	1.1

Station ID	LS11-MW09D	LS11-N	/W10D	LS11-MW11D	LS11-MW13D	LS11-MW14D	LS11-MW17D	LS11-MW18Y	LS11-MW19Y	LS11-MW20Y	LS11-M	IW23D
Sample ID Sample Date	LS11-MW09D-05D 10/10/2005	LS11-MW10D-05A 03/29/2005	LS11-MW10D-05D 10/10/2005	LS11-MW11D-05A 03/29/2005	LS11-MW13D-05A 03/29/2005	LS11-MW14D-05A 03/29/2005	LS11-MW17D 03/29/2005	LS11-MW18Y-05A 03/29/2005	LS11-MW19Y-05A 03/29/2005	LS11-MW20Y-05A 03/30/2005	LS11-MW23D-05A 03/31/2005	LS11-MW23D-05D 10/10/2005
Field Parameters	<u>" </u>											
Dissolved Oxygen (mg/L)	1	2.4	1.5	2	3.3	1.6	1	6.6	0.7	0.5	0.5	0.5
Depth to Water (ft)	6.25	6.07	5.91	NA	NA	5.81	4.81	5.29	NA	3.89	4.8	5
ORP (mV)	-48	265	306	208	56	217	365	149	-59	-115	-331	-410
Flow Rate (GPM)	0.066	0.066	0.079	0.066	0.071	0.066	0.066	0.066	0.066	0.066	0.046	0.053
Gallons Purged (GAL)	2.5	2.5	3	2	2.5	NA	2	NA	2	1.5	NA	. 5
Salinity (%)	0.01	0	0.01	0	0	0	0	0	0	0	0	0.02
pН	5.86	5.32	5.27	5.68	10.05	5.77	6.07	9.08	10.93	7	8.78	8.33
Specific Conductance (ms/cm)	0.348	0.225	0.242	0.143	0.245	0.275	0.405	0.286	0.952	0.446	0.394	0.502
Temperature (C)	20.95	17.3	22.95	21.88	17.7	18.1	15.7	17.8	19.3	15	14.6	20.11
Turbidity (NTU)	2.6	0.2	4.7	0	3.3	0	0	0	1.9	1.9	5.3	0.5

Station ID	LS11-MW24D			LS11-M	1W26D	LS11-MW27D	LS11-MW28D	LS11-MW29D	LS11-MW30D	LS11-MW36D	LS11-MW37D	LS11-GP804
Sample ID Sample Date	LS11-MW24D-05A 03/31/2005	LS11-MW25D-05A 03/31/2005	LS11-MW25D-05D 10/11/2005		LS11-MW26D-05D 10/11/2005	LS11-MW27D-05A 03/31/2005	LS11-MW28D-05A 03/30/2005	LS11-MW29D-05A 03/31/2005	LS11-MW30D-05A 03/31/2005	LS11-MW36D-05A 04/01/2005	LS11-MW37D-05A 04/01/2005	LS11-GW804-LC 10/07/05
Field Parameters	3.A <sup>-</sup>	:										
Dissolved Oxygen (mg/L)	0.4	0.5	0.31	0.5	0.2	0.4	0.4	0.4	0.4	2	1.3	3.56
Depth to Water (ft)	5.55	5.12	5.22	4.61	4.77	4.88	5.18	5.1	4.7	8	10.41	NA
ORP (mV)	-151	-141	-167	-303	-221	-226	-197	-155	-286	151	156	-60
Flow Rate (GPM)	0.052	0.052	0.082	0.046	0.066	0.046	0.066	0.039	0.052	0.052	0.039	NA
Gallons Purged (GAL)	NA	NA	2	NA	NA NA	NA	1.5	NA	NA	NA	NA	NA NA
Salinity (%)	0.1	0.1	0.12	0	0.04	0	0.1	0	0	0	0	0.01
рН	6.44	6.63	6.65	7.93	7.04	7.22	7	6.59	7.53	5.37	5.42	5.25
Specific Conductance (ms/cm)	1.36	2.52	2.52	0.751	0.819	0.999	1.48	0.624	0.92	0.173	0.197	204
Temperature (C)	14.2	15.4	19.33	14.4	19.05	13.7	14.1	14	14.1	18.7	18.9	22.34
Turbidity (NTU)	13	0	1.8	0	8.3	0.5	0.2	0	1	3.8	4.1	423

### Table 2 Water Level Survey (April 2005) Pre-Feasibility Study Investigations NAB Little Creek Virginia Beach, Virginia

			<u> </u>
MONITORING WELL	TOP OF PVC (msl)	DEPTH TO WATER (ft)	GROUNDWATER ELEVATION (msl)
COLUMBIA AQU	IFER WELLS	. : : : : : : : : : : : : : : : : : : :	
LS11-MW01T	8.13	4.99	3.14
LS11-MW02S	6.97	3.81	3.16
LS11-MW03T	6.44	3.28	3.16
LS11-MW04D	9.2	6.09	3.11
LS11-MW05S	8.03	4.88	3.15
LS11-MW05D	8.36	5.22	3.14
LS11-MW06D	6.76	3.64	3.12
LS11-MW07D	8.86	5.79	3.07
LS11-MW08D	9.06	6.03	3.03
LS11-MW09D	8.88	6.06	2.82
LS11-MW10D	8.19	5.25	2.94
LS11-MW11D	9.89	7.32	2.57
LS11-MW13D	8.09	5.1	2.99
LS11-MW14D	8.57	5.66	2.91
LS11-MW17D	8.12	4.82	3.3
LS11-MW23D	7.59	4.46	3.13
LS11-MW24D	8.22	5.1	3.12
LS11-MW25D	7.92	4.78	3.14
LS11-MW26D	7.65	4.52	3.13
LS11-MW27D	7.6	4.46	3.14
LS11-MW28D	7.95	4.84	3.11
LS11-MW29D	8.05	4.93	3.12
LS11-MW30D	7.63	4.51	3.12
LS11-MW36D	9.17	*	NA
LS11-MW37D	9.00	*	NA
YORKTOWN AQU	JIFER WELLS		
LS11-MW18Y	8.75	4.88	3.87
LS11-MW19Y	8.38	4.35	4.03
LS11-MW20Y	7.05	3.26	3.79

msl = mean sea level

ft = feet

<sup>\* =</sup> Directional wells. Exact depth to water is estimated.

Table 3
Location and DPT Sample Depths (October 2005)
Pre-Feasibility Study Investigations
NAB Little Creek
Virginia Beach, Virginia

Location	Sample	Depth
LS11-GP801	LS11-SB801-UC	6-8'bgs
	LS11-SB801-LC	18-19.5'bgs
	LS11-SB801-YC	20-22'bgs
LS11-GP802	LS11-SB802-UC	13-15'bgs
	LS11-SB802-LC	18-20'bgs
	LS11-SB802-YC	20.5-22.5'bgs
LS11-GP803	LS11-SB803-LC	18-20'bgs
LS11-GP804	LS11-SB804-LC	22-24'bgs
	LS11GW804-LC	22-24*
LS11-GP805	LS11-SB805-UC	10-12'bgs
	LS11-SB805-LC	24-28'bgs
	LS11-SB805-YC	24-28'bgs
LS11-GP806	LS11-SB806-UC	8-10'bgs
	LS11-SB806-LC	16-18bgs
	LS11-SB806-YC	20-22'bgs
LS11-GP807	LS11-SB807-UC	13-15'bgs
	LS11-SB807-LC	20-22'bgs
	LS11-SB807-YC	22-24'bgs

<sup>\*</sup> Estimated sample interval

### Table 4 Detects in Groundwater (March and October 2005) Pre-Feasibility Study Investigations NAB Little Creek Virginia Beach, Virginia

tation ID	Y	LS11-GP804	LS11-MW01T	LS11-MW02S	LS11-MW03T	LS11-N	MWO4D	1	LS11-MW05D	·	LS11-MW05S	LS11-MW06D	LS11-MW07D	LS11-MW08D	,	LS11-MW09D	
Sample ID	MCL-	LS11-GP804 LS11-GW804-LC	LS11-MW01T-05A	LS11-MW02S	LS11-MW03T-05A	LS11-MW04D-05A	LS11-MW04D-05D	LS11-MW05D-05A	LS11-MW05D-05D	LS11-MW05DP-05D	LS11-MW05S-05A	LS11-MW06D-05A	LS11-MW07D LS11-MW07D-05A	LS11-MW08D LS11-MW08D-05A	LS11-MW09D-05A	LS11-MW09DP-05A	LS11-MW09D-05D
Sample Date	Groundwater	10/07/05	04/01/05	04/01/05	03/31/05	03/30/05	10/10/05	03/30/05	10/10/05	10/10/05	03/30/05	03/30/05	03/30/05	03/30/05	03/30/05	03/30/05	10/10/05
		10/07/05	1 04/01/05	04/01/05	1 03/3//03	03/30/05	10/10/05	1 03/30/03	10/10/05	10/10/03	03/30/03	1 03/30/03	03/30/05	03/30/05	1 03/30/05	03/30/03	10/10/03
Chemical Name			<del>-</del>	<del> </del>	l						ļ —	ļ					
				<b>-</b>	<del></del>						1						
VOCs (UG/L)				· · · · · · · · · · · · · · · · · · ·				100		<u> </u>	40.11	40.11	40.11				
1,1,1-Trichloroethane	200	2 J	13		10 U	320	64	190	28	27	10 U	10 U	10 U	1 J	280	240	42
1,1,2-Trichloroethane	5	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	1 J	10 U
1,1-Dichloroethane		2.5 J	35	10 U	10 U	600	340	280	160	180	10 U	10 U	10 U	2 J	150	150 J	54
1,1-Dichloroethene		10 U	2 J		10 U		77	29	. 8 J		10 U	1 J	10 U	. 4 J	220	240	30
1,2-Dichloroethane	5	10 U	10 U	10 U	10 U	1 J	10 U		10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 UJ	10 U
1,2-Dichloroethene (total)	70	NA NA	25	5 J	2 J		NA NA	1,000	NA	NA NA	10 U	26	10 U	29	540	560	NA NA
1,2-Dichloropropane	5	10 U	10 U	1	10 U	10 U	10 U	+	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 UJ	10 U
2-Butanone		10 U	10 U		10 U	10 U	10 U		10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 UJ	10 U
4-Methyl-2-pentanone		7.8 J	3 J	<del></del>	10 U	1,800	40 J	· · · · · · · · · · · · · · · · · · ·	200 J	200 J	10 U	10 U	10 U	10 U	10 U	10 UJ	10 U
Acetone		10 U	10 U		10 U	41	10 U		190	160	10 U	5 J	10 U	10 U	10 U	10 UJ	10 U
Bromodichloromethane	80	10 U	10 U		10 U	10 U	10 U		10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 UJ	10 U
Carbon tetrachloride	5	10 U	10 U		10 U	10 U	10 U	<del>                                     </del>	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 UJ	10 U
Chloroethane		10 U	10 U	+	10 U	10 U	10 U		10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 UJ	10 U
Chloroform	80	10 U	10 U		10 U	1 J	10 U	·	10 U	10 U	10 U	10 U	10 U	10 U	2 J	2 J	10 U
Methyl acetate		10 U	10 U		10 U	10 U	10 U		10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 UJ	10 U
Methylcyclohexane	. =	10 U	10 U	<b></b>	10 U	10 U	10 U	+	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 ບປ	10 U
Methylene chloride	5	10 U	1 B		10 U	10 U	10 U	+	10 U	10 U	10 U	1 B	10 U	10 U	10 U	1 B	10 U
Toluene .	1,000	10 U	10 U		10 U	. 10 U	10 U		10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 UJ	10 U
Trichloroethene		22	21	10 U	10 U	. 29	883		10 U	10 U	10 U	30	10 U	20	1,500	1,400	140
Vinyl chloride	2	10 U	49	1 J	10 U	74	100 J		170 J	180 J 326	10 U	6 J	10 U	10 U	10 U	10 UJ	10 U
cis-1,2-Dichloroethene	70	15	25	5 J	2 J	3,500	1,200	1,000	310		10 U	26	10 U	29	540	560	360
Irans-1,2-Dichloroethene	100	10 U	10 U	10 U	10 U	7 J	10 U	3 J	10 U	10 U	10 U	10 U	10 U	10 U	3 J	3 J	10 U
								ļ				ļ					
Total Metals (UG/L)		·						ļ									
Calcium		NA NA	NA NA	, NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA_	NA NA	NA NA	NA NA	, NA	NA NA	. NA	NA NA
Cobalt		NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
Cyanide	200	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA_	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
Iron	·	NA NA	NA NA	NA NA	7,620 J	15,900 J	NA	48,800 J	NA	NA NA	NA NA	NA NA	290 J	511 J	454 J	NA NA	NA NA
Magnesium		NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA_	NA_	NA_	NA NA	NA NA	NA	NA NA	NA	NA NA
Manganese	-	NA.	NA NA	NA NA	731 J	3,860	NA NA	7,180	NA	NA NA	NA NA	NA NA	640	52.7	972	NA NA	NA NA
Nickel		NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	, NA	NA NA	NA NA	NA NA	NA NA
Potassium		NA NA	NA NA	NA NA	NA NA	NA NA	NA_	NA I	NA NA	NA NA	NA NA	NA .	NA NA	NA	NA NA	NA NA	NA NA
Silver		NA NA	NA NA	NA NA	NA NA	NA NA	NA.	NA NA	NA NA	NA NA	NA NA	NA_	NA NA	NA NA	NA NA	NA I	NA NA
Sodium		NA NA	NA NA	NA_	NA NA	NA NA	NA	NA NA	NA_	NA NA	NA NA	NA NA	NA NA	NA .	NA NA	NA I	NA NA
					ļ												
Dissolved Metals (UG/L)				<u> </u>				<b> </b>		<b>!</b>							
Calcium		NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
Cobalt		NA NA	NA NA	NA_	NA NA	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA NA	NA .
Iron		NA	NA NA	NA NA	7,740 J	16,700	NA NA	47,100	NA	NA NA	NA NA	12.4 B	33.7 B	11.7 B	191	NA NA	NA NA
Magnesium	<u> </u>	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA.	NA.	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA
Manganese	<u> </u>	93.6	NA NA	NA NA	810 J	3,930	3,590	7,260	6,900	6,830	NA NA	0.91 U	65.6	27	916	, NA	2,300
Nickel	I	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA
Potassium	<del>-</del> -	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NANA	NA NA	NA NA
Sodium		NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
	I							ļ				ļ					
Wet Chemistry (MG/L)				<u> </u>						<b></b>		ļ					
Alkalinity	-	14	NA NA	NA	NA_	NA .	180	NA NA	94	96	NA NA	NA NA	. NA	NA NA	NA NA	NA NA	70
Chloride		26	NA NA	NA NA	NA NA	NA NA	55 L	NA NA	44 L	38 L	NA.	NA NA	NA NA	NA NA	NA NA	NA NA	31 L
Ethane		0.0062 U	NA NA	NA	0.01 U	0.01 U	0.0062 U	0.01 U	0.0062 U	0.0062 U	NA NA	NA NA	0.01 U	0.01 U	0.01 U	NA NA	0.0062 U
Ethene		0.0058 U	NA	NA NA	0.01 U	0.074	0.066	0.18	0.061	0.072	NA.	NA NA	0.01 U	0.01 U	0.01 U	NA	0.0058 U
Methane	-	0.0063 J	, NA	NA NA	0.19	0.49	0.32 J	0.56	0.15 J	0.17 J	NA NA	NA NA	0.081	0.069	0.037	NA NA	0.013 J
Nitrate	10	0.1 ปป	NA NA	NA	0.05 U	0.05 U	0.14	0.42	0.16	0.15	NA NA	NA NA	0.58	0.3	0.53	NA	0.1 U
Nitrite	1	NA	NA	NA.	0.05 U	0.0056 J	NA	0.022 J	NA NA	NA NA	NA	NA	0.05 U	0.05 U	0.05 U	NA	NA
Sulfate	_	11	NA NA	NA	19	21	20	1.7	16	14	NA.	NA	17	19	25	NA NA	30
Sulfide	_	NA	NA NA	NA.	1 U		NA	1 U	NA NA	NA.	NA.	NA	0.25 J	1 U	1 U	NA	NA
Total organic carbon (TOC)		4	NA NA	NA NA	13	570	520	260	220	230	NA NA	NA NA	0.71 J	4.5	210	NA	180
				<del>-</del>	•							***					

U- Analyte not detected
J- Reported value is estimated

UJ- Analyte not detected. Quantitation limit may be imprecise

L- Reported value is estimated

B- Possible blank contamination

NA- Not analyzed

Shading represents exceedance of MCL Screening Criteria

No criteria established

"P" Identifier on sample ID indicates a duplicate sample
VOCs- Volatile Organic Compounds

### Table 4 Detects in Groundwater (March and October 2005) Pre-Feasibility Study Investigations NAB Little Creek Virginia Beach, Virginia

									Virginia Beach, Vi								
Station ID	MCL-	ŁS11-M	MW10D	LS11-	MW11D	LS11-MW13D	LS11-MW14D	LS11-MW17D	LS11-MW18Y	LS11	-MW19Y	LS11-MW20Y	LS11-l	MW23D	LS11-MW24D	LS11-	MW25D
Sample ID	Groundwater	LS11-MW10D-05A	LS11-MW10D-05D	LS11-MW11D-05A	LS11-MW11DP-05A	LS11-MW13D-05A	LS11-MW14D-05A	LS11-MW17D-05A	LS11-MW18Y-05A	LS11-MW19Y-05A	LS11-MW19YP-05A	LS11-MW20Y-05A	LS11-MW23D-05A	LS11-MW23D-05D	LS11-MW24D-05A	LS11-MW25D-05A	LS11-MW25D-05D
Sample Date	Gloundwater	03/29/05	10/10/05	03/29/05	03/29/05	03/29/05	03/29/05	03/29/05	03/29/05	03/29/05	03/29/05	03/30/05	03/31/05	10/10/05	03/31/05	03/31/05	10/11/05
Chemical Name			1				f	1			1	1			1	1	1
							-				† · · · · · · · · · · · · · · · · · · ·		· ·		<u> </u>		
VOCs (UG/L)					•						<del>                                     </del>	·	· · · · · · · · · · · · · · · · · · ·		<del> </del>		<del> </del>
1,1,1-Trichloroethane	200	10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	84	7.4 J	50 U	32,000	27,000 L
1,1,2-Trichloroethane	5	10 U	10 U	10 U	NA NA	10 U	<del> </del>	10 U	10 U	10 U	· · · · · · · · · · · · · · · · · · ·	<del></del>	1 J	10 U	50 U	2,000 U	14"
1,1-Dichloroethane		3 J	3.9 J	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	<del></del>	<del></del>	760	1,300	10 J	12,000	
1,1-Dichloroethene	7	34	- 58	10 U	NA NA	10 U	10 U	10 U	10 U		<del> </del>	10 U	780	1,300	8.3	2,700	12,000 L
1,2-Dichloroethane	5	10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U		<del> </del>		2 J	2.7 J			3,900 L
1,2-Dichloroethene (total)	70	10 J	NA NA	10 U	NA NA	10 U	10 U	10 U	10 U	10 U			1,000	<del></del>	50 U 160	2,000 U	10 U
1,2-Dichloropropane	, ,	10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	· · · · · · · · · · · · · · · · · · ·			NA AG LI		300,000	NA NA
2-Butanone		10 U	10 U	10 U	NA NA	10 U	10 U	10 U		<del>}</del>	10 U	10 U	10 U	10 U	50 U	2,000 U	22
4-Methyl-2-pentanone		10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U		10 U	7 J	10 U	26 J	2,000 U	10 U
Acetone		5 J	10 U	10 U	NA NA		10 U		10 U		<del></del>	10 U	110	160 J	1,900	1,500 J	2,200 J
	90	10 U	10 U			10 U	<del>+</del>	10 U	10 U	10 U	10 U	10 U	140	10 U	51	1,500 J	1,100 J
Bromodichloromethane	- 00			10 U	NA NA	10 U	10 U	10 U	10 U		10 U	10 U	10 U	10 U	50 U	2,000 ∪	10 U
Carbon tetrachloride		10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U		10 U	10 U	10 U	50 U	2,000 U	1,309 J
Chloroethane Chloroform		10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	970	2,000 U	10 U
Chloroform	80	1 J	1.4 J	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	50 U	2,000 U	10 U
Methyl acetate		10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	50 U	2,000 U	10 U
Methylcyclohexane		10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	50 U	2,000 U	2 J
Methylene chloride	5	10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	2 B	10 U	12 B	2,800	7,100 L
Toluene	1,000	10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	1.3 J	50 U	2,000 U	16
Trichloroethene	5	280	370	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	7.1	13	50 U	2,000 U	1,200 J
Vinyl chloride	2	10 U	10 U	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	3,200	5,500	3,000	1,600 J	7,600 L
cis-1,2-Dichloroethene	70	10 J	12	10 U	NA NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	1,000	1,100	160	300,000	260,000 L
trans-1,2-Dichloroethene	100	10 U	10 U	10 U	NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	13	11	50 U	550 J	300 J
											<u> </u>						
Total Metals (UG/L)											ļ						
Calcium		NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA
Cobalt		NA NA	NA NA	NA NA	NA NA	NA_	NA NA	NA NA	NA NA	NA.	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
Cyanide	200	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA	NA NA	NA NA	NA .	NA NA	NA NA	NA NA	NA NA	NA NA
iron		187 J	ŅA	21.3 B	33.7 B	443 J	NA .	NA NA	NA NA	NA NA	NA NA	NA NA	6,090 J	NA NA	NA NA	71,900 J	NA NA
Magnesium	~	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA.	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA_	NA NA
Manganese	_	215	NA NA	90.1	. 99	8.6	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	51.4 J	NA NA	NA NA	8,880 J	NA NA
Nickel		NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA_
Potassium		NA .	NA .	NA NA	NA NA	NA NA	NA NA	NA NA	NA.	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA	NA NA
Silver		NA :	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA
Sodium		NA .	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA
Dissolved Metals (UG/L)																	
Calcium		NA .	NA NA	NA NA	NA NA	NA.	NA NA	NA NA	. NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA
Cobalt		NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA
iron		8 B	NA NA	7.73 U	7.73 U	11.5 B	NA NA	NA NA	NA NA	NA	NA NA	NA NA	472 J	NA NA	NA	67,000 J	NA NA
Magnesium		NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA.	NA NA	NA NA	NA	NA	NA	NA.
Manganese		51.2	128	61.1	62.5	1.4 B	NA NA	NA NA	NA NA	NA	NA	NA NA	35.3 J	142	NA	8,610 J	9,150
Nickel		NA NA	NA NA	NA NA	NA .	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA .	NA.
Potassium		NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA.	NA	NA	NA	NA	NA
Sodium		NA NA	NA NA	NA	NA NA	NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA NA	NA	NA	NA	NA
Wet Chemistry (MG/L)																	
Alkalinity		NA	8.3	NA	NA NA	NA	NA.	NA NA	NA NA	NA.	NA NA	NA NA	NA	87	NA NA	NA	470
Chloride		NA	49 L	NA	NA NA	NA	NA	NA	NA	NA NA	NA	NA NA	NA	59 L	NA	NA	400
Ethane	-	0.01 U	0.0062 U	0.01 U	NA NA	0.01 U	NA NA	NA .	NA	NA	NA	NA NA	0.01 U	0.0062 U	NA	0.01	0.0062 U
Ethene		0.01 U	0.0058 U	0.01 U	NA	0.01 U	NA NA	NA	NA	NA	NA NA	NA NA	0.42	0.46	NA NA	0.043	0.066 J
Methane		0.01 U	0.011 J	0.01 U	NA	0.01 U	NA NA	NA	NA.	NA	NA NA	NA NA	0.74	0.86 J	NA NA	0.32	0.37 J
Nitrate	10	1	0.18	1.8	NA	1.5	NA NA	NA.	NA	NA	NA NA	NA NA	0.05 ປ	0.15	NA NA	0.05 U	0.14 L
Nitrite	1	0.05 U	NA.	0.05 U	NA	0.054	NA	NA NA	NA.	NA	NA NA	NA NA	0.014 J	NA.	NA NA	0.14	NA NA
Sulfate		41	44	21	NA NA	13	NA .	NA	NA.	NA	NA NA	NA NA	5 U	0.1 U	NA NA	1 U	0.12
Sulfide		1.2	NA	1.2	NA NA	1	NA NA	NA NA	NA.	NA NA	NA NA	NA NA	1 U	NA NA	NA NA	0.7 J	NA NA
Total organic carbon (TOC)		0.45 J	1 U	0.49 J	NA	2.4	NA .	NA NA	NA	NA NA	NA NA	NA NA	120	230	NA NA	2,600	3,600
Notes:									· · · · · · · · · · · · · · · · · · ·	<del></del>	· · · · · · · · · · · · · · · · · ·	·				2,000	0,000

Notes: U- Analyte not detected

J- Reported value is estimated

UJ- Analyte not detected. Quantitation limit may

L- Reported value is estimated B- Possible blank contamination NA- Not analyzed

Shading represents exceedance of MCL Screer

No criteria established

"P" Identifier on sample ID indicates a duplicate
VOCs- Volatile Organic Compounds

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Table 4
Detects in Groundwater (March and October 2005)
Pre-Feasibility Study Investigations
NAB Little Creek
Virginia Beach, Virginia

									virginia Beach, V	
Station ID		LS11-	MW26D	LS11-MW27D	LS11-MW28D	LS11-MW29D	LS11-MW30D	LS11-	MW36D	LS11-MW37D
Sample ID	MCL-	LS11-MW26D-05A	LS11-MW26D-05D	LS11-MW27D-05A	LS11-MW28D-05A	LS11-MW29D-05A	LS11-MW30D-05A	LS11-MW36D-05A	LS11-MW36DP-05A	LS11-MW37D-05A
Sample Date	Groundwater	03/31/05	10/11/05	03/31/05	03/30/05	03/31/05	03/31/05	04/01/05	04/01/05	04/01/05
<del></del>		05/01/00	10/11/00	03/3/1/03	05/50/05	1 03/31/03	1 03/3//03	04/01/03	04/01/03	04/01/03
Chemical Name										
				· · · · · · · · · · · · · · · · · · ·						
VOCs (UG/L)					Name :					ļ
1,1,1-Trichloroethane	200	17	190 J	10 UJ	12,000	1,400	2,100	10 U	10 U	10 U
1,1,2-Trichloroethane		10 U	10 U	10 UJ	4 Y5	100 U	200 U	10 U	10 U	10 U
1,1-Dichloroethane		170	940 L	520	4,000	920	2,600	10 U	10 U	4 J
1,1-Dichloroethene	7	1.3	10.1	10 UJ	2,700	87 J	390	10 U	10 U	.70
1,2-Dichloroethane	5	10 U	1.9 J	10 UJ	23	100 U	200 U	10 ป	10 U	10 U
1,2-Dichloroethene (total)	70	19	NA	25 J	60,000 +	3,600	47,000	10 U	10 U	3 J
1,2-Dichloropropane	5	10 U	10 U	10 UJ	e j	100 U	200 U	10 U	10 U	10 U
2-Butanone		4 J	45 J	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
4-Methyl-2-pentanone		640 J	1,600 L	37 J	3,000	1,000	610	10 U	10 U	10 U
Acetone		51	820 L	10 UJ	370	240	340	10 U	10 U	10 U
Bromodichloromethane	80	10 U	10 U	10 UJ	10 U	100 U	200 U	2 J	2 J	10 U
Carbon tetrachloride		10 U	10 U	10 UJ	10 U	100 U		10 U		<b>4</b>
					<del></del>		200 U	<del></del>	10 U	10 U
Chloroethane		10 U	10 U	32 J	10 U	100 U	200 U	10 U	10 U	10 U
Chloroform	80	10 U	10 U	10 UJ	13	100 U	200 U	6 J	6 J	10 U
Methyl acetate		10 U	16 J	10 UJ	10 U	100 U	200 ป	10 U	10 U	10 U
Methylcyclohexane		10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Methylene chloride	. 5	10 U	10 U	2 B	470 %	23 B	140 J	10 U	10 U	2 B
Toluene	1,000	10 U	1.6 J	10 UJ	5 J	100 U	200 U	4 J	2 J	5 J
Trichloroethene	5	1 J	67 J	2 J	24	100 U	53 J	3 J	2 J	230
Vinyl chloride	2	280 J	3,500 L	140 J	2,300	4.400	5,400	10 U	10 U	10 U
cis-1,2-Dichloroethene	70	19	330 L	25 J	60,000	3,000	47,000	10 ປ	10 U	3 J
trans-1,2-Dichloroethene	100	10 U	5.9 J	10 UJ	450	100 U	120 Ú	10 U	10 U	10 U
						1				
Total Metals (UG/L)					1					
Calcium		NA ·	NA.	NA	NA NA	NA.	NA	NA	NA	NA
Cobalt		NA	NA	NA	NA	NA	NA	NA	NA NA	NA NA
Cyanide	200	NA	NA	NA	NA NA	NA NA	NA.	NA	NA NA	NA.
Iron	-	10,500 J	NA	NA	NA	NA	NA.	174 J	142 J	630 J
Magnesium		NA	NA.	NA	NA NA	NA NA	NA	NA	NA.	NA NA
Manganese		871 J	NA	NA	NA NA	NA NA	NA NA	61.4 J	58.2 J	160 J
Nickel		NA NA	NA	NA	NA NA	NA.	NA	NA	NA.	NA NA
Potassium		NA	NA NA	NA	NA NA	NA NA	NA.	NA	NA	NA NA
Silver	_	NA	NA	NA	NA	NA NA	NA NA	NA	NA	NA NA
Sodium		NA	NA NA	NA	NA NA	NA NA	NA.	NA	NA.	NA
Dissolved Metals (UG/L)							· · · · · · · · · · · · · · · · · · ·			
Calcium		NA	NA	NA.	NA NA	NA NA	NA NA	NA.	NA	NA
Cobalt		NA NA	NA.	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
Iron		9,120 J	NA NA	NA NA	NA NA	NA NA	NA NA	19.2 B	17.7 B	39.9 B
Magnesium		NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA
Manganese		1,110 J	3,340	NA NA	NA NA	NA NA	NA NA	52.9 J	53.1 J	156 J
Nickel		1,110 J	9,340 NA	NA NA	NA NA	NA NA	NA NA	32.9 J	33.1 J NA	NA NA
Potassium		NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
Sodium		NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	
Socium		INA	INA	INA .	NA.	N/A	<u> </u>	NA.	INA	NA NA
Wet Chemistry (MG/L)		· · · · · · · · · · · · · · · · · · ·								
			252				<del> </del>			
Alkalinity		NA NA	250	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
Chloride		NA	66	NA	NA NA	NA NA	NA	NA	NA NA	NA NA
Ethane		0.01 U	0.0062 U	NA	NA NA	NA	NA NA	0.01 U	NA NA	0.01 U
Ethene		0.15	0.44 J	NA NA	NA NA	NA NA	NA NA	0.01 U	, NA	0.01 U
Methane		4.9	5.8 J	NA NA	NA NA	NA NA	NA NA	0.01 U	NA NA	0.076
Nitrate	10	0.05 U	0.12 L	NA_	NA NA	NA NA	NA NA	0.046 J	NA NA	0.028 J
Nitrite	1	U 800.0	NA NA	NA	NA NA	NA NA	NA	0.05 U	NA	0.05 U
Sulfate	-	1 U	0.24	NA NA	NA NA	NA	NA NA	11	NA	12
Sulfide		1 U	NA	NA	NA	NA NA	NA NA	1 U	NA	1 U
Total organic carbon (TOC)		290	720	NA	NA NA	NA NA	. NA	1.3	NA NA	0.55 J
Notes:										

U- Analyte not detected

J- Reported value is estimated

UJ- Analyte not detected. Quantitation limit may

L- Reported value is estimated B- Possible blank contamination NA- Not analyzed

Shading represents exceedance of MCL Screen

-- No criteria established "P" Identifier on sample ID indicates a duplicate

VOCs- Volatile Organic Compounds

Table 4 Delects in Groundwater

Table 5 Detections in Soil (October 2005)
Pre-Feasibility Study Investigations
NAB Little Creek
Virginia Beach, Virginia

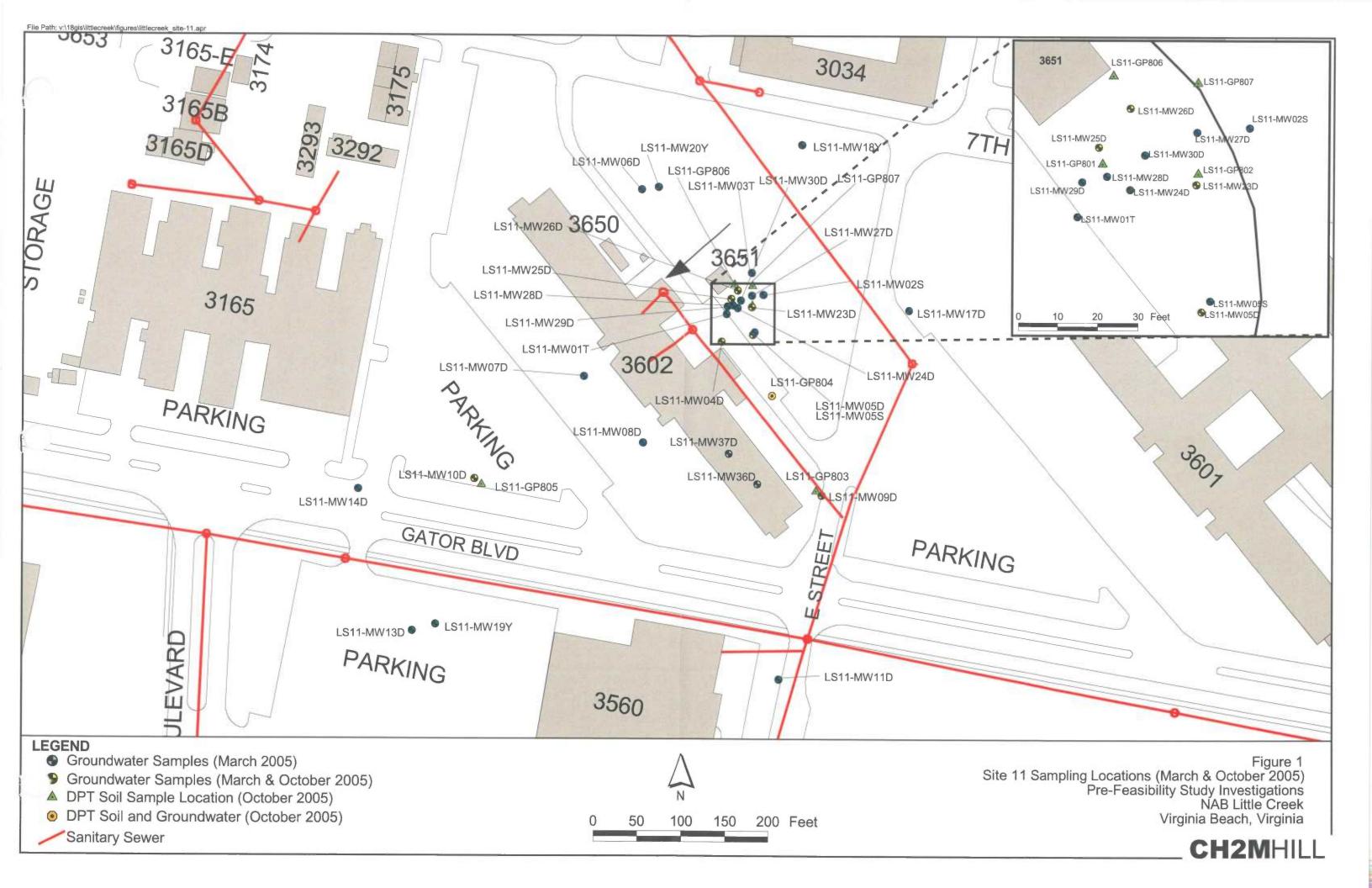
Station ID		LS11-GP801		·····	LS11-GP802		LS11-GP803	LS11-GP804		
Sample ID	LS11-SB801-UC	LS11-SB801-LC	LS11-SB801-YC	LS11-SB802-YC	LS11-SB802-LC	LS11-SB802-UC	LS11-SB803-LC	LS11-SB804-LC	LS11-SB804P-LC	
Sample Date	10/07/05	10/07/05	10/07/05	10/08/05	10/08/05	10/08/05	10/08/05	10/07/05	10/07/05	
Chemical Name										
Volatile Organic Compounds (UG/KG)										
1,1,1-Trichloroethane	12 U	93	16 U	15 J	16 J	12 U	12 U	12 U	13 ∪	
1,1-Dichloroethane	2.5 J	38	2,100	3,000 J	7.3 J	7.1 J	4.9 J	12 U	13 U	
1,1-Dichloroethene	12 U	5 J	210	620 J	11 U	12 U	12 U	12 U	13 U	
1,2-Dichloroethane	12 U	12 U	16 U	6.8 J	11 U	12 U	12 U	12 U	13 U	
2-Butanone	12 U	13	16 U	17 U	11 U	12 U	12 U	12 U	13 U	
4-Methyl-2-pentanone	12 U	200	16 U	72 J	14	8.5 J	12 U	12 U	13 U	
Acetone	12 U	510 J	16 J	210 J	11 U	12 U	12 U	12 U	13 U	
Carbon disulfide	12 U	12 U	21	40	11 U	12 U	12 U	12 U	13 U	
Chloroethane	12 U	12 U	11 J	34	11 U	12 U	12 U	12 U	13 U	
Ethylbenzene	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U	13 U	
Methylene chloride	12 U	22 J	530 J	240	11 U	12 U	12 U	12 U	13 U	
Toluene	12 U	12 U	16 U	3.7 J	11 U	12 U	12 U	12 U	13 U	
Trichloroethene	6 J	22 J	18,000	25,000 J	17 J	18	23	3 J	25	
Vinyl chloride	12 U	7.3 J	16 U	140	11 U	12 U	12 U	12 U	13 U	
Xylene, total	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U	13 U	
cis-1,2-Dichloroethene	11 J	600	130 J	8,100 J	130 J	26	16	8.2 J	9.9 J	
trans-1,2-Dichloroethene	12 U	12 U	16 U	42	11 U	12 U	12 U	12 U	13 U	
Wet Chemistry (MG/KG)										
% Solids	86	86	63	60	90	83	85	82	77	
Total organic carbon (TOC)	NA	950	12,000	22,000	1,300	NA	NA	_ NA	NA	

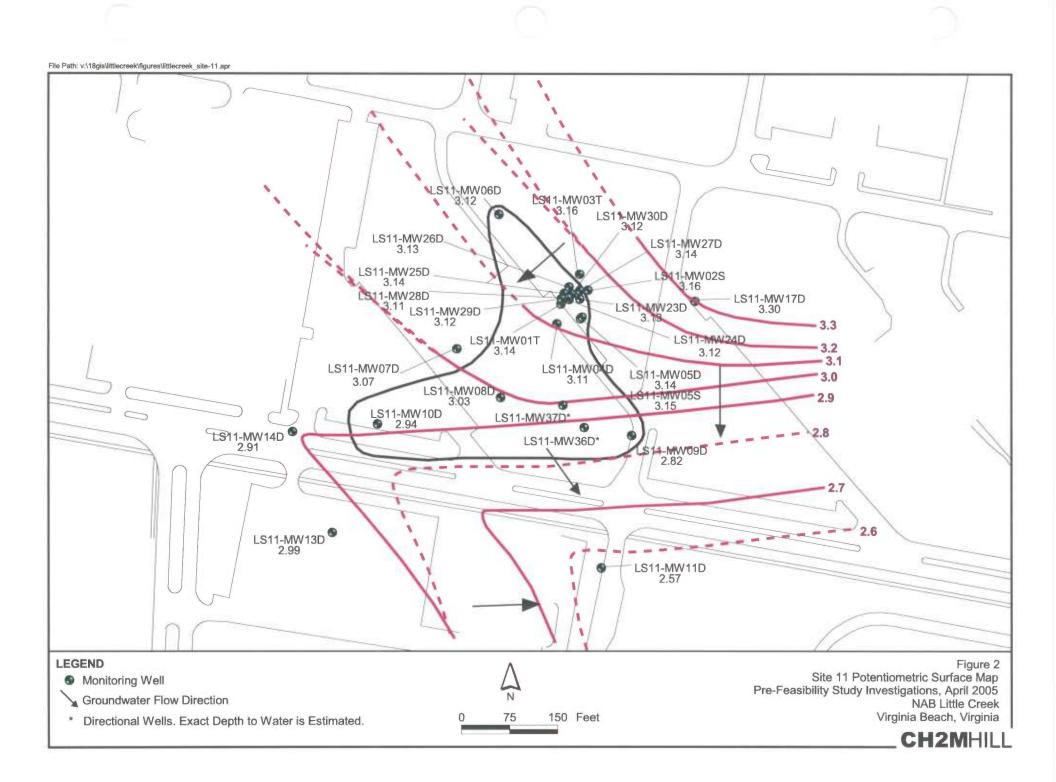
NA - Not analyzed J - Reported value is estimated U - Analyte not detected

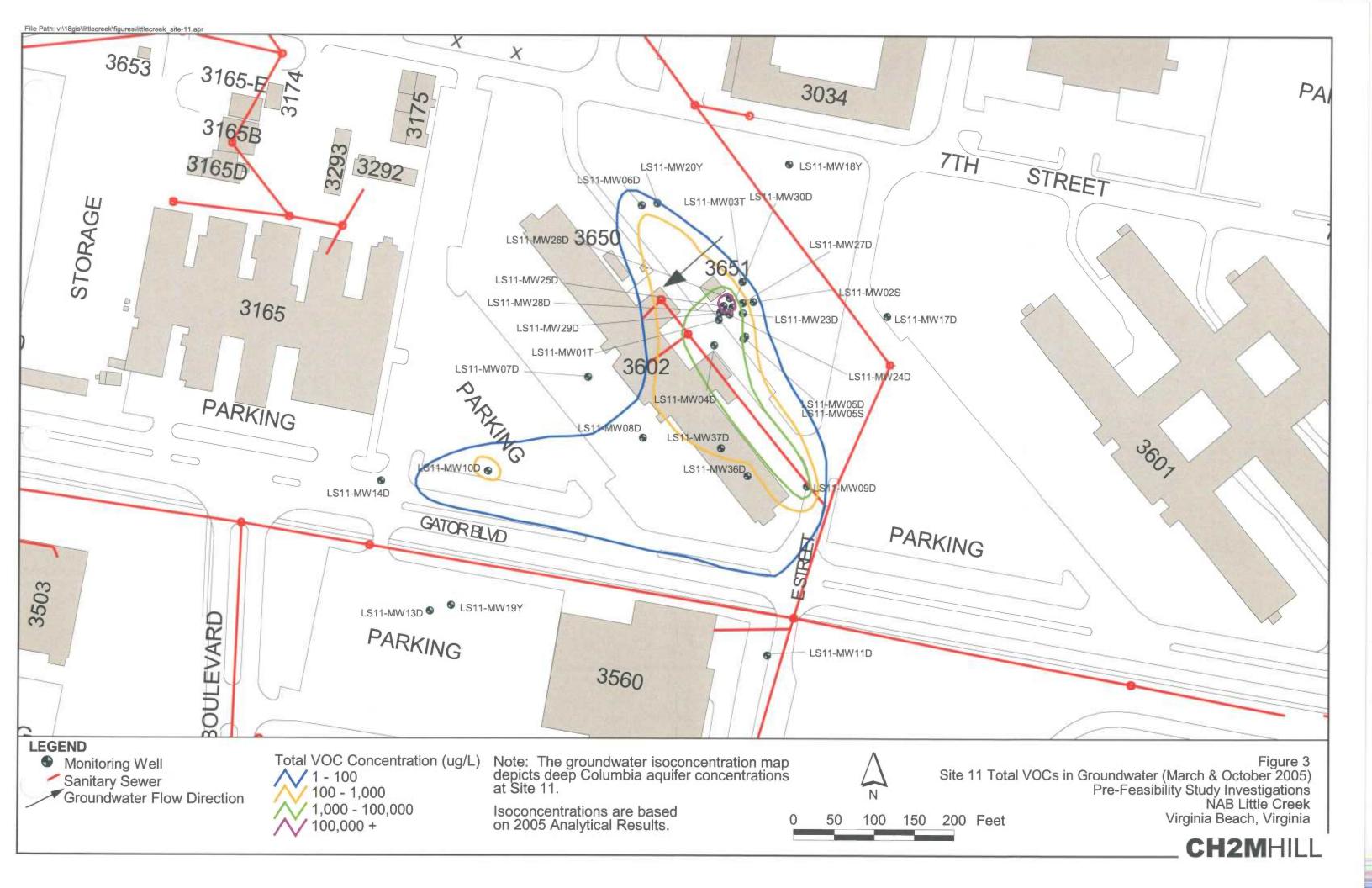
## Table 6 Total Oxidant Demand (TOD) (October 2005) Pre-Feasibility Study Investigations NAB Little Creek Virginia Beach, Virginia

Well Number	Oxidant (g/kg)
LS11-SB801-LC	3.7 Na persulfate
LS11-SB801-YC	> 19.5 Na persulfate
LS11-SB805-LC	1.9 Na persulfate
LS11-SB805-YC	11 Na persulfate
LS11-SB806-LC	2.5 Na persulfate
LS11-SB806-YC	12 Na persulfate

**Figures** 







**Attachment A** 



Project Number 157234.FS.DR Well Number LS11-MW36D

Sheet 1 of 1

### **DIRECTIONAL WELL COMPLETION DIAGRAM**

PROJECT: NAB Little C	reek		LOCATI	ON: Site 11	
DRILLING CONTRACTO		NORTHING:		EASTING:12	2169560.79
	ID EQUIPMENT USED: 41/4				
WATER LEVELS:	START: 3/28/2		): 3/29/2005 1250	LOGGER: C. V	Vhite
1 - Ground elevation at v 2 - Top of casing elevation 3 - Wellhead protection of a - drain tub b - concrete 4 - Dia/type of well casin	well: 9.62 on: 9.17 cover: Flushmount pe? e pad dimensions: 2' diame			17.8' 19.3'	VIIICE
6 - Type screen filter:	#1 silica pack	, , , , , , , , , , , , , , , , , , ,			
a - quantity	used:		5		
7 - Type of seal: a) quantity ( 8 - Grout:	Bentonite 3/8* chips used:		6—		
a) grout mix	used: Bentonite/g	rout mix		<b>\</b>	
	of placement: Tremmie			28.2'	
'	ell casing grout:	<del></del>	,	~48	3° ( )
Development method:	Whale Pump				
Development time:	53 minutes				
Estimated purge vol:	65 gallons				
Comments:	Well set at an angle of ~48	degrees			



Project Number

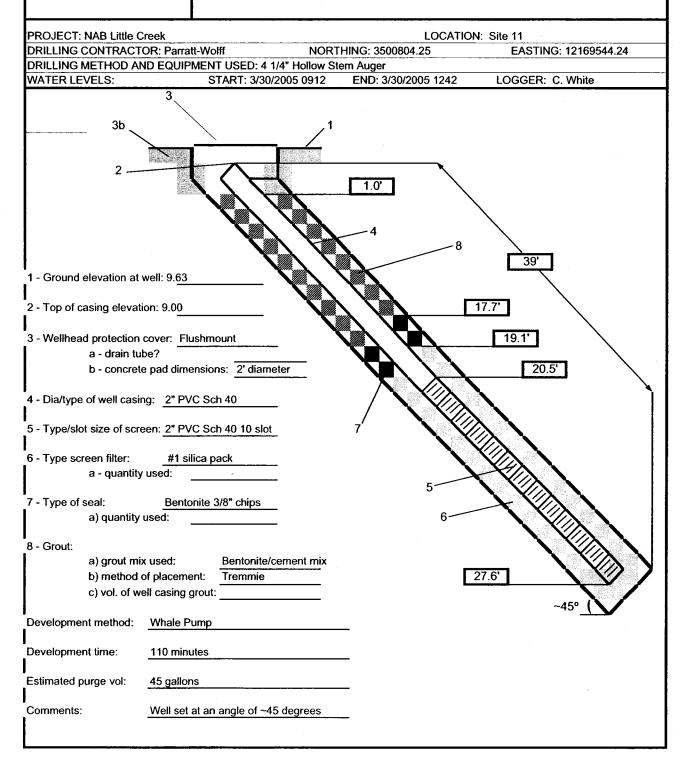
157234.FS.DR

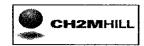
Well Number

LS11-MW37D

Sheet 1 of 1

### **DIRECTIONAL WELL COMPLETION DIAGRAM**





PROJECT NUMBER 157234.FS.FR BORING NUMBER LS11-MW36D

SHEET 1

OF 2

### **SOIL BORING LOG**

PROJECT : CTO-021 DRILLING CONTRACTOR : Parratt-Wolff LOCATION : NAB Little Creek Site 11
ELEVATION : 9.17 NORTHING: 3500880.96 EASTING: 12169560.79

DRILLING METHOD AND EQUIPMENT USED :4 1/4" Hollow Stem Auger - Directional Drilling at 45 degree angle

WATER	R LEVELS of Auger D	3:		START:	44" Hollow Stem Auger - Directional Drilling at 45 ( 3/28/05 1225 END: 3/28/05 1600 CORE DESCRIPTION	LOGGER : J. Butter/C. White  COMMENTS
	INTERVA		- ( - 7	STANDARD	COME DESCRIPTION	CONTINUE 1413
	1	RECOVE	RY (IN) #/TYPE	PENETRATION TEST RESULTS 6"-6"-6" 6"	SOIL NAME, USCS GROUP SYMBOL, COLOR, MOISTURE CONTENT, RELATIVE DENSITY, OR CONSISTENCY, SOIL STRUCTURE, MINERALOGY.	DEPTH OF CASING, DRILLING RATE, DRILLING FLUID LOSS, TESTS, AND INSTRUMENTATION. OVM (ppm): Breathing Zone Above Hole
5	10-14'	42 <sup>-</sup>	1		10.0'-13.0' Silty SAND, SM, light yellowish brown (2.5Y 6/4), saturated, loose	PID = 0.0ppm
- - 15	14-18'	48"	2		13.0'-14.0' Silty coarse SAND, SW, olive yellow (2.5Y 6/6), saturated, loose 14.0'-18.0' Coarse SAND, SW, olive yellow (2.5Y 6/6) some silt, saturated, loose	_
_ _ _ 20	18-22'	48*	3		 18.0'-22.0' Coarse silty SAND, SW, olive yellow (2.5Y 6/6), saturated, loose 	
- - - 25	22-26'	48"	4		22.0'-23.0' Silty SAND, SM, olive yellow (2.5Y 6/6), saturated, loose 23.0'-26.0' Gravelly SAND, SW, olive yellow (2.5Y 6/6), saturated, loose, dark brown (2.5Y 4/3) lense 25.0' 25.5'	-
  	26-30'	48"	5		26.0'-27.0' Coarse SAND, SW, olive yellow (2.5Y 6/6) saturated, loose 27.0'-29.0' Silty SAND, SM, light brownish grey (2.5Y 6/2), saturated, loose	PID = 0.0ppm -



PROJECT NUMBER 157234.FS.FR

BORING NUMBER
LS11-MW36D

SHEET 2

OF 2

### **SOIL BORING LOG**

**DRILLING CONTRACTOR: Parratt-Wolff** LOCATION: NAB Little Creek Site 11 PROJECT: CTO-021 **ELEVATION: 9.17** NORTHING: 3500880.96 EASTING: 12169560.79 DRILLING METHOD AND EQUIPMENT USED :4 1/4" Hollow Stem Auger - Directional Drilling at 45 degree angle WATER LEVELS: START: 3/28/05 1225 END: 3/28/05 1600 LOGGER: J. Butler/C. White Length of Auger Drilled @ 48° (FT) STANDARD CORE DESCRIPTION COMMENTS INTERVAL (FT) PENETRATION RECOVERY (IN) TEST SOIL NAME, USCS GROUP SYMBOL, COLOR, DEPTH OF CASING, DRILLING RATE, #/TYPE RESULTS MOISTURE CONTENT, RELATIVE DENSITY, DRILLING FLUID LOSS, 6"-6"-6"-6" OR CONSISTENCY, SOIL STRUCTURE, TESTS, AND INSTRUMENTATION. MINERALOGY. OVM (ppm): Breathing Zone Above Hole (N) 29.0'-30.0' Silty SAND, SM, light yellowish grey (2.5Y 30\_ 6/3), saturated, loose 30-34 30.0'-32.5' Silty SAND, SM, light yellowish brown (2.5Y PID = 0.0ppm 48" 6 6/4), saturated, loose 32.5'-34.0' Medium silty SAND, SM, light yellowish brown (2.5Y 6/4), moderately dense 34-38' PID = 0.0ppm 48" 7 35 34.0'-35.5' Medium SAND, SM, light yellowish brown (2.5Y 6/4), saturated, loose, 1" clay layer at 34.5' 35.5' 37.5' Coarse SAND, SM, light yellowish brown (2.5Y 6/4), saturated, loose 37.5'-38.0' Medium SAND, SC, light yellowish brown (2.5Y 6/4), saturated, medium dense 48" 38-42' 8 38.0'-40.0' Medium SAND, SC, light yellowish brown PID = 0.0ppmMonitoring well LS11-MW36D installed (2.5Y 6/4), saturated, medium dense, mottling 40\_ 40.0'-41.0' Clayey SILT, CL, strong brown (7.5YR 5/6), saturated 41.0-42.0 Silty CLAY, CL, very dark grey (GLEY 1 3/N), saturated, medium dense End of boring @ 42.0' of rods Note: Boring completed at ~48° angle, thus the boring dept is indicative of the length of auger (not vertical distance bgs). 45\_ 50



ROJECT NUMBER	BORING NUMBER
57234.FS.FR	LS11-MW37D

SHEET 1 OF 2

### **SOIL BORING LOG**

PROJECT : CTO-021	DRILLING CONTRACTOR : Pa	arratt-Wolff	LOCATION : NAB Little Creek Site 11
ELEVATION:9.00	NORTHING: 3500804.25	EASTING: 12169544.24	
DRILLING METHOD AND EQUIPME	NT USED :4 1/4" Hollow Stem Auger-	Directional Drilling at 45 degre	e angle

	LEVELS		LGOII WIL		"Hollow Stem Auger- Directional Drilling at 45 de 3/29/05 1705 END: 3/30/05 0912	LOGGER : C. White
ength of	Auger @ 4	野(FT)		STANDARD	CORE DESCRIPTION	COMMENTS
	INTERVA	L (FT) RECOVE	RY (IN) #/TYPE	PENETRATION TEST RESULTS 6"-6"-6" (N)	SOIL NAME, USCS GROUP SYMBOL, COLOR, MOISTURE CONTENT, RELATIVE DENSITY, OR CONSISTENCY, SOIL STRUCTURE, MINERALOGY.	DEPTH OF CASING, DRILLING RATE, DRILLING FLUID LOSS, TESTS, AND INSTRUMENTATION. OVM (ppm): Breathing Zone Above Hole
- -					- -	
- 5					- -	
- -				: :	-	
- 10	10-14'	42*	1		10.0' 12.0' Silty SAND, SM, light yellowish brown (2.5Y 6/4), moist, loose, some coarse sand	PID = 0.0ppm
- - 15	14-18'	48"	2		12.0'-12.5' Silty SAND, SM, olive yeloow (2.5Y 6/6), moist, loose 12.5'-13.5' Silty SAND, SM, light yellowish brown (2.5Y 6/4), moist, loose, some coarse sand 13.5'-14.0' No Recovery 14.0'-18.0' Silty SAND, SM, light yellowish brown (2.5Y 6/4), saturated, loose, some coarse sand	PID = 0.0ppm
- - 20	18-22'	48"	3		18.0'-22.0' Silty SAND, SM, light yellowish brown (2.5Y 6/4), saturated, loose, some coarse sand	PID = 0.0ppm
- - -	22-26'	48"	4		22.0-23.0' Silty SAND, SM, light yellowish brown (2.5' 6/4), saturated, loose 23.0'-26.0' Silty SAND, SM, yellowish brown (10YR 5/6), saturated, loose, some coarse sand	YPID = 0.0ppm
25 ~ ~	26-30'	36"	5		26.0'-28.0' Silty SAND, SM, yellowish brown (10YR 5/6), saturated, loose, some coarse sand	PID = 0.0ppm
-					28.0'-29.0' Silty SAND, SM, yellowish brown (10YR 5/6), saturated, loose	



PROJECT NUMBER 157234.FS.FR BORING NUMBER LS11-MW37D

SHEET 2

OF 2

### **SOIL BORING LOG**

PROJECT: CTO-021 **DRILLING CONTRACTOR: Parratt-Wolff** LOCATION: NAB Little Creek Site 11 **ELEVATION:9.00** NORTHING: 3500804.25 EASTING: 12169544.24 DRILLING METHOD AND EQUIPMENT USED :4 1/4" Hollow Stem Auger- Directional Drilling at 45 degree angle WATER LEVELS START: 3/29/05 1705 END: 3/30/05 0912 LOGGER: C. White Length of Auger @ 45°(FT) STANDARD CORE DESCRIPTION COMMENTS INTERVAL (FT) PENETRATION RECOVERY (IN) TEST SOIL NAME, USCS GROUP SYMBOL, COLOR, DEPTH OF CASING, DRILLING RATE, #/TYPE RESULTS MOISTURE CONTENT, RELATIVE DENSITY, DRILLING FLUID LOSS, 6"-6"-6"-6" OR CONSISTENCY, SOIL STRUCTURE, TESTS, AND INSTRUMENTATION. MINERALOGY. (N) OVM (ppm): Breathing Zone Above Hole 29.0'-30.0' No Recovery 30\_ 30-34' 48" 6 30.0'-33.0' Clayey SAND, SC, dark grey (2.5Y 4/1), PID = 0.0ppmsaturated, medium dense, some organics 33.0'-34.0' Silty SAND, SM, grey (2.5Y 5/1), saturated, PID = 0.0ppm 34-38' 48" 7 34.0'-35.0' SAA 35 35.0'-36.0' Clayey SAND, SC, dark grey (2.5Y 4/1), saturated, medium dense 36.0'-38.0' Silty SAND, SM, light yellowish brown (2.5') 6/4), saturated, loose, some coarse sand 38-42' 48" 8 38.0'-39.5' Clayey SAND, SC, light yellowish brown PID = 0.0ppm(2.5Y 6/4), moist, medium dense Monitoring well LS11-MW37D installed 40 39.5'-42.0' CLAY, CL, greenish grey (GLEY 1 5/5GY), moist, dense, trace silt End of boring @ 42.0' of rods Note: Boring completed at ~45° angle, thus the boring length is indicative of the length of auger. 45 50\_



PROJECT NUMBER 329752.SI.WP BORING NUMBER LS11-GP801

SHEET 1

OF 1

### **SOIL BORING LOG**

		Little Cre	ek		G CONTRACTOR : Parratt-Wolff	LOCATION : Site 11
	ION :9.50		EQ111014		ING: 3500881.61 EASTING: 12169477.	18
	LEVELS		EQUIPM		7/4' Acetate Sleeve 10/7/05 END: 10/7/05	LOGGER : A. Jones/M. Ost
		RFACE (F	T)	STANDARD	CORE DESCRIPTION	COMMENTS
	INTERVA	<u>`</u>	•,	PENETRATION	GONE BEGONS HOW	
		RECOVE	RY (IN)	TEST	SOIL NAME, USCS GROUP SYMBOL, COLOR,	DEPTH OF CASING, DRILLING RATE,
	•		#/TYPE	RESULTS	MOISTURE CONTENT, RELATIVE DENSITY,	DRILLING FLUID LOSS,
	l			6"-6"-6"-6"	OR CONSISTENCY, SOIL STRUCTURE,	TESTS, AND INSTRUMENTATION.
*				(N)	MINERALOGY.	OVM (ppm): Breathing Zone Above Hole
	0'-4'	27"	1		0'-1.75' No recovery	
_			:		1.75'-2.0' FILL, medium to coarse sand, some gravel	
-	  - 				2.0'-4.0' No recovery	-
_	4'-8'	36"	2		4.0'-5.0' No recovery	
5					5.5'-6.5' Silty CLAY with organics, ML, dark grey	1
_	]				(10YR 4/1)	Sample LS11-SB801-UC collected from
			1		6.5'-7.1' Silty SAND with gravel, SM, light yellow	6.0-8.0' bgs
_				ŀ	brown (2.5Y 6/4)	<del></del>
_	l				7.1'-8.0' Medium SAND, SB, light yellow brown (2.5Y	1
	8'-12'	42"	3		6/3), moist	
-	1				8.0'-8.5' No Recovery 8.5'-10.0' Medium SAND, SB, light yellow brown	
10	]				(2.5Y 6/3), moist	
_	ļ				10.0'-12.0' Medium SAND, SB, olive yellow (2.5Y	
-	1				6/6), moist	
_	12'-16'	48"	4		12'.0-13.0' Silty fine SAND, SM, light yellowish brown	1
_	1				(2.5Y 6/3)	
	1				13.0'-14.5' Silty fine SAND, SM, olive yellow (2.5Y 6/8), saturated	1
-	ł				oroj, seimates	
15					14.5'-16.0' SAND, SP, light grey (5Y 7/1)	-[
		ļ				
_	16'-20'	24"	5	1	16.0'-18.0' No recovery	1
_			1		-	-
	1		1			i
-					- 18.0'-18.7' Coarse SAND, SP, grey (FY 6/1)	Sample LS11-SB801-LC collected from
_			1		18.7'-19.5' Coarse-med. SAND, SW, light olive	18.0'-19.5' bgs
20					brown(2.5Y 5/4), well graded, coarsening	1
20	20'-24'	48"	6		19.5'-20.0' Silty clay, ML, dark grey (2.5Y 4/1), medium dense, shell hash at 19.7'	
		~			20.0'-21.0' CLAY, CL, dark grey (2.5Y 4/1), dense,	Sample LS11-SB801-YC collected from
_	1				shell hash	20.0'-22.0' bgs
_	l		1		21.0'-24.0' CLAY, CL, dark grey (2.5Y 4/1), dense, some organics	1
	l	1	1	1	Some organica	i
_			1		-	1
	ŀ		ļ	<b></b>	End of Boring @ 24' bgs	
25				l		
_		t	1	1		1
_	1			·	-	-
			]		į	1
-	}			i	-	1
_	l				_	
-		. <u>-</u>				

PROJECT NUMBER 329752.SI.WP

**BORING NUMBER** LS11-GP802

SHEET 1

OF 1

### **SOIL BORING LOG**

PROJECT : NAB Little Creek DRILLING CONTRACTOR: Parratt-Wolff LOCATION: Sites 11 ELEVATION: 9.50 NORTHING: 3500828.91 EASTING: 12169519.40 DRILLING METHOD AND EQUIPMENT USED: DPT/4' Acetate Sleeve WATER LEVELS : 9.0' bgs START: 10/8/05 END: 10/8/05 LOGGER: A. Jones/M. Ost DEPTH BELOW SURFACE (FT) STANDARD CORE DESCRIPTION COMMENTS INTERVAL (FT) PENETRATION RECOVERY (IN) TEST SOIL NAME, USCS GROUP SYMBOL, COLOR, DEPTH OF CASING, DRILLING RATE, #/TYPE **RESULTS** MOISTURE CONTENT, RELATIVE DENSITY, DRILLING FLUID LOSS, 6"-6"-6"-6" OR CONSISTENCY, SOIL STRUCTURE, TESTS, AND INSTRUMENTATION. MINERALOGY. OVM (ppm): Breathing Zone Above Hole 0'-4' 48" 1 0.0'-0.5' Organic topsoil 0.5'-2.5' SILT, ML, very dark greyish brown (2.5Y 3/2) 2.5'-4.0' Clayey SILT, ML, light yellowish brown (2.5Y 6/4) 4'-8' 42" 2 4.0'-4.5' No Recovery 5 4.5'-5.5' Silty CLAY, ML, light yellowish brown (2.5Y 6/4) 5.5'-7.0' Clayey SILT, ML, grey (2.5Y 6/1), medium dense 7.0'-8.0' Fine SAND, SP, olive yellow (2.5Y 6/6) 8'-12' 36" 3 8.0'-9.0' No Recovery 9.0'-10.0' Coarse SAND, SP, grey (2.5Y 6/1), 10 saturated 10.0'-12.0' Fine SAND, SP, grey (2.5Y 6/1) 12'-16' 48" 4 12.0'-13.0' No recovery Sample LS11-SB802-UC collected from 13.0'-16.0' Fine SAND, SP, grey (2.5Y 6/1) 13.0'-15.0' bgs 15\_ 16'-20' 48" 5 16.0'-18.0' Fine SAND, SP, grey (2.5Y 6/1) 18.0'-20.0' Coarse SAND with gravel, SW, grey (2.5 Sample LS11-SB802-LC collected from 18.0'-20.0' bgs 6/1) 20\_ 20'-24' 48" 6 20.0'-20.2' Coarse sand with gravel, SW, gray (2.5Y Sample LS11-SB 802-YC collected from 20.5'-22.5' bgs 20.2'-24.0' N/A End of boring @ 24.0' bgs 25

PROJECT NUMBER 329752.SI.WP BORING NUMBER LS11-GP803

SHEET 1 OF 1

### **SOIL BORING LOG**

			-	<u> </u>		·
		Little Cre	ek		G CONTRACTOR : Parratt-Wolff	LOCATION : Site 11
ELEVAT			EOI IIDM		NG: 3500796.29 EASTING: 12169592.1 7 4' Acetate Sleeve	16
		: 6.0' bgs		START:		LOGGER : A. Jones/M. Ost
		RFACE (F		STANDARD	CORE DESCRIPTION	COMMENTS
	INTERVA		<del>.,</del>	PENETRATION		
		RECOVE	RY (IN)	TEST	SOIL NAME, USCS GROUP SYMBOL, COLOR,	DEPTH OF CASING, DRILLING RATE,
			#/TYPE	RESULTS	MOISTURE CONTENT, RELATIVE DENSITY,	DRILLING FLUID LOSS,
		l		6"-6"-6"-6"	OR CONSISTENCY, SOIL STRUCTURE,	TESTS, AND INSTRUMENTATION.
				(N)	MINERALOGY.	OVM (ppm): Breathing Zone Above Hole
_	0'-4'	48"	1	U	0.0'-0.5' Top soil 0.5'-2.0' SILT, ML, light olive brown (2.5Y 5/3), medium dense	·
-					2.0'-2.5' Fill material 2.5'-4.0' Sitty CLAY, ML, olive yellow (2.5Y 6/6)	-
5	4'-8'	48"	2		4.0'-5.0' Silty CLAY, ML, olive yellow (2.5Y 6/6), mois 5.0'-8.0' Medium SAND, SP, light grey (2.5Y 6/8), saturated, iron staining	
- -	8'-12'	48"	3			-
- 10	0-12	40			saturated (2.31 of 5)	- -
- - -	12'-16'	36"	4	:	12.0'-13.0' No recovery  13.0'-16.0' Medium SAND, SP, olive yellow (2.5Y 6/8), saturated	- -
15 - -	16'-20'	18"	5			_ - -
- 20_					18.5'-19' Coarse SAND with gravel, SW, yellowish brown (10YR 5/8)  19'-20' Silty sand, SM 20' Clay, CL, greenish gray (Gley1 5/5GY 5/1)	Sample LS11-SB803 collected from 18.0'- 20.0' bgs
-					End of boring @ 20.0' bgs	-
-					<u>-</u>	-
_					-	-
25						_
					-	-
-					-	-
_					_	- -

PROJECT NUMBER 329752.SI.WP BORING NUMBER LS11-GP804

SHEET 1

OF 1

### **SOIL BORING LOG**

PROJECT : NAB Little Creek DRILLING CONTRACTOR: Parratt-Wolff LOCATION: Site 11 ELEVATION: 9.50 NORTHING: 3500857.18 EASTING: 12169581.12 DRILLING METHOD AND EQUIPMENT USED: DPT/4' Acetate Sleeve WATER LEVELS: 7.5' bgs START: 10/7/05 END: 10/7/05 LOGGER: A. Jones/M. Ost DEPTH BELOW SURFACE (FT) STANDARD CORE DESCRIPTION COMMENTS INTERVAL (FT) PENETRATION RECOVERY (IN) TEST SOIL NAME, USCS GROUP SYMBOL, COLOR, DEPTH OF CASING, DRILLING RATE, #/TYPE RESULTS MOISTURE CONTENT, RELATIVE DENSITY, DRILLING FLUID LOSS, 6"-6"-6"-6" OR CONSISTENCY, SOIL STRUCTURE. TESTS, AND INSTRUMENTATION. (N) MINERALOGY. OVM (ppm): Breathing Zone Above Hole 0.0'-0.5' Asphalt 0.5'-2.7' SILT with clay lens, ML, brown (10YR 4/3), 0'-4' 36" 1 dry, medium dense 2.7'-3.2' Fine SAND, SP, light olive brown (2.5Y 5/3), dry, loose 3.2'-4.0' Clayey SILT, ML, light olive brown (2.5Y 5/6) 4'-8' 48" 2 dense 5\_ 4.0'-5.6' Silty CLAY, ML, light olive brown (2.5Y 5/6), medium dense, sand lense at 4.2'-4.4' 5.6'-8.0' Medium SAND, SP, light yellowish brown (2.5Y 6/4), saturated, loose, iron staining 8'-12' 3 36" 8.0'-9.0' No recovery 9.0'-11.0' Medium SAND, SP, light yellowish brown 10\_ (2.5Y 6/4) 11.0'-12.0' Fine SAND, SP, light yellowish brown (2.5Y 6/4), saturated 12'-16' 36" 4 12.0'-13.0' No recovery 13.0'-15.5' Fine SAND, SP, light grey (2.5Y 7/1), iron staining 15\_ 16'-20' 36" 5 15.5'-16.0' Coarse clean SAND, SP, grey (2.5Y 5/1) 16.0'-17.0' No recovery 17 0'-19 5' Fine SAND, SP, grey (2.5Y 5/1), coarsening upward 20 20'-24' 48" 6 19.5'-20.0' Fine SAND, SP, olive yellow (2.5Y 6/8) 20.0'-24.0' Fine SAND, SP, olive yellow (2.5Y 6/8), iron staining from 23.5' to 24.0' Samples LS11-SB804-LC and LS11-SB804P-LC collected from 22.0'-24.0' Sample LS11-GW804 collected 24'-28' 48" 7 24.0'-24.5' Clay, CL, olive yellow (2.5Y 6/6) 25 24.5'-25.0' Clay, CL, dark grey (2.5Y 4/1), some organics 25.0'-28.0' Clay, Ct., yellow (2.5Y 7/6), dense

End of boring @ 28.0' bgs



PROJECT NUMBER BORING NUMBER
329752.SI.WP LS11-GP805

SHEET 1

OF 1

### **SOIL BORING LOG**

PROJECT : NAB Little Creek DRILLING CONTRACTOR : Parratt-Wolff LOCATION : Site 11

ELEVATION: 9.50 NORTHING: 3500906.49 EASTING: 12169541.48
DRILLING METHOD AND EQUIPMENT USED: DPT/4' Acetate Sleeve

DEPTH BELOW SURFACE (FT)	WATER LEVELS : 6.0' bgs	QUIPMENT USED : DPT	10/8/05 END: 10/8/05	LOGGER : A. Jones/M. Ost
NTERVAL (FT)				
RECOVERY ((b)   TEST   RESULTS   FESTLES   FG-5-6-6-7 (N)   TEST   RESULTS   FG-5-6-7 (N)   TEST   RESULTS   TEST   TEST			OGIAL DEGOTAL HOLY	OSMILLE VIO
10	RECOVERY	Y (IN) TEST */TYPE RESULTS 6"-6"-6"-6"	MOISTURE CONTENT, RELATIVE DENSITY, OR CONSISTENCY, SOIL STRUCTURE,	DRILLING FLUID LOSS, TESTS, AND INSTRUMENTATION.
4'-8' 48' 2 4.0'-5.5' SILT, ML, olive yellow (2.5Y 6/6), very dense 5.5'-8.0' Medium SAND, SP, pale yellow (2.5Y 7/3), saturated, iron staining 5.5'-8.0' No recovery 9.0'-10.0' Fine SAND, SP, yellow (2.5Y 7/6) 10.0'-12.0' Fine SAND, SP, yellow (2.5Y 6/8) 5.0'-12.0' Fine SAND, SP, olive yellow (2.5Y 6/8) 10.0'-12.0' Silty SAND, SM, olive yellow (2.5Y 6/8) 10.0'-12.0' bigs 15.0'-15.4' Coarse SAND with gravel, SW, fight grey (2.5Y 7/1) 15.4'-15.6' Coarse SAND with gravel, SW, fight grey (2.5Y 6/8) 15.6'-16.0' Coarse SAND with gravel, SW, olive (2.5Y 6/8) 15.6'-16.0' Coarse SAND with gravel, SW, olive (2.5Y 6/8) 17.5'-19.5' Silty CLAY, ML, pale yellow (2.5Y 6/8) 17.5'-19.5' Silty CLAY, ML, pale yellow (2.5Y 7/1), Sample LS11-SB805-LC collected very gravel, mottled with red inclusions 20.0'-22.5' Fine silty SAND, SM, light grey (2.5Y 7/1), Sample LS11-SB805-LC collected very gravel, mottled with red inclusions 20.0'-22.5' Fine silty SAND, SM, light grey (2.5Y 7/1), Sample LS11-SB805-LC collected very gravel, mottled with red inclusions 20.0'-22.5' Fine silty SAND, SM, light grey (2.5Y 7/1), Sample LS11-SB805-LC collected very gravel, mottled with red inclusions 20.0'-22.5' Fine silty SAND, SM, light grey (2.5Y 7/1), Sample LS11-SB805-LC collected very gravel, with organics, dark grey (2.5Y 7/1), Sample LS11-SB805-LC collected very gravel, with organics, CL				***
5.5 SILT, ML, olive yellow (2.5Y 6/6), very dense 5.5'-8.0' Medium SAND, SP, pale yellow (2.5Y 7/3), saturated, iron staining    8'-12' 36' 3 8.0'-9.0' No recovery   9.0'-10.0' Fine SAND, SP, yellow (2.5Y 7/6)   10.0'-12.0' Fine SAND, SP, olive yellow (2.5Y 6/8)   Sample LS11-SB805-UC collected from 10.0'-12.0' bgs    12'-16' 36' 4 12.0'-13.0' No recovery   13.0'-15.0' Silty SAND, SM, olive yellow (2.5Y 6/8), dark purple stain at 13.5' 15.0'-15.4' Coarse SAND with gravel, SW, light grey (2.5Y 7/1) 15.4'-15.6' Coarse SAND with gravel, SW, olive (2.5Y 4/4) 15.6'-16.0' Coarse SAND with gravel, SW, olive (2.5Y 6/8) 16.0'-17.5' Fine SAND, SP, olive yellow (2.5Y 6/8) 17.5'-19.5' Silty CLAY, ML, pale yellow (2.5Y 6/8) 17.5'-19.5' Silty CLAY, ML, pale yellow (2.5Y 7/1), Sample LS11-SB805-LC collected 22.5'-24.0' Silty CLAY with organics, dark grey (2.5Y 7/1), Sample LS11-SB805-LC collected 22.5'-24.0' Silty CLAY with organics, dark grey (2.5Y 4/1)	_ 0'-4' 48* - -	1	_	- - -
8'-12' 36" 3 8.0'-9.0' No recovery 9.0'-10.0' Fine SAND, SP, yellow (2.5Y 7/6) 10.0'-12.0' Fine SAND, SP, olive yellow (2.5Y 6/8)  12'-16' 36" 4 12'-13.0' No recovery 13.0'-15.0' Silty SAND, SM, olive yellow (2.5Y 6/8), dark purple stain at 13.5' 15.0'-15.4' Coarse SAND with gravel, SW, light grey (2.5Y 7/1) 15.4'-15.6' Coarse SAND with gravel, SW, olive (2.5Y 6/8) 16'-20' 48" 5 15.6'-16.0' Coarse SAND with gravel, SW, olive yellow (2.5Y 6/8) 16.0'-17.5' Fine SAND, SP, olive yellow (2.5Y 6/8) 17.5'-19.5' Silty CLAY, ML, pale yellow (2.5Y 6/8) 17.5'-19.5' Silty CLAY, ML, pale yellow (2.5Y 7/1), Sample LS11-SB805-LC collected loose  20'-22.5' Fine silty SAND, SM, light grey (2.5Y 7/1), Sample LS11-SB805-LC collected loose  225'-24.0' Silty CLAY with organics, dark grey (2.5Y 7/1), Sample LS11-SB805-LC collected loose		2	5.5'-8.0' Medium SAND, SP, pale yellow (2.5Y 7/3),	- - -
10.0°-12.0° Fine SAND, SP, olive yellow (2.5Y 6/8)  12°-16° 36° 4  12.0°-13.0° No recovery  13.0°-15.0° Silty SAND, SM, olive yellow (2.5Y 6/8), dark purple stain at 13.5° 15.0°-15.4° Coarse SAND with gravel, SW, light grey (2.5Y 7/1), 15.4°-15.6° Coarse SAND with gravel, SW, olive (2.5Y 4/4)  15— 16°-20° 48° 5  16°-20° 48° 5  16°-20° 5 6/8)  16.0°-17.5° Fine SAND, SP, olive yellow (2.5Y 6/8) 16.0°-17.5° Fine SAND, SP, olive yellow (2.5Y 6/8) 17.5°-19.5° Silty CLAY, ML, pale yellow (2.5Y 7/3), some gravel, mottled with red inclusions 20.0°-22.5° Fine silty SAND, SM, light grey (2.5Y 7/1), Sample LS11-SB805-LC collected loose  22.5°-24.0° Silty CLAY with organics, dark grey (2.5Y 7/1), Sample LS11-SB805-LC collected 22.5°-24.0° Silty CLAY with organics, dark grey (2.5Y 7/1), Sample LS11-SB805-LC collected 24.0°-28.0° Clay with organics, CL	- 8'-12' 36"	3	8.0'-9.0' No recovery	- -
dark purple stain at 13.5'   15.0'-15.4'   Coarse SAND with gravel, SW, light grey (2.5Y 7/1)   15.4'-15.6'   Coarse SAND with gravel, SW, olive (2.5Y 4/4)   15.6'-16.0'   Coarse SAND with gravel, SW, olive yellow (2.5Y 6/8)   16.0'-17.5'   Fine SAND, SP, olive yellow (2.5Y 6/8)   17.5'-19.5'   Silty CLAY, ML, pale yellow (2.5Y 7/3), some gravel, mottled with red inclusions 20.0'-22.5'   Fine silty SAND, SM, light grey (2.5Y 7/1), Sample LS11-SB805-LC collected loose   22.5'-24.0'   Silty CLAY with organics, dark grey (2.5Y 4/1)   - 24'-28'   48"   7   24.0'-28.0'   Clay with organics, CL	-	4	-	
20'-24' 48" 6 gravel, mottled with red inclusions 20.0'-22.5' Fine silty SAND, SM, light grey (2.5Y 7/1), Sample LS11-SB805-LC collected loose 22.5'-24.0' Silty CLAY with organics, dark grey (2.5Y 4/1) - 24'-28' 48" 7 24.0'-28.0' Clay with organics, CL		5	dark purple stain at 13.5' 15.0'-15.4' Coarse SAND with gravel, SW, light grey (2.5Y 7/1) 15.4'-15.6' Coarse SAND with gravel, SW, olive (2.5Y 4/4) 15.6'-16.0' Coarse SAND with gravel, SW, olive yellow (2.5Y 6/8) 16.0'-17.5' Fine SAND, SP, olive yellow (2.5Y	- 6/8)
4/1)  - 24'-28' 48" 7 24.0'-28.0' Clay with organics, CL		6	gravel, mottled with red inclusions 20.0'-22.5' Fine silty SAND, SM, light grey (2.5Y 7/1),	Sample LS11-SB805-LC collected
		7	4/1)	- Sample LS11-SB805-YC collected -
End of boring @ 28 0' bgs	-		End of boning @ 28 0' bgs	<u>-</u>



PROJECT NUMBER BORING NUMBER LS11-GP806 329752.SI.WP

SHEET 1 OF 1

### **SOIL BORING LOG**

PROJECT: NAB Little Creek

DRILLING CONTRACTOR: Parratt-Wolff

LOCATION : Site 11

ELEVATION: 9.40 NORTHING: 3500973.76

DRILLING METHOD AND FOUIPMENT USED: DPT/4' Acetate Sleeve

NORTHING: 3500973.76

EASTING: 12169532.73

DRILLIN	G METH	OD AND	EQUIPME	ENT USED : DPT	7 4' Acetate Sleeve	
WATER	LEVELS	: 6.0 bgs		START:	10/7/05 END: 10/7/05	LOGGER: A. Jones/M. Ost
DEPTH B	ELOW SU	RFACE (F	Т)	STANDARD	CORE DESCRIPTION	COMMENTS
	INTERVA	L (FT)		PENETRATION		
		RECOVE	RY (JN)	TEST	SOIL NAME, USCS GROUP SYMBOL, COLOR,	DEPTH OF CASING, DRILLING RATE,
ŀ	i		#/TYPE	RESULTS	MOISTURE CONTENT, RELATIVE DENSITY,	DRILLING FLUID LOSS,
	l			6"-6"-6"-6"	OR CONSISTENCY, SOIL STRUCTURE,	TESTS, AND INSTRUMENTATION.
1		1	1	(N)	MINERALOGY.	OVM (ppm): Breathing Zone Above Hole
_	0'-4'	48"	1	•	0.0'-3.0' SILT, ML, Olive Brown (2.5Y 4/3), loose	-
,	İ	ŀ				
_					-	<del>.</del>
	İ					
-					3.0'-4.0' Silty CLAY, ML, light olive brown (2.5Y 5/6),	_
_	ŀ				dense	
	4'-8'	48"	2		4.0'-4.4' SILT, ML, light olive brown (2.5Y 6/6)	
5					4.4'-6.0' CLAY, CL, ofive yellow (2.5Y 6/6), dense,	_
					some silt	
I -		l	1			PID = 5 ppm -
			1		loose, some gravel	
-					_	_
I -	<b>.</b>					-
I	8'-12'	36"	3		8.0'-9.0' No recovery	Sample LS11-SB806-UC collected from 8.0'-10.0' bgs
-	ŀ	ŀ			9.0'-12.0' Medium SAND, SP, loose, iron staining	6.0-10.0 bgs
10		1			11.0'-12.0' Mediditi SAND, 3F, 10058, Iron staining	
'-		[	ŀ		7.70	-
I _		1			_	_
	l	İ				
_	40/ 40/	40"	1 .			_
1	12'-16'	48"	4		12.0'-13.0' Medium SAND, SP, loose	
_						-
	1	ł			loose	
_						
15_					15.0'-15.3' Medium SAND, SP, light grey (2.5Y 6/6),	_
					loose	
-	16-'20'	48"	5	•	15.3'-16.0' Medium SAND, SP, light grey (2.5Y 7/2) 16.0'-20.0' Gravelly, medium to coarse SAND, SW,	Sample LS11-SB806-LC collected from
	10-20	70			light olive brown (2.5Y 5/3), saturated, loose	16.0'-18.0' bgs
I -					, , , ,	_
_			]		_	
1	1					
-					-	-
20						·
1 20-	20'-24'	48"	6		_	Sample LS11-SB806-YC collected
1	]				20.0'-24.0' CLAY, CL, dark drey (2.5Y 4/1), sheli hash	•
<u> </u>					· · · · · · · · · · · · · · · · · · ·	
_						_
I						
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1						
					End of boring @ 24.0' bgs	
25						
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-					_	-
l						
-					_	-
1						
i -					_	_

PROJECT NUMBER 329752.SI.WP BORING NUMBER LS11-GP807

SHEET 1

OF 1

### **SOIL BORING LOG**

DBO IEC	T - NIAD I	Little Cree	a k	DBILLIN	IG CONTRACTOR : Parratt-Wolff LOCATION : Site 11	
	T : NAB   ION : 9.5		ek		G CONTRACTOR : Parratt-Wolff LOCATION : Site 11  NG: 3500919.18 EASTING: 12169563.42	
			EQUIPME		7 4' Acetate Sleeve	
	LEVELS			START:		
DEPTH B	ELOW SU	RFACE (F	T)	STANDARD	CORE DESCRIPTION COMMENTS	
	INTERVA	L (FT) RECOVE		PENETRATION TEST RESULTS 6"-6"-6" (N)	SOIL NAME, USCS GROUP SYMBOL, COLOR, MOISTURE CONTENT, RELATIVE DENSITY, OR CONSISTENCY, SOIL STRUCTURE, MINERALOGY.  DEPTH OF CASING, DRILLING RATE, DRILLING FLUID LOSS, TESTS, AND INSTRUMENTATION. OVM (ppm): Breathing Zone Above Hole	
_	0'-4'	48°	1		0.0'-0.5' Topsoil	-
- 5	4'-8'	48*	2		3.0'-4.0' Construction fill  4.0'-4.6' Clayey SILT, ML, light olive brown (2.5Y 5/6) some construction fill  4.6'-7.2' Clayey SILT, ML, grey (2.5Y 6/1), saturated	-
- - - 10	8'-12'	48 <b>"</b>	3		4.6'-7.2' Clayey SILT, ML, grey (2.5Y 6/1), saturated  7.2'-8.0' Silty SAND with gravel, SW, light olive brown (2.5Y 5/4)  8.0'-11.0' Clean SAND, SP, olive yellow (2.5Y 6/6)	-
- - - 15_	12'-16'	36"	4		11.0'-12.0' Clean SAND, SP, pale yellow (2.5Y 7/3) — 12.0'-13.0' No recovery ————————————————————————————————————	-
-	16'-20'	42"	5		16.0'-16.5' No Recovery 16.5'-18.0' Medium SAND, SP, greyish brown (2.5Y 5/2) 18.0'-20.0' Coarse SAND, SP, greyish brown (2.5Y 5/2)	-
- 20 - -	20'-24'	48"	6		20.0'-21.0' Coarse SAND, SW, grey (2.5Y 5/1)  Sample LS11-SB807-LC collected from 20.0'-22.0' bgs  21.0'-21.5' Clayey SILT, ML, olive brown (2.5Y 4/3) 21.5'-24.0' CLAY with shells, CL, loose  Sample LS11-SB807-LC collected from 22.0'-24.0' bgs	-
25					End of boring @ 24.0' bgs	<u> </u>
- ; -					- - -	-

**Attachment B** 

	16		<del></del>				ia Beach, Virginia					T	T	<del></del>	
Station ID	LS11-GP804	LS11-MW01T	LS11-MW02S	LS11-MW03T	LS11-M		1044144055.054	LS11-MW05D	1044 1040500 050	LS11-MW05S	LS11-MW06D	LS11-MW07D	LS11-MW08D	L C44 ABAGOD OSA	LS11-MW09D LS11-MW09DP-05A
Sample ID	LS11-GW804-LC	LS11-MW01T-05A	LS11-MW02S-05A	LS11-MW03T-05A	LS11-MW04D-05A	LS11-MW04D-05D	LS11-MW05D-05A	LS11-MW05D-05D	LS11-MW05DP-05D	LS11-MW05S-05A	LS11-MW06D-05A	LS11-MW07D-05A	LS11-MW08D-05A	LS11-MW09D-05A	03/30/05
Sample Date	10/07/05	04/01/05	04/01/05	03/31/05	03/30/05	10/10/05	03/30/05	10/10/05	10/10/05	03/30/05	03/30/05	03/30/05	03/30/05	03/30/05	03/30/05
Chemical Name								ļ							
	<b>_</b>										<del></del>				
Volatile Organic Compounds (UG/L)					200		400			40.11	40.11	40.11		200	240
1,1,1-Trichloroethane	2 J	1 J	10 U	10 U	320	64	190	28	27	10 U	10 U	10 U	,1 J	280	240
1,1,2,2-Tetrachloroethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
1,1,2-Trichloro-1,2,2-trifluoroethane(Freon-113)	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
1,1,2-Trichloroethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U		10 U	2 J	150	150 J
1,1-Dichloroethane	2.5 J	35	10 U	10 U	600	340	280	160	180 8.8 J	10 U	10 U	10 U	4 J	220	240
1,1-Dichloroethene	10 U	2 J	10 U	10 U	140	77	29 10 U	8 J 10 U	8.8 J	10 U	1 J 10 U	10 U	10 U	10 U	10 U
1,2,4-Trichlorobenzene	10 U	10 U	10 U	10 U	10 U	10 U		10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
1,2-Dibromo-3-chloropropane	10 U	10 R	10 R	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
1,2-Dibromoethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
1,2-Dichlorobenzene	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
1,2-Dichloroethane	10 U	10 U			3,500	NA	1,000	NA NA	NA NA	10 U	26	10 U	29	540	560
1,2-Dichloroethene (total)	NA 10 U	25 10 U	5 J 10 U	2 J 10 U	3,500 10 U	10 U	1,000 10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
1,2-Dichloropropane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
1,3-Dichlorobenzene	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	<del> </del>
1,4-Dichlorobenzene	10 U	10 U	10 U	10 U	10 U	10 U	6 J	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
2-Butanone	10 υ	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.					
2-Hexanone	7.8 J	3 J	10 U	10 U	1,800	40 J	530	200 J	200 J	10 U	10 U	10 U	10 U		10 U.
4-Methyl-2-pentanone	10 U	10 U	10 U	10 U	41	10 U	180	190	160	10 U	5 J	10 U	10 U	10 U	10 U.
Acetone	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	<del>                                     </del>
Benzene Bromodichloromethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Bromoform	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Bromomethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Carbon disulfide	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Carbon tetrachloride	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Chlorobenzene	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Chloroethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Chloroform	10 U	10 U	10 U	10 U	1 J	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	2 J	2 J
Chloromethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Cumene	10 U	10 U	10 U	10 U	10 U	10 U	10 ປ	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Cyclohexane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Dibromochloromethane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Dichlorodifluoromethane (Freon-12)	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Ethylbenzene	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Methyl acetate	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Methyl-tert-butyl ether (MTBE)	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Methylcyclohexane	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Methylene chloride	10 U	1 B	1 B	10 U	10 U	10 U	10 U	10 U	10 U	10 U	, 1 B	10 U	10 U	10 U	1 B
Styrene	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.
Tetrachloroethene	10 U	10 <u>U</u>	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U.				
Toluene	10 U	10 U		10 U	10 U	10 U	10 U	10 U	10 U	+	10 U	10 U	· · · · · · · · · · · · · · · · · · ·	<del></del>	<del></del>
Trichloroethene	22	21	10 U	10 U	29	8.8 J	10 U	10 U	10 U	10 U	30	10 U	20	1,500	1,400
Trichlorofluoromethane(Freon-11)	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U		10 U	10 U	10 U		
Vinyl chloride	10 U	49	1 J	10 U	74	100 J	400	170 J	180 J	10 U	6 J	10 U	10 U	<del> </del>	10 U.
Xylene, total	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U		10 U	10 U
cis-1,2-Dichloroethene	15	25	5 J	2 J	3,500	1,200	1,000	310	320		26	<del> </del>	<del></del>		10 U
cis-1,3-Dichloropropene	10 U	10 U		10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	<del>                                     </del>	
m- and p-Xylene	NA NA	10 U		10 U	10 U	NA NA	10 U	NA NA	NA NA	10 U	10 U	10 U	10 U	+·	10 U
o-Xylene	NA	10 U		10 U	10 U	10 U	10 U	NA 10 U	10 U	10 U	10 U	10 U	10 U	<del></del>	
trans-1,2-Dichloroethene	10 U	10 U		10 U	7 J	10 U	3 J 10 U	10 U	10 U	10 U	10 U	10 U			10 U.
trans-1,3-Dichloropropene	10 U	10 ປ	10 U	10 U	10 U	10 0	10 0	100	10 0	10 0	10 0	100	100	100	10 0

Station ID	LS11-MW13D LS11-MW13D-05A 03/29/05  10 U 10 U 10 U 10 U 10 U 10 U 10 U 10	LS11-MW14D LS11-MW14D-05A 03/29/05 10 U 10 U 10 U 10 U 10 U 10 U 10 U 10	LS11-MW17D LS11-MW17D-05A 03/29/05 10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	LS11-MW18Y LS11-MW18Y-05A 03/29/05 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	MW19Y LS11-MW19YP-05A 03/29/05 10 U 10 U 10 U	LS11-MW20Y LS11-MW20Y-05A 03/30/05 10 U 10 U 10 U 10 U	LS11-MV23D-05A 03/31/05 03/31/05 84 10 U 10 U	LS11-MW23D-05D 10/10/05 7.4 J 10 U
Sample Date	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U	03/29/05 10 U 10 U 10 U 10 U 10 U	03/30/05 10 U 10 U 10 U 10 U	03/31/05 84 10 U 10 U	10/10/05 7.4 J 10 U
Volatile Organic Compounds (UG/L)	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U	84 10 U 10 U	7.4 J 10 U
Volatile Organic Compounds (UG/L)	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U	10 U 10 U 10 U	10 U 10 U	10 U
1,1,1-Trichtoroethane       42       10 U       10 U       10 U       10 U       NA         1,1,2,2-Tetrachloroethane       10 U       10 U       10 U       10 U       10 U       NA         1,1,2-Trichloro-1,2,2-trifluoroethane(Freon-113)       10 U       10 U       10 U       10 U       10 U       NA         1,1-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,1-Dichloroethane       30       34       58       10 U       NA         1,2-Trichlorobenzene       10 U       10 U       10 U       10 U       NA         1,2-Trichlorobenzene       10 U       10 U       10 U       10 U       NA         1,2-Uchloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane (total)       NA       10 J       NA       10 U       NA       10 U       NA         1,2-Dichloroethane       10 U       10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U	10 U 10 U 10 U	10 U 10 U	10 U
1,1,1-Trichtoroethane       42       10 U       10 U       10 U       10 U       NA         1,1,2,2-Tetrachloroethane       10 U       10 U       10 U       10 U       10 U       NA         1,1,2-Trichloro-1,2,2-trifluoroethane(Freon-113)       10 U       10 U       10 U       10 U       10 U       NA         1,1-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,1-Dichloroethane       30       34       58       10 U       NA         1,2-Trichlorobenzene       10 U       10 U       10 U       10 U       NA         1,2-Trichlorobenzene       10 U       10 U       10 U       10 U       NA         1,2-Uchloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane (total)       NA       10 J       NA       10 U       NA       10 U       NA         1,2-Dichloroethane       10 U       10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U	10 U 10 U 10 U	10 U 10 U	10 U
1,1,1-Trichtoroethane       42       10 U       10 U       10 U       10 U       NA         1,1,2,2-Tetrachloroethane       10 U       10 U       10 U       10 U       10 U       NA         1,1,2-Trichloro-1,2,2-trifluoroethane(Freon-113)       10 U       10 U       10 U       10 U       10 U       NA         1,1-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,1-Dichloroethane       30       34       58       10 U       NA         1,2-Trichlorobenzene       10 U       10 U       10 U       10 U       NA         1,2-Trichlorobenzene       10 U       10 U       10 U       10 U       NA         1,2-Uchloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethane (total)       NA       10 J       NA       10 U       NA       10 U       NA         1,2-Dichloroethane       10 U       10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U	10 U 10 U 10 U	10 U 10 U	10 U
1,1,2,2-Tetrachloroethane       10 U       NA       11 U       10 U       NA       11 U       10 U       10 U       NA       11 U       NA       11 U       NA       11 U       NA       11 U       NA       11 U       NA       11 U       NA       11 U       NA       NA       11 U       NA       NA       11 U       NA       NA       12 U       NA       12 U       NA       NA       NA       12 U       NA       NA	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U	10 U 10 U 10 U	10 U 10 U	10 U	
1,1,2-Trichloroethane       10 U       10 U       10 U       10 U       10 U       NA         1,1-Dichloroethane       54       3 J       3.9 J       10 U       NA         1,1-Dichloroethene       30       34       58       10 U       NA         1,2,4-Trichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         1,2-Dibromo-3-chloropropane       10 U       10 U       10 U       10 U       10 U       NA       NA       1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA       NA       1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA       NA       1,2-Dichloroethene (total)       NA       10 U       10 U       10 U       NA       1,2-Dichloroethene (total)       NA       10 U       10 U       NA       10 U       NA       1,2-Dichloroethene (total)       NA       10 U       10 U       NA       1,2-Dichloroethene (total)       NA       10 U       10 U       NA       1,2-Dichloroethene (total)       NA       10 U       10 U       10 U       NA       1,2-Dichloroethene (total)       NA       10 U       10 U       10 U       NA       1,2-Dichloroethene (total)       NA       10 U       10 U <th>10 U 10 U</th> <th>10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U</th> <th>10 U 10 U 10 U 10 U 10 U 10 U</th> <th>10 U 10 U 10 U 10 U</th> <th>10 U 10 U 10 U</th> <th>10 U 10 U</th> <th>10 U</th> <th></th> <th>10 U</th>	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U	10 U 10 U 10 U	10 U 10 U	10 U		10 U
1,1,2-Trichloroethane       10 U       10 U       10 U       10 U       10 U       NA         1,1-Dichloroethane       54       3 J       3.9 J       10 U       NA         1,1-Dichloroethene       30       34       58       10 U       NA         1,2,4-Trichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         1,2-Dibromo-3-chloropropane       10 U       10 U       10 U       10 U       10 U       NA       NA       1,2-Dichloropenzene       10 U       10 U       10 U       10 U       NA       NA       1,2-Dichloropenzene       10 U       10 U       10 U       10 U       NA       NA       1,2-Dichloropethane (total)       NA       10 U       10 U       10 U       NA       1,2-Dichloropethene (total)       NA       10 U       10 U       NA       1,2-Dichloropethene (total)       NA       10 U       10 U       NA       10 U       NA       1,2-Dichloropethene (total)       NA       10 U       10 U       NA       1,2-Dichloropethene (total)       NA       10 U       10 U       NA       1,2-Dichloropethene (total)       NA       10 U       10 U       10 U       NA       1,2-Dichloropethene (total)       NA       10 U       10 U <t< th=""><th>10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U</th><th>10 U 10 U 10 U 10 U 10 U 10 U 10 U</th><th>10 U 10 U 10 U 10 U 10 U</th><th>10 U 10 U 10 U</th><th>10 U 10 U</th><th>10 U</th><th></th><th>1 J</th><th></th></t<>	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U	10 U 10 U	10 U		1 J	
1.1-Dichloroethene   30   34   58   10 U	10 U 10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U	10 U 10 U	10 U		10 11 1		10 U
1.2.4-Trichlorobenzene         10 U         10	10 U 10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U	10 U		40.11		760	1,300
1.2.4-Trichlorobenzene       10 U       10 U       10 U       10 U       10 U       10 U       NA         1.2-Dibromo-3-chloropropane       10 U       10 U       10 U       10 U       10 U       10 U       NA         1.2-Dibromoethane       10 U       10 U       10 U       10 U       10 U       10 U       NA         1.2-Dichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         1.2-Dichlorobethene (total)       NA       10 J       NA       10 U       NA       10 U       NA       10 U       NA         1.2-Dichloropropane       10 U       10 U       10 U       10 U       10 U       NA       10 U       NA       1.2-Dichlorobenzene       10 U       10 U       10 U       10 U       NA       1.2-Dichlorobenzene       10 U       10 U       10 U       10 U       NA       1.2-Dichlorobenzene       10 U       10 U       10 U       10 U       NA       1.2-Dichlorobenzene       10 U       10 U       10 U       10 U       NA       1.2-Dichlorobenzene       10 U       10 U       10 U       10 U       NA       1.2-Dichlorobenzene       10 U       10 U       10 U       10 U       NA       1.2-Dichlorobenzene       10	10 U 10 U 10 U 10 U 10 U 10 U	10 U 10 U 10 U 10 U	10 U 10 U			10 U	10 U	14	17
1,2-Dibromoethane	10 U 10 U 10 U 10 U	10 U 10 U 10 U	10 U	10 U i	10 U	10 U	10 U	10 U	10 U
1.2-Dichlorobenzene         10 U         NA           1.2-Dichloroethane         10 U         10 U         10 U         10 U         10 U         NA         NA         10 U         NA         NA         10 U         NA         NA         10 U         NA         NA<	10 U 10 U 10 U 10 U	10 U 10 U			10 U	10 U	10 U	200 R	10 U
1,2-Dichloroethane         10 U         10 U         10 U         10 U         10 U         NA         NA         10 U         NA         NA         10 U         NA         NA <th>10 U 10 U 10 U</th> <th>10 U</th> <th></th> <th>10 U</th> <th>10 U</th> <th>10 U</th> <th>10 U</th> <th>10 U</th> <th>10 U</th>	10 U 10 U 10 U	10 U		10 U	10 U	10 U	10 U	10 U	10 U
1,2-Dichloroethane       10 U       10 U       10 U       10 U       NA         1,2-Dichloroethene (total)       NA       10 J       NA       10 U       NA         1,2-Dichloropropane       10 U       10 U       10 U       10 U       10 U       NA         1,3-Dichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         1,4-Dichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         2-Butanone       10 U       10 U       10 U       10 U       10 U       NA         2-Hexanone       10 U       10 U       10 U       10 U       NA         4-Methyl-2-pentanone       10 U       10 U       10 U       10 U       NA         Acelone       10 U       5 J       10 U       10 U       NA         Benzene       10 U       10 U       10 U       10 U       NA	10 U 10 U		10 U	10 U	10 U	10 U	10 U	10 U	10 U
1,2-Dichloroethene (total)       NA       10 J       NA       10 U       NA         1,2-Dichloropropane       10 U       10 U       10 U       10 U       10 U       10 U       NA         1,3-Dichlorobenzene       10 U       10 U       10 U       10 U       10 U       10 U       NA         1,4-Dichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         2-Butanone       10 U       10 U       10 U       10 U       10 U       NA         2-Hexanone       10 U       10 U       10 U       10 U       10 U       NA         4-Methyl-2-pentanone       10 U       10 U       10 U       10 U       NA         Acetone       10 U       5 J       10 U       10 U       NA         Benzene       10 U       10 U       10 U       10 U       NA	10 U		10 U	10 U	10 U	10 U	10 U	2 J	2.7 J
1,2-Dichloropropane       10 U       10 U       10 U       10 U       10 U       NA         1,3-Dichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         1,4-Dichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         2-Butanone       10 U       10 U       10 U       10 U       10 U       NA         2-Hexanone       10 U       10 U       10 U       10 U       10 U       NA         4-Methyl-2-pentanone       10 U       10 U       10 U       10 U       NA         Acetone       10 U       5 J       10 U       10 U       NA         Benzene       10 U       10 U       10 U       10 U       NA		10 U	10 U	10 U	10 U	10 U	10 U	1,000	NA
1,3-Dichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         1,4-Dichlorobenzene       10 U       10 U       10 U       10 U       10 U       NA         2-Butanone       10 U       10 U       10 U       10 U       10 U       NA         2-Hexanone       10 U       10 U       10 U       10 U       10 U       NA         4-Methyl-2-pentanone       10 U       10 U       10 U       10 U       NA         Acetone       10 U       5 J       10 U       10 U       NA         Benzene       10 U       10 U       10 U       10 U       NA	10.11	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
1,4-Dichlorobenzene         10 U         10 U         10 U         10 U         10 U         NA           2-Butanone         10 U         10 U         10 U         10 U         10 U         NA           2-Hexanone         10 U         10 U         10 U         10 U         10 U         NA           4-Methyt-2-pentanone         10 U         10 U         10 U         10 U         NA           Acetone         10 U         5 J         10 U         10 U         NA           Benzene         10 U         10 U         10 U         10 U         NA	100	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
2-Butanone         10 U         10 U         10 U         10 U         10 U         NA           2-Hexanone         10 U         10 U         10 U         10 U         NA           4-Methyl-2-pentanone         10 U         10 U         10 U         10 U         NA           Acetone         10 U         5 J         10 U         10 U         NA           Benzene         10 U         10 U         10 U         10 U         NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
2-Hexanone         10 U         10 U         10 U         10 U         NA           4-Methyl-2-pentanone         10 U         10 U         10 U         10 U         NA           Acetone         10 U         5 J         10 U         10 U         NA           Benzene         10 U         10 U         10 U         10 U         NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	7 J	10 U
Acetone         10 U         5 J         10 U         10 U         NA           Benzene         10 U         10 U         10 U         10 U         NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Acelone         10 U         5 J         10 U         10 U         NA           Benzene         10 U         10 U         10 U         10 U         NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	110	160 J
Benzene 10 U 10 U 10 U NA	10 U	10 U	10 U	10_U	10 U	10 U	10 U	140	10 U
	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Bromoform 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Bromomethane 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Carbon disulfide 10 U 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Carbon tetrachloride 10 U 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Chlorobenzene 10 U 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Chloroethane 10 U 10 U 10 U 10 U NA	10 U	10 U	10 ប	10 U	10 U	10 U	10 U	10 U	1 <u>0 U</u>
Chloroform 10 U 1 J 1 4 J 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Chloromethane 10 U 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Currene 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Cyclohexane 10 U 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Dibromochloromethane 10 U 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Dichlorodifluoromethane (Freon-12) 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Ethylbenzene 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Methyl acetate 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Methyl-tert-butyl ether (MTBE) 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Methylcyclohexane 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U
Methylene chloride 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	2 B	10 U
Styrene 10 U 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	. 10 U	10 U	10 U	10 U	10 U
Tetrachloroethene 10 U 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U		10 U	10 U	10 U	10 U
Toluene 10 U 10 U 10 U NA	10 U	10 U		10 U				10 U	
Trichloroethene 140 280 370 10 U NA	10 U	10 U	<del> </del>	10 U		10 U		7 J	13
Trichlorofluoromethane(Freon-11) 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U		10 U		10 U	10 U
Vinyl chloride 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U		3,200	5,500
Xylene, total 10 U 10 U 10 U NA	10 U	10 U	<del></del>	10 U	10 U	10 U	<del></del>	10 U	
ois-1,2-Dichloroethene 360 10 J 12 10 U NA	10 U	10 U		10 U	10 U	10 U		1,000	1,100
cis-1,3-Dichloropropene 10 U 10 U 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U		10 U	10 U
m- and p-Xylene NA 10 U NA 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	<del></del>	· · · · · · · · · · · · · · · · · · ·	NA_
o-Xylene NA 10 U NA 10 U NA	10 U	10 U	10 U	10 U	10 U	10 U	10 U	10 U	NA NA
trans-1.2-Dichloroethene 10 U 10 U 10 U 10 U NA		10 U	40.11						
trans-1,3-Dichloropropene 10 U 10 U 10 U NA	10 U		10 U	10 U	10 U	10 U	10 U	13	11
100 100 100 100 100 100 100 100 100 100	10 U	10 U		10 U 10 U			10 U	<del> </del>	11 10 U

					Virginia Beach, Vir	yına			<del></del>		<del></del>	
Station ID	LS11-MW24D	LS11-N	1W25D	LS11-I	MW26D	LS11-MW27D	LS11-MW28D	LS11-MW29D	LS11-MW30D		MW36D	LS11-MW37D
Sample ID	LS11-MW24D-05A	LS11-MW25D-05A	LS11-MW25D-05D	LS11-MW26D-05A	LS11-MW26D-05D	LS11-MW27D-05A	LS11-MW28D-05A	LS11-MW29D-05A	LS11-MW30D-05A	LS11-MW36D-05A	LS11-MW36DP-05A	LS11-MW37D-05A
Sample Date	03/31/05	03/31/05	10/11/05	03/31/05	10/11/05	03/31/05	03/30/05	03/31/05	03/31/05	04/01/05	04/01/05	04/01/05
Chemical Name												
			-									
Volatile Organic Compounds (UG/L)												
1,1,1-Trichloroethane	50 U	32,000	27,000 L	17	190 J	10 UJ	12,000	1,400	2,100	10 U	10 U	10 U
1,1,2,2-Tetrachloroethane	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
1,1,2-Trichloro-1,2,2-trifluoroethane(Freon-113)	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
1,1,2-Trichloroethane	50 U	2,000 U	14	10 U	10 U	10 UJ	15	100 U	200 U	10 U	10 U	10 U
1,1-Dichloroethane	10 J	12,000	12,000 L	170	940 L	520	4,000	920	2,600	10 U	10 U	4 J
1,1-Dichloroethene	8 J	2,700	3,900 L	1 J	10 J	10 UJ	2,700	87 J	390	10 U	10 U	70
1,2,4-Trichlorobenzene	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 ป	10 U	10 U	10 U
1,2-Dibromo-3-chloropropane	50 U	2,000 U	10 U	10 U	10 U	50 R	10 U	100 R	5,000 R	10 R	10 R	40 R
1,2-Dibromoethane	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
1,2-Dichlorobenzene	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
1,2-Dichloroethane	50 U	2,000 U	10 U	10 U	1.9 J	10 UJ	23	100 ປ	200 U	10 U	10 U	10 ປ
1,2-Dichloroethene (total)	160	300,000	NA	19	NA NA	25 J	60,000	3,600	47,000	10 U	10 U	3 J
1,2-Dichloropropane	50 U	2,000 U	22	10 U	10 U	10 UJ	8 J	100 U	200 U	10 U	10 U	10 U
1,3-Dichlorobenzene	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
1,4-Dichlorobenzene	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
2-Butanone	26 J	2,000 U	10 U	4 J	45 J	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
2-Hexanone	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
4-Methyl-2-pentanone	1,900	1,500 J	2,200 J	640 J	1,600 L	37 J	3,000	1,000	610	10 U	10 U	10 U
Acetone	51	1,500 J	1,100 J	51	820 L	10 UJ	370	240	340	10 U	10 U	10 U
Benzene	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Bromodichloromethane	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	2 J	2 J	10 U
Bromoform	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Bromomethane	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Carbon disulfide	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Carbon tetrachloride	50 U	2,000 U	1,300 J	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Chlorobenzene	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Chloroethane	970	2,000 U	10 U	10 U	10 U	32 J	10 U	100 U	200 U	10 U	10 U	10 U
Chloroform	50 U	2,000 U	10 U	10 U	10 U	10 UJ	13	100 U	200 U	6 J	6 J	10 ປ
Chloromethane	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Cumene Cyclohexane	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Dibromochloromethane	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Dichlorodifluoromethane (Freon-12)	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
····································	50 U	2,000 U	3.8 B	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Ethylbenzene Methyl acetate	50 U	2,000 U	3.8 B	10 U	16 J	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Methyl-tert-butyl ether (MTBE)	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
	50 U	2,000 U	2 J	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Methylcyclohexane	12 B	2,800 0	7,100 L	10 U	10 U	2 B	470	23 B	140 J	10 U	10 U	2 B
Methylene chloride	50 U	2,800 U	7,100 L 10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Styrene	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Tetrachloroethene	50 U	2,000 U	16	10 U		10 UJ	5 J	100 U	200 U		2 J	5 J
Trichlereethans	50 U	2,000 U	1,200 J	10 U	6.7 J	2 J	24	100 U	53 J	3 J	2 J	230
Trichlorofthoromethono(Front 11)	50 U	2,000 U	1,200 J	10 U		10 UJ	10 U	100 U	200 U	10 U	10 U	10 U
Trichlorofluoromethane(Freon-11)	<u> </u>		7,600 L	280 J	3,500 L	140 J	2,300	4,400	5,400	10 U	10 U	10 U
Vinyl chloride	3,000	1,600 J			<del></del>	10 UJ	2,300 10 U	100 U	200 U	<del></del>	10 U	10 U
Xylene, total	50 U	2,000 U	16 B	10 U	330 L	25 J	60,000	3,000	47,000	10 U	10 U	3 J
cis-1,2-Dichloroethene	160	300,000	260,000 L	19	<del> </del>			3,000 100 U	47,000 200 U		10 U	10 U
cis-1,3-Dichloropropene	50 U	2,000 U	10 U	10 U		10 UJ	10 U	<del>                                       </del>	200 U	10 U	10 U	10 U
m- and p-Xylene	50 U	2,000 U	NA NA	10 U	+	10 UJ	10 U				10 U	10 U
o-Xylene	50 U	2,000 U	NA .	10 U	<del> </del>	10 UJ	10 U	100 U	200 U		10 U	10 U
trans-1,2-Dichloroethene	50 U	550 J	300 J	10 U	<del></del>	10 UJ	450	100 U	120 J	10 U	<del> </del>	
trans-1,3-Dichloropropene	50 U	2,000 U	10 U	10 U	10 U	10 UJ	10 U	100 U	200 U	10 U	10 U	10 U

NAB	Little C	reek	
/iroinia	Reach	Virginia	

						Virgit	nia Beach, Virginia				·				1.044.1411/000
Station ID	LS11-GP804	LS11-MW01T	LS11-MW02S	LS11-MW03T	LS11-1	AW04D		LS11-MW05D		LS11-MW05S	LS11-MW06D	LS11-MW07D	LS11-MW08D		LS11-MW09D LS11-MW09DP-05A
Sample ID	LS11-GW804-LC	LS11-MW01T-05A	LS11-MW02S-05A	LS11-MW03T-05A	LS11-MW04D-05A	LS11-MW04D-05D	LS11-MW05D-05A	LS11-MW05D-05D	LS11-MW05DP-05D	LS11-MW05S-05A	LS11-MW06D-05A	LS11-MW07D-05A	LS11-MW08D-05A	LS11-MW09D-05A	03/30/05
Sample Date	10/07/05	04/01/05	04/01/05	03/31/05	03/30/05	10/10/05	03/30/05	10/10/05	10/10/05	03/30/05	03/30/05	03/30/05	03/30/05	03/30/05	03/30/05
Chemical Name															
Total Metals (UG/L)		-		<u> </u>											
Aluminum	NA NA	NA NA	NA NA	NA	NA	NA	NA NA	NA.	NA NA	NA NA	NA NA	NA_	NA NA	NA	NA NA
Antimony	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA NA	NA NA	NA	NA_	NA	NA NA	NA_
Arsenic	NA NA	NA NA	NA	NA.	NA	NA	NA	NA	NA NA	NA NA	NA	NA	NA NA	NA NA	NA NA
Barium	NA NA	NA	NA NA	NA	NA	NA	NA	NA NA	NA_	NA NA	NA NA	NA	NA	NA NA	NA NA
Beryllium	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA NA	NA NA	NA NA	NA_	NA NA	NA_	NA NA	NA NA
Cadmium	NA NA	NA	NA NA	NA	NA	NA	NA	NA NA	NA_	NANA	NA	NA_	NA NA	NA NA	NA NA
Calcium	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA	NA NA	NA NA	NA.	NA NA	NA	NA NA
Chromium	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA NA	NA NA	NA.	NA NA	NA	NA
Cobalt	NA NA	NA	NA NA	NA NA	NA NA	NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA_
	NA NA	NA NA	NA NA	NA.	NA	NA	NA	NA	NA_	NA NA	NA	NA_	NA NA	NA NA	NA
Copper Cyanide	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA NA	NA	NA_	NA NA	NA NA	NA
ron	NA NA	NA NA	NA NA	7,620 J	15,900 J	NA.	48,800 J	NA	NA NA	NA	NA NA	290 J	511 J	454 J	NA
Lead	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA	NA	NA NA	NA NA	NA_	NA NA	NA NA
	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA	NA NA	NA NA	NA_	NA NA	NA NA	NA NA
Magnesium Manganese	NA NA	NA NA	NA NA	731 J	3,860	NA NA	7,180	NA	NA	NA	NA NA	640	52.7	972	NA _
Mercury	NA NA	NA NA	NA.	NA	NA	NA.	NA	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA
Nickel	NA NA	NA.	NA.	NA	NA NA	NA	NA	NA	NA	NA NA	NA	NA_	NA NA	NA NA	NA NA
Potassium	NA NA	NA NA	NA NA	NA	NA	NA.	NA	NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA
Selenium	NA NA	NA NA	NA.	NA	NA NA	NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA_	NA NA	NA
Silver	NA NA	NA NA	NA NA	NA	NA NA	NA	NA	NA NA	NA	NA	NA NA	NA.	NA NA	NA NA	NA NA
Sodium	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA	NA NA	NA NA	NA	NA
Thallium	NA NA	NA NA	NA NA	NA NA	NA NA	NA.	NA	NA	NA	NA NA	NA	NA_	NA NA	NA NA	NA
Vanadium	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA_	NA
	NA NA	NA NA	NA NA	NA NA	NA.	NA	NA.	NA	NA NA	NA NA	NA NA	NA NA	NA_	NA NA	NA NA
Zinc								<u> </u>							<b></b>
Dissolved Metals (UG/L)										<u>                                     </u>	ļ				
Aluminum	NA NA	NA NA	NA NA	NA	NA	NA NA	NA	NA	NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA
Antimony	NA NA	NA NA	NA NA	NA NA	NA.	NA	NA	NA NA	NA NA	NA NA	NA _	NA NA	NA NA	NA NA	NA NA
Arsenic	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA.	NA	NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA
Barium	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA_	NA NA	NA NA
Berytlium	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA.	NA NA	NA NA	NA_	NA NA	NA NA	NA NA	NA
Cadmium	NA NA	NA NA	NA NA	NA NA	NA NA	NA.	NA	NA.	NA NA	NA	NA NA	NA_	NA NA	NA	NA NA
Calcium	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA_	NA_	NA NA	NA NA	NA	NA NA
Chromium	NA NA	NA NA	NA NA	NA	NA NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA
Cobalt	NA NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA	NA_	NA NA	NA NA	NA NA
Copper	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA	NA_	NA NA	NA NA	NA	NA NA
Iron	NA NA	NA NA	NA NA	7,740 J	16,700	NA	47,100	NA NA	NA NA	NA NA	12.4 B	33.7 B	11.7 B	<del></del>	NA NA
Lead	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA	NA NA
Magnesium	NA NA	NA NA	NA NA	NA	NA NA	NA	NA	NA	NA NA	NA_	NA NA				
Manganese	93.6	NA NA	NA NA	810 J	3,930	3,590	7,260	6,900	6,830	NA NA	0.91 U	65.6	27	916	NA NA
	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA	NA	NA NA	NA NA	NA_	NA_	NA_	NA NA	NA NA
Nickel	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA.	NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
Potassium	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA	NA
Selenium	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA NA	NA NA	NA_	NA NA	NA NA
	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA NA	NA NA	NA_	NA NA	NA NA	NA
Silver Sodium	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA
Sodium Thallium	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA	NA	NA	NA NA	NA	NA NA	NA_	NA NA	NA NA
	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA	NA	NA NA	NA NA	NA NA	NA NA	NA _	NA NA
Vanadium	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA	NA.	NA	NA	NA	. NA	NA NA	NA NA	NA NA
Zinc	- NA	NA NA	INA.		† ·	1	1						1		<u> </u>

Second   S	
Seminary   Seminary	
Semigen   10-1000   10-2	LS11-MW23D-05D
The second sec	10/10/05
Total Market (GRC)	
Montemy	
March   Marc	NA NA
Margaret   Mart   Mar	NA NA
March   Marc	NA
Service   March   Ma	NA
March   Marc	NA
Common	NA
Calcium   No.	NA
Common	NA.
Cabal	NA NA
Compose	NA
Cyangle   No.	NA
From   No.   107	NA NA
Bard	NA
Magnesism	NA_
Margariese	NA
Merclay	NA NA
Palassum	NA
Profession	NA NA
Selection	NA
Silver	NA_
Sodium	NA NA
Institution	NA_
Na	NANA
Company   Comp	NA_
Aluminum	
Aluminum NA NA NA NA NA NA NA NA NA NA NA NA NA	
Allimory	NA
Arsenic	NA
Architecture	NA NA
Sarulm	NA NA
Seryillum	NA NA
Calcium	NA NA
Calculation	NA
Cobalt	NA NA
Cobalt         NA         NA <th< td=""><td>NA NA</td></th<>	NA NA
Copper	NA NA
	NA NA
Lead NA NA NA NA NA NA NA NA NA NA NA NA NA	NA NA
	NA 110
Magnesium 148 151 128 1511 162.5 1.4.B NA NA NA NA NA NA NA NA NA NA NA NA NA	142
Manganese 2,500 312 12 12 NA NA NA NA NA NA NA NA NA NA NA NA NA	NA NA
Mercury NA NA NA NA NA NA NA NA NA NA NA NA NA	NA NA
NICKEI NA NA NA NA NA NA NA NA NA NA NA NA NA	NA.
POLSSIUM NA NA NA NA NA NA NA NA NA NA NA NA NA	NA NA
Selenium NA NA NA NA NA NA NA NA NA NA NA NA NA	NA NA
Silver NA NA NA NA NA NA NA NA NA NA NA NA NA	NA NA
SOGIUM NA NA NA NA NA NA NA NA NA NA NA NA NA	NA NA
Thaillium NA NA NA NA NA NA NA NA NA NA NA NA NA	NA NA
Vanadium NA NA NA NA NA NA NA NA NA NA NA NA NA	NA_
Zinc NA NA NA NA NA NA NA NA NA NA NA NA NA	L

										, <del></del>		
Station ID	LS11-MW24D	LS11-N			MW26D	LS11-MW27D	LS11-MW28D	LS11-MW29D	LS11-MW30D	<del></del>	MW36D	LS11-MW37D
Sample ID	LS11-MW24D-05A	LS11-MW25D-05A	LS11-MW25D-05D	LS11-MW26D-05A	LS11-MW26D-05D	LS11-MW27D-05A	LS11-MW28D-05A	LS11-MW29D-05A	LS11-MW30D-05A	LS11-MW36D-05A	LS11-MW36DP-05A	LS11-MW37D-05A
Sample Date	03/31/05	03/31/05	10/11/05	03/31/05	10/11/05	03/31/05	03/30/05	03/31/05	03/31/05	04/01/05	04/01/05	04/01/05
Chemical Name				·	<u>-</u>							
Total Metals (UG/L)												
Aluminum	NA NA	NA NA	NA	NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA
Antimony	NA	NA	NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA	NANA	NA NA	NA NA
Arsenic	NA	NA NA	NA NA	NA NA	, NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA	NA .
Barium	NA NA	NA NA	NA .	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA
Beryllium	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA				
Cadmium	NA	NA NA	NA	NA	NA NA	NA	NA NA	NA	NA.	NA_	NA NA	NA
Calcium	NA	NA NA	NA NA	NA NA	NA NA	NA	NA NA	NA NA	NA NA	NA NA	NA NA	NA NA
Chromium	NA	NA NA	NA NA	NA NA	NA							
Cobalt	NA	NA.	NA	NA	NA	NA	NA	NA ·	NA	NA	NA	NA
Copper	NA	NA NA	NA	NA	NA NA	NA NA	NA	NA NA	NA	NA NA	NA_	NA
Cyanide	NA	NA NA	. NA	NA								
lron	NA	71,900 J	NA	10,500 J	NA	NA	NA	NA	, NA	174 J	142 J	630 J
Lead	NA NA	NA	NA	NA	NA	NA	NA	NA -	NA	NA	NA	NA
Magnesium	NA.	NA NA	NA	NA	NA NA	NA	NA	NA	NA	NA NA	NA	NA
Manganese	NA	8,880 J	NA	871 J	NA	NA	NA	NA NA	NA	61.4 J	58.2 J	160 J
Mercury	NA	NA	NA	NA	NA	NA	NA NA	NA NA	NA	NA	NA NA	NA
Nickel	NA	NA.	NA	NA	NA	NA	· NA	NA .	NA	NA	NA	NA
Potassium	NA NA	NA NA	NA	NA	NA	NA	NA	NA NA	NA	NA NA	NA	NA NA
Selenium	NA	NA NA	NA	NA	NA	NA						
Silver	NA NA	NA.	NA	NA	NA NA	NA	NA	NA ·	NA	NA NA	NA	NA
Sodium	NA	NA NA	NA	NA NA	NA	NA .						
Thallium	NA	NA NA	NA .	NA								
Vanadium	NA	NA NA										
Zinc	NA	NA.	NA	NA	NA	NA	NA	NA NA	NA	NA NA	NA	NA
							- "					
Dissolved Metals (UG/L)												
Aluminum	NA	NA.	NA	NA	NA	NA NA	NA	NA NA	NA	NA	NA	NA
Antimony	NA NA	NA	NA	NA	NA.	NA	NA	NA NA	NA	NA NA	NA	NA NA
Arsenic	NA	NA	NA	NA.	NA.	NA	NA	NA NA	NA	NA NA	NA NA	NA
Barium	. NA	NA .	NA	NA	NA	NA	NA	NA NA	NA	NA NA	NA	NA
Beryllium	NA NA	NA	NA	NA	NA	NA	NA	NA	NA	NA NA	NA	NA
Cadmium	NA	NA	NA	NA	NA.	NA	NA	NA.	NA	NA NA	NA	NA .
Calcium	NA	NA NA	NA	NA	NA NA	NA NA						
Chromium	NA	NA	NA	NA	NA NA	NA	NA.	NA.	NA	NA	NA	NA
Cobalt	NA	NA	NA	. NA	NA	NA	NA	NA.	NA	NA	NA	NA NA
Copper	NA	NA	- NA	NA	NA	NA	NA	N/A	NA	NA NA	NA	NA
Iron	NA	67,000 J	NA	9,120 J	NA	NA	NA	NA NA	NA NA	19.2 B	17.7 B	39.9 B
Lead	NA NA	NA	NA	NA NA	NA	NA	NA	NA .	NA NA	NA	NA	NA
Magnesium	NA NA	NA .	NA	NA NA	NA	NA.	NA NA	NA.	NA NA	NA NA	NA NA	NA NA
Manganese	NA	8,610 J	9,150	1,110 J	3,340	NA.	NA	NA NA	NA NA	52.9 J	53.1 J	156 J
Mercury	NA NA	NA NA	NA	NA NA	NA	NA	NA NA	NA.	NA	NA NA	NA	NA NA
Nickel	NA NA	NA NA	NA	NA NA	NA	NA.	NA	NA NA	NA	NA NA	NA	NA NA
Potassium	NA NA	NA NA	NA.	NA NA	NA	NA.	NA	NA NA	NA.	NA NA	NA NA	NA.
Selenium	NA NA	NA.	NA.	NA NA	NA	NA NA						
Silver	NA NA	NA	NA.	NA.	NA NA	NA NA	NA NA					
Sodium	NA NA	NA	NA NA	NA NA	NA.	NA NA	NA NA	NA NA				
Thallium	NA NA	NA.	NA NA	NA NA	NA NA	NA NA	NA NA					
Vanadium	NA NA	NA NA										
Zinc	NA NA	NA .	NA NA	NA NA								
Z.IIIQ	T- NA	I INA	11/1	14/	14/1	180	1	13/2	1471	INC	11/0	
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Virginia Beach, Virginia

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Station ID	LS11-GP804	LS11-MW01T	LS11-MW02S	LS11-MW03T	LS11-1	MW04D		LS11-MW05D		LS11-MW05S	LS11-MW06D	LS11-MW07D	LS11-MW08D		LS11-MW09D
Sample ID	LS11-GW804-LC	LS11-MW01T-05A	LS11-MW02S-05A	LS11-MW03T-05A	LS11-MW04D-05A	LS11-MW04D-05D	LS11-MW05D-05A	LS11-MW05D-05D	LS11-MW05DP-05D	LS11-MW05S-05A	LS11-MW06D-05A	LS11-MW07D-05A	LS11-MW08D-05A	LS11-MW09D-05A	LS11-MW09DP-05A
Sample Date	10/07/05	04/01/05	04/01/05	03/31/05	03/30/05	10/10/05	03/30/05	10/10/05	10/10/05	03/30/05	03/30/05	03/30/05	03/30/05	03/30/05	03/30/05
Chemical Name															
Wet Chemistry (MG/L)															
Alkalinity	14	NA	NA	NA.	NA	180	NA	94	96	NA NA	NA NA	NA_	NA NA	NA NA	NA
Chloride	26	NA	NA.	NA	NA.	55 L	NA	44 L	38 L	NA NA	NA NA	NA NA	NA	NA NA	NA
Ethane	0.0062 ป	NA NA	NA	0.01 U	0.01 U	0.0062 U	0.01 U	0.0062 U	0.0062 U	NA	NA_	0.01 ป	0.01 U	0.01 U	NA NA
Ethene	0.0058 U	NA	NA NA	0.01 U	0.074	0.066	0.18	0.061	0.072	NA NA	NA	0.01 U	0.01 U	0.01 U	NA
Methane	0.0063 J	NA	NA	0.19	0.49	0.32 J	0.56	0.15 J	0.17 J	NA	NA ·	0.081	0.069	0.037	NA
Nilrale	0.1 UJ	NA	NA NA	0.05 U	0.05 U	0.14	0.42	0.16	0.15	NA NA	NA	0.58	0.3	0.53	NA
Nitrite	NA.	NA	NA	0.05 บ	0.0056 J	NA NA	0.022 J	NA	NA	NA NA	NA	0.05 U	0.05 ป	0.05 U	NA
Nitrogen	0.1 UJ	NA	NA	NA	NA	0.1 U	NA	0.1 U	0.1 U	NA NA	NA	NA NA	NA NA	NA NA	NA
Sulfate	11	NA	NA	19	21	20	1.7	16	14	NA	NA	17	19	25	NA
Sulfide	NA NA	NA NA	NA	1 U	1 U	NA	1 ປ	NA	NA	NA	NA	0.25 J	1 U	1 U	NA NA
Total organic carbon (TOC)	4	NA.	NA.	13	570	520	260	220	230	NA	NA	0.71 J	4.5	210	NA

- U- Analyte not detected
- J- Reported value is estimated
- UJ- Analyte not detected. Quantitation limit may be imprecise
- L- Reported value is estimated
- B- Possible blank contamination
- NA- Not analyzed
- "P" Identifier on sample ID indicates a duplicate sample
- R- Unreliable result

						Virginia Deach, Virg	3*****							
Station ID		LS11-I	MW10D	LS11-	MW11D	LS11-MW13D	LS11-MW14D	LS11-MW17D	LS11-MW18Y	LS11-	MW19Y	LS11-MW20Y	LS11-N	AW23D
Sample ID	LS11-MW09D-05D	LS11-MW10D-05A	LS11-MW10D-05D	LS11-MW11D-05A	LS11-MW11DP-05A	LS11-MW13D-05A	LS11-MW14D-05A	LS11-MW17D-05A	LS11-MW18Y-05A	LS11-MW19Y-05A	LS11-MW19YP-05A	LS11-MW20Y-05A	LS11-MW23D-05A	LS11-MW23D-05D
Sample Date	10/10/05	03/29/05	10/10/05	03/29/05	03/29/05	03/29/05	03/29/05	03/29/05	03/29/05	03/29/05	03/29/05	03/30/05	03/31/05	10/10/05
Chemical Name														
Wet Chemistry (MG/L)														
Alkalinity	70	NA	8.3	NA	NA	NA NA	NΑ	NA NA	NA NA	NA NA	NA NA	NA	NA NA	87
Chloride	31 L	NA	49 L	NA NA	NA	NA	NA	NA	NA	NA	NA	NA	NA.	59 L
Ethane	0.0062 U	0.01 U	0.0062 U	0.01 U	NA	0.01 U	NA	NA	NA NA	NA NA	NA	NA	0.01 U	0.0062 U
Ethene	0.0058 U	0.01 U	0.0058 U	0.01 U	NA	0.01 U	NA	NA	NA	NA	NA	NA NA	0.42	0.46
Methane	0.013 J	0.01 U	0.011 J	0.01 U	NA	0.01 U	NA	NA NA	NA	NA NA	NA	NA NA	0.74	0.86 J
Nitrate	0.1 U	1	0.18	1.8	NA	1.5	NA	NA	NA	· NA	NA NA	NA.	0.05 ป	0.15
Nitrite	NA	0.05 U	NA	0.05 ป	NA	0.054	NA	NA	NA	NA	NA	NA NA	0.014 J	NA
Nitrogen	0.1 ป	NA	0.1 U	NA	NA	NA	NA	NA	NA	NA	NA NA	NA NA	NA .	0.1 U
Sulfate	30	41	44	21	NA	13	NA NA	NA	NA	NA	NA	NA	5 U	0.1 U
Sulfide	NA	1.2	NA NA	1.2	NA	1	NA	NA	NA	, NA	NA	NA	1 U	NA
Total organic carbon (TOC)	180	0.45 J	1 U	0.49 J	NA	2.4	NA	NA	NA NA	NA	NA.	NA	120	230

- U- Analyte not detected
- J- Reported value is estimated
- UJ- Analyte not detected. Quantitation limit may be imp
- L- Reported value is estimated
- B- Possible blank contamination
- NA- Not analyzed
- "P" Identifier on sample ID indicates a duplicate sample
- R- Unreliable result

					Virginia Beach, Vi	rgima						
Station ID	LS11-MW24D	LS11-	MW25D	LS11-l	MW26D	LS11-MW27D	LS11-MW28D	LS11-MW29D	LS11-MW30D	LS11-	MW36D	LS11-MW37D
Sample ID	LS11-MW24D-05A	LS11-MW25D-05A	LS11-MW25D-05D	LS11-MW26D-05A	LS11-MW26D-05D	LS11-MW27D-05A	LS11-MW28D-05A	LS11-MW29D-05A	LS11-MW30D-05A	LS11-MW36D-05A	LS11-MW36DP-05A	LS11-MW37D-05A
Sample Date	03/31/05	03/31/05	10/11/05	03/31/05	10/11/05	03/31/05	03/30/05	03/31/05	03/31/05	04/01/05	04/01/05	04/01/05
Chemical Name												
Wet Chemistry (MG/L)												
Alkalinity	NA	NA	470	NA	250	NA NA	NA NA	NA	NA	NA NA	NA_	NA NA
Chloride	NA	NA	400	NA	66	NA	NA NA	NA.	NA NA	NA.	NA NA	NA.
Ethane	NA	0.01	0.0062 U	0.01 U	0.0062 U	NA	NA	NA NA	NA NA	0.01 U	NA NA	0.01 U
Ethene	NA	0.043	0.066 J	0.15	0.44 J	NA	NA	NA	NA.	0.01 U	NA NA	0.01 U
Methane	NA	0.32	0.37 J	4.9	5.8 J	NA	NA	NA	NA	0.01 U	NA NA	0.076
Nitrate	NA	0.05 U	0.14 L	0.05 U	0.12 L	NA	NA	NA .	NA NA	0.046 J	NA NA	0.028 J
Nitrite	NA	0.14	NA NA	0.008 J	NA	NA	NA	NA	NA_	0.05 U	NA NA	0.05 U
Nitrogen	NA	NA	0.1 U	NA	0.1 U	NA NA	NA	NA	NA NA	NA	NA	NA NA
Sulfate	NA	1 U	0.12	1 U	0.24	NA	NA	NA	NA	11	NA NA	12
Sulfide	NA	0.7 J	NA	1 U	NA NA	NA	NA	NA	NA	1 U	NA	1 ป
Total organic carbon (TOC)	NA	2,600	3,600	290	720	NA	NA	NA	NA	1.3	NA	0.55 J

- U- Analyte not detected
- J- Reported value is estimated
- UJ- Analyte not detected. Quantitation limit may be imp
- L- Reported value is estimated
- B- Possible blank contamination
- NA- Not analyzed
- "P" Identifier on sample ID indicates a duplicate sample
- R- Unreliable result

Attachment B-2 Soil Anaytical Results (October 2005) Pre-Feasibility Study Investigations NAB Little Creek Virginia Beach, Virginia

	***	1044 0004	<del></del>	<del></del>	LS11-GP802		LS11-GP803	I IS11	-GP804	-	LS11-GP805			LS11-GP806			LS11-GP807	
Station ID	1011 00001110	LS11-GP801	LS11-SB801-YC	LS11-SB802-YC	LS11-SB802-LC	LS11-SB802-UC	LS11-SB803-LC	LS11-SB804-LC	LS11-SB804P-LC	LS11-SB805-UC	LS11-SB805-LC	LS11-SB805-YC	LS11-SB806-UC	LS11-SB806-LC	LS11-SB806-YC	LS11-SB807-YC	LS11-SB807-UC	LS11-SB807-LC
Sample ID	LS11-SB801-UC	LS11-SB801-LC	10/07/05	10/08/05	10/08/05	10/08/05	10/08/05	10/07/05	10/07/05	10/08/05	10/08/05	10/08/05	10/07/05	10/07/05	10/07/05	10/08/05	10/08/05	10/08/05
Sample Date	10/07/05	10/07/05	10/07/05	10/06/05	10/00/03	10/00/03	10,00,00	10.07.00			1			Ī				
Chemical Name		-						<del> </del>	<del>                                     </del>									
								<b></b>			ļ —							
Volatile Organic Compounds (UG/KG)						1			1 1			1	<u> </u>					1
1,1,1-Trichloroethane	12 U	93	16 U	15 J	16 J	12 U	12 U	12 U	13 U	12 U	12 U	14 U	12 U	12 U	14 U		3.7 J	180
1,1,2,2-Tetrachloroethane	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U	13 U	12 U	12 U	14 U	12 U	12 U	14 U	15 U	13 U	12 U
1,1,2-Trichloro-1,2,2-												l.,				45 11	13 U	12 U
trifluoroethane(Freon-113)	12 U	12 U	16 U	17 U	11 U	12 U	12 U		13 U	12 U	12 U		<del></del>	12 U	14 U		13 U	
1,1,2-Trichloroethane	12 U	12 U	16 U	17 U	11 UJ	12 U	12 U		13 U	12 U	12 U		<del></del>	12 U	14 U		4.8 J	110
1,1-Dichloroethane	2.5 J	38	2,100	3,000 J	7.3 J	7.1 J	4.9 J	12 U	13 U	12 U	12 U		<del></del>	4.1 J	1,800	600 J	4.8 J	<del></del>
1,1-Dichloroethene	12 U	5 J	210	620 J	11 U	12 U	12 U	12 U	13 U	12 U	12 U			12 U		75		
1,2,4-Trichlorobenzene	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U	13 U	12 U	12 U			12 U			13 U	4
1,2-Dibromo-3-chloropropane	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U	13 U	12 U		<del></del>	<del></del>	12 U	14 U	<del> </del>	13 U	
1,2-Dibromoethane	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U	13 U	12 U		<b></b>		12 U	+		13 U	
1,2-Dichlorobenzene	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U	13 U	12 U	12 U	<del></del>	12 U	12 U			13 U	<del></del>
1,2-Dichloroethane	12 U	12 U	16 U	6.8 J	11 U	12 U	12 U	12 U		12 U	· · · · · · · · · · · · · · · · · · ·		12 U	12 U	14 U	<del></del>	13 U	
1,2-Dichloropropane	12 U	12 U	16 U	17 U	11 UJ	12 U	12 U	12 U	13 U	12 U			12 U	12 U		+	13 U	
1.3-Dichlorobenzene	12 U	12 U	16 U	17 U	11 U	12 U	12 U		<del></del>	12 U			12 U	12 U	14 U		13 U	+
1,4-Dichlorobenzene	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U		12 U			12 U	12 U	14 U	15 U	13 U	
2-Butanone	12 U	13	16 U	17 U	11 U	12 U	12 U	12 U	13 U	12 U			12 U	12 U	14 U			<b>—————————————————————————————————————</b>
2-Hexanone	12 U	12 U	16 U	17 U	11 UJ	12 U	12 U			12 U			12 U	12 U	14 U		13 U 5.7 J	240 J
4-Methyl-2-pentanone	12 U	200	16 U	72 J	14	8.5 J	12 U	12 U	13 U	12 U				<del></del>	14 U			
Acetone	12 U	510 J	16 J	210 J	11 U	12 U	12 U	12 U	13 U	12 U					14 U	<del></del>	13 U	
Benzene	12 U	12 U	16 U	17 U	11 UJ	12 U	12 U	12 U	13 U		+						13 U	
Bromodichloromethane	12 U	12 U	16 U	17 U	11 UJ	12 U	12 U	12 U		12 U	+		-+	· · · · · · · · · · · · · · · · · · ·	14 U		13 U 13 U	+
Bromoform	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U		12 U		+	<del></del>				13 U	
Bromomethane	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U	<del></del>	12 U		<del></del>	<del></del>	<del></del>	+		13 U	
Carbon disulfide	12 U	12 U	21	40	11 U	12 U	12 U	12 U	+	12 U	+	+		<del></del>		22	13 U	
Carbon tetrachloride	12 U	12 U	16 U	17 U	11 UJ	12 U	12 U	12 U	+	12 U	+		<del></del>				13 U	
Chlorobenzene	12 U	12 U	16 U	17 U	11 U	12 U	12 U			12 U	<del> </del>	<del></del>	+	+		15 U	13 U	
Chloroethane	12 U	12 U	11 J	34	11 U	12 U	12 U		+	12 U	<del></del>	14 U		12 U	<del></del>	<del></del>	13 U	12 11
Chloroform	12 U	12 U	16 U	17 U	11 U	12 U	12 U			12 U		14 U		12 U			13 U	<del> </del>
Chloromethane	12 U	12 U	16 U	17 U	11 U	12 U	12 U	12 U		12 U	<del></del>	14 U		12 U	+	15 U	13 U	1211
Cumene	12 U	12 U	16 U	17 U	11 U	12 U	12 U			12 U	<del></del>	14 U		12 U		15 U	13 U	
Cyclohexane	12 U	12 U	16 U	17 U	11 UJ	12 U	12 U		<del></del>	12 U		14 0		12 U	<del>+</del>	15 U	13 U	
Dibromochloromethane	12 U	12 U	16 U	17 U	11 UJ	12 U	12 U	12 U	13 U	12 U	12 U	14 L	J 12 U	12 0	140	130		1
										12 U	12 U	14 (	12 U	12 U	14 U	15 U	13 U	
Dichlorodifluoromethane (Freon-12)	12 U	12 U			11 U	12 U	12 t			+	<del></del>	<u> </u>		+		15 U	13 U	12 U
Ethylbenzene	12 U	12 U			11 U	12 U	12 L					<b>+</b>				· · · · · · · · · · · · · · · · · · ·	13 U	12 U
Methyl acetate	12 U	12 U			11 U	12 U	12 L	+···						+	+·		13 U	
Methyl-tert-butyl ether (MTBE)	12 U	12 U			11 U	12 U	12 U	<del></del>				<del> </del>				<del></del>	13 U	12 U
Methylcyclohexane	12 U	12 U	<b>_</b>	17 U	11 UJ	12 U	12 \		<b>—</b> — — — — — — — — — — — — — — — — — —		+		<del> </del>	4	<del></del>	30	13 U	
Methylene chloride	12 U	22 J	530 J	240	11 U	12 U	<del></del>		<del></del>			<del></del>	+				13 U	12 U
Styrene	12 U	12 U		17 U	<del>                                     </del>	12 U			+			<del> </del>	- · · · · · · · · · · · · · · · · · · ·				13 U	12 U
Tetrachloroethene	12 U	12 U			11 U	12 U	12 L					14 (	<del></del>			15 U	13 U	12 U
Toluene	12 U			3.7 J	11 U 17 J	12 U	23	3 3		55		3.4 J		12 U	11,000	5,400 J	12 J	
Trichloroethene	6 J	22 J	18,000	25,000 J	1/J	18	23	33	23	- 33	1	1	<del></del>		1			
L		12 U	16 U	17 U	11 U	12 0	12 L	) 12 U	13 6	12 L			J 12 U			15 U	13 U	
Trichlorofluoromethane(Freon-11)	12 U	7.3 J	16 U		11 U	12 L					J 12 U	14 U			70	15 U		
Vinyl chloride					11 U	12 L				1			J 12 U	12 L				12 U
Xylene, total	12 U	600	130 J		130 J	26	16	8.2 J		<del></del>	) 12 U	14 L	J 4.2 J	14 J	7,700	600 J		370
cis-1,2-Dichloroethene	11 J		<del></del>		·	12 t		<del></del>					J 12 U	12 L				
cis-1,3-Dichloropropene	12 U			<del>                                     </del>	11 U	12 0									J 14 L	J 15 U	13 L	
trans-1,2-Dichloroethene	12 U			<del></del>	<del> </del>	12 0		<del></del>		- <del></del>		14 (	J 12 U	12 L	J 14 U	J 15 U	13 L	J 12 U
trans-1,3-Dichloropropene	12 U	12 0	100	1/10	1103		1 1	+	1									1
	<b></b>	<del> </del>	<del>   -</del>	1		<del> </del>	1	1	1 1	† · · · · · · · · · · · · · · · · · · ·	1							1
Wet Chemistry (MG/KG)	<u> </u>			60	90	83	85	82	77	82	82	72	87	85	72	66	80	80
% Solids	86	86	63	22,000	1,300	NA NA	NA NA	NA NA	NA	NA NA	1,400	13,000	NA	970	10,000	10,000	NA NA	1,600
Total organic carbon (TOC)	NA	950	12,000	22,000	1,300	14/4	14/	1	1									

NA - Not analyzed J - Reported value is estimated U - Analyte not detected UJ - Not detected, quantitation limit may be inaccurate



2340 Stock Creek Blvd. Rockford TN 37853-3044 Phone: (865) 573-8188 Fax: (865) 573-8133 Email: info@microbe.com

### **Analysis Report**

Client:

Felicia Arroyo

Phone:

(757) 671-8311

CH2M HILL

5700 Cleveland Street

Suite 101

Virginia Beach, VA 23462

Fax:

(757) 497-6885

MI Identifier: 013CJ

Date Rec: 10/08/2005

**Report Date: 10/28/2005** 

Greg a Daris

Client Project #: 329752.SI.FQ

Client Project Name: NAB Little Creek Site 11 (CTO-103)

Purchase Order #:

**Analysis Requested:** 

CENSUS (final), PLFA, VFA

Comments:

All samples within this data package were analyzed under U.S. EPA Good Laboratory Practice Standards: Toxic Substances Control Act (40 CFR part 790). All samples were processed according to standard operating procedures. Test results submitted in this data package meet the quality assurance requirements established by Microbial Insights, Inc.

Reported By:

Reviewed By:

NOTICE: This report is intended only for the addressee shown above and may contain confidential or privileged information. If the recipient of this material is not the intended recipient or if you have received this in error, please notify Microbial Insights, Inc. immediately. The data and other information in this report represent only the sample(s) analyzed and are rendered upon condition that it is not to be reproduced without approval from Microbial Insights, Inc. Thank you for your cooperation.

### MICROBIAL INSIGHTS, INC.

2340 Stock Creek Blvd. Rockford, TN 37853-3044

Tel: (865) 573-8188; Fax: (865) 573-8133

**CENSUS** 

Client:

**CH2M HILL** 

Project:

NAB Little Creek Site 11 (CTO-103)

MI Project Number:

013CJ

Date Received:

10/08/2005

### Sample Information

Client Sample ID:		LS11-SB801-LC	LS11-SB801-YC	LS11-SB806- LC	LS11-SB802-L C	LS11-SB805-L C
Sample Date:		10/07/2005	10/07/2005	10/07/2005	10/08/2005	10/08/2005
Units:		cells/g	cells/g	cells/g	cells/g	cells/g
Dechlorinating Bacteria						
Dehalococcoides spp (1)	DHC	1.11E+06	<9.58E+02	3.24E+05	2.73E+03	1.75E+03
Dehaiobacter spp.	DHB	4.01E+06	8.5E+05	7.72E+06	3.26E+06	2.65E+06
unctional Genes			·			
BAV1 VC R-Dase (1)	BVC	1.35E+05	<9.58E+02	5.1E+04	<9.52E+02	<8.28E+02

### Legend:

NA = Not Analyzed NS = Not Sampled J = Estimated gene copies below PQL but above LQL

I = Inhibited

< = Result not detected

<sup>1</sup> Bio-Dechlor Census technology was developed by Dr. Loeffler and colleagues at Georgia Institute of Technology and was licensed for use through Regenesis.

### MICROBIAL INSIGHTS, INC.

2340 Stock Creek Blvd. Rockford, TN 37853-3044

Tel: (865) 573-8188; Fax: (865) 573-8133

**CENSUS** 

Client:

**CH2M HILL** 

Project:

NAB Little Creek Site 11 (CTO-103)

**MI Project Number:** 

013CJ

Date Received:

10/08/2005

Sample Information

Client Sample ID:

LS11-SB807-LC

Sample Date:

10/08/2005

Units:

cells/g

**Dechlorinating Bacteria** 

Dehalococcoides spp (1)

DHC

3.95E+03

Dehalobacter spp.

DHB

5.66E+06

**Functional Genes** 

BAV1 VC R-Dase (1)

BVC

<9.03E+02

Legend:

NA = Not Analyzed NS = Not Sampled J = Estimated gene copies below PQL but above LQL

I = Inhibited

< = Result not detected

Notes:

1 Bio-Dechlor Census technology was developed by Dr. Loeffler and colleagues at Georgia Institute of Technology and was licensed for use through Regenesis.

2340 Stock Creek Blvd. Rockford, TN 37853-3044

Tel: (865) 573-8188; Fax: (865) 573-8133

**PLFA** 

Client:

**CH2M HILL** 

Project:

NAB Little Creek Site 11 (CTO-103)

**MI Project Number:** 

Date Received:

**013CJ** 10/08/2005

**Sample Information** 

Sample Name:

LS11-SB801-LC

LS11-SB801-YC

LS11-SB806-

LS11-SB805-L

Sample Date:

10/07/2005 Soil

24.05

10/07/2005

LC 10/07/2005

10/08/2005

D: .

Sample Matrix:

Soil

17.47

25.25

8.33

2.33

37.24

10/07/2005 Soil

2.51E+07

20.73

36.24

3.85

5.30

Soil

**Biomass** 

Cells/g

6.01E+06 2.18E+06

\_\_\_\_

Soil

2.02E+06

13.86

24.51

8.49

3.78

42.38

6.99

Community Structure (% total PLFA)

Firmicutes (TerBrSats)
Proteobacteria (Monos)

Proteobacteria (Monos) 30.54

Anaerobic metal reducers (BrMonos) 4.90

SRB/Actinomycetes (MidBrSats) 1.50

SRB/Actinomycetes (MidBrSats) 1.50
General (Nsats) 35.12

General (Nsats)
Eukaryotes (polyenoics)

3.91 9.38

23.42 10.47

Physiological Status (Proteobacteria only)

Slowed Growth

Decreased Permeability

1.48 0.18

1.55 0.18

Legend:

NA = Not Analyzed

NS = Not Sampled

**PLFA** 

2340 Stock Creek Blvd. Rockford, TN 37853-3044

Tel: (865) 573-8188; Fax: (865) 573-8133

Client:

**CH2M HILL** 

Project: NAB Little Creek Site 11 (CTO-103) **MI Project Number:** 

013CJ

Date Received:

10/08/2005

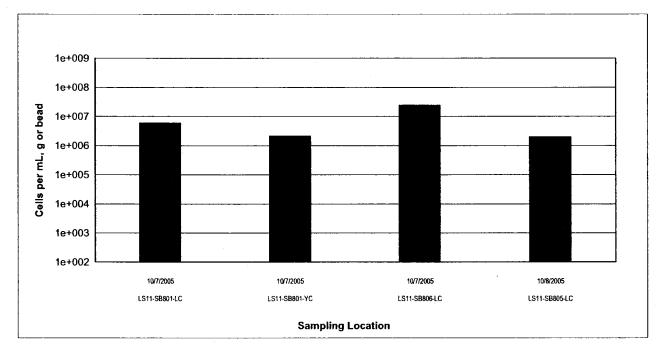


Figure 1. Biomass content is presented as a cell equivalent based on the total amount of phospholipid fatty acids (PLFA) extracted from a given sample. Total biomass is calculated based upon PLFA attributed to bacterial and eukaryotic biomass (associated with higher organisms).

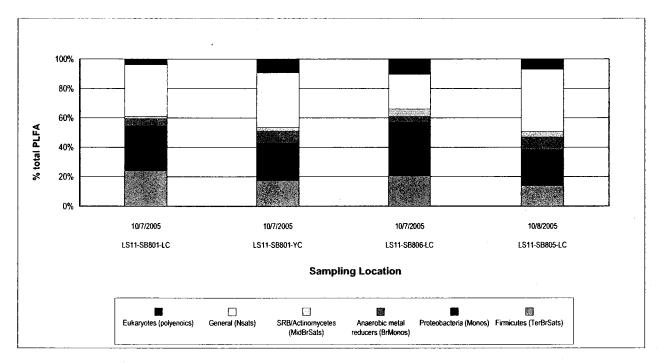


Figure 2. Relative percentages of total PLFA structural groups in the samples analyzed. Structural groups are assigned according to PLFA chemical structure, which is related to fatty acid biosynthesis.



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### **Analysis Summary Report**

	Date	Date	Arrival	Arrival Metabolic Acids (mg/L)					
Sample Name:	Sampled:	Received:	Condition:	Pyruvic	Lactic	Formic	Acetic	Propionic	Butyric
LS11-GW804-LC	10/07/2005	10/08/2005	Intact	<4	<1	<1	<1	. <1	<1
LS11-MW10D-05D	10/10/2005	10/12/2005	Intact	<4	<1	<1	<1	<1	<1
LS11-MW09D-05D	10/10/2005	10/12/2005	Intact	<4	<1	<1	<1	<1	<1
LS11-MW25D-05D	10/11/2005	10/12/2005	Intact	<4	<1	<1	269.5	192.4	13.8



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### **Quality Control Report**

Compound	MS Recovery	MSD Recovery	RPD	LCS Recovery	
	%	%	%	%	
Pyruvic	86.9	85.1	2.1	86.5	
Lactic	90.7	91.5	0.9	93.9	
Formic	56.9	55.2	3.0	61.0	
Acetic	89.8	89.4	0.4	90.1	
Propionic	91.3	91.0	0.3	89.1	
Butyric	83.1	81.8	1.6	83.5	

### LABORATORY DATA SUMMARY

Project:

CH2M Hill Project Number:

CH2M Hill Laboratory Testing 2005 329753.SI.FQ, CTO # 103 Navy Clean Prime III Contract N62470-02-D-4401

Number: Date:

3687-110

10/27/05

SAMPLE NUMBER	LS11-SB805-LC	LS11-SB805-YC	LS11-SB806-LC	LS11-SB806-YC
SAMPLE DEPTH	16 to 20	22 to 26	16 to 20	20 to 24
SAMPLE CLASSIFICATION	SM	ML	SP	СН
MOISTURE CONTENT (%)	19.6	43.0	18.5	55.9
% FINER THAN NO. 200 SIEVE	12:8	83.4	4.8	90.5
SPECIFIC GRAVITY	2.651	2.699	2.860	2.722
WET UNIT WEIGHT (pcf)	108.9	115.6	121.4	104,7
DRY UNIT WEIGHT (pcf)	91.1	80.8	102.5	67.2
BULK DENSITY (g/ml)	1.859	1.791	1,874	1.754
POROSITY (%)	49.4	52.0	38.0	60.6

### **SIEVE ANALYSIS**

Project Name:

**CH2M HILL Laboratory Testing 2005** 

Number:

3687-110

Project Number:

329752.SI.FQ, CTO #: 103 and "Navy Clean III Prime Contract N62470-02-D-4401"

Sample Number:

LS11-SB806-YC

Sample Depth:

20 to 24 feet

Sample Description: Silty CLAY (CH), Dark Gray, Trace Fine Sand, Shell Fragments and Organics

Test Method:

**ASTM D 422** 

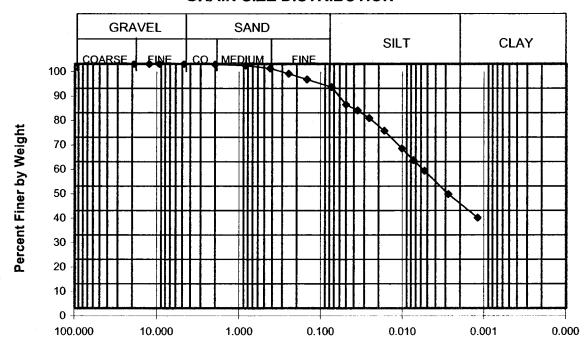
### Sieve Analysis Data

SIEVE NO.	PERCENT PASSING
3/4 Inch	100.0
1/2 Inch	100.0
3/8 Inch	100.0
4	100.0
10	99.8
20	99.3
40	98.2
60	96.0
100	93.6
200	90.5

### **Hydrometer Analysis Data**

DIAMETER (mm)	PERCENT FINER
0.0502	83.4
0.0361	81.0
0.0261	77.8
0.0171	72.7
0.0103	65.5
0.0075	60.6
0.0054	56.3
0.0028	46.7
0.0012	37.0

### **GRAIN SIZE DISTRIBUTION**



Grain Size (mm)

### **SIEVE ANALYSIS**

Project Name:

**CH2M HILL Laboratory Testing 2005** 

Number:

3687-110

Project Number:

329752.SI.FQ, CTO #: 103 and "Navy Clean III Prime Contract N62470-02-D-4401"

Sample Number:

LS11-SB806-LC

Sample Depth:

16 to 20 feet

Sample Description: SAND (SP), Tan and Olive Gray, Fine to Medium, Trace Silt

Test Method:

**ASTM D 422** 

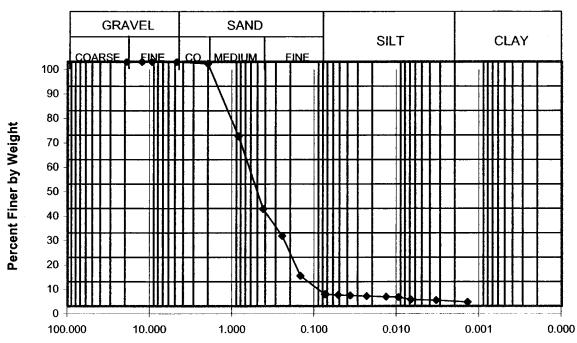
### Sieve Analysis Data

# **Hydrometer Analysis Data**

SIEVE NO.	PERCENT PASSING	
3/4 Inch	100.0	
1/2 Inch	100.0	
3/8 Inch	100.0	
4	100.0	
10	99.2	
20	69.5	
40	39.8	
60	28.8	
100	12.4	
200	4.8	

DIAMETER (mm)	PERCENT FINER
0.0735	4.8
0.0520	4.6
0.0368	4.4
0.0233	4.1
0.0135	3.9
0.0095	3.6
0.0068	2.7
0.0033	2.4
0.0014	1.7

### **GRAIN SIZE DISTRIBUTION**



Grain Size (mm)

### **SIEVE ANALYSIS**

Project Name:

**CH2M HILL Laboratory Testing 2005** 

Number:

3687-110

Project Number:

329752.SI.FQ, CTO #: 103 and "Navy Clean III Prime Contract N62470-02-D-4401"

Sample Number:

LS11-SB805-YC

Sample Depth:

22 to 26 feet

Sample Description:

Sandy SILT (ML), Dark Gray, with Clay, Trace Organics

Test Method:

**ASTM D 422** 

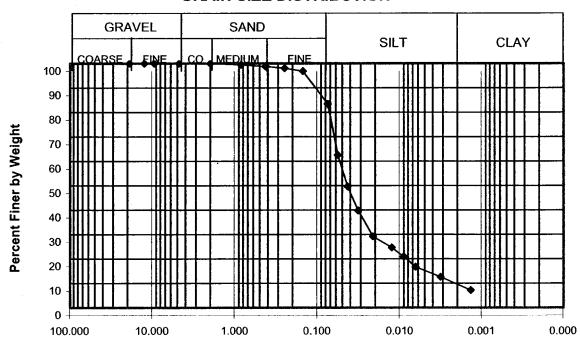
### **Sieve Analysis Data**

SIEVE NO.	PERCENT PASSING
3/4 Inch	100.0
1/2 Inch	100.0
3/8 Inch	100.0
4	100.0
10	100.0
20	99.5
40	98.8
60	98.1
100	96.9
200	83.4

### **Hydrometer Analysis Data**

DIAMETER (mm)	PERCENT FINER
0.0567	62.6
0.0432	49.7
0.0321	39.8
0.0213	29.2
0.0125	24.7
0.0090	20.9
0.0065	16.7
0.0032	12.6
0.0014	7.2

### **GRAIN SIZE DISTRIBUTION**



Grain Size (mm)

### **SIEVE ANALYSIS**

Project Name:

**CH2M HILL Laboratory Testing 2005** 

Number:

3687-110

Project Number:

329752.SI.FQ, CTO #: 103 and "Navy Clean III Prime Contract N62470-02-D-4401"

Sample Number:

LS11-SB805-LC

Sample Depth:

16 to 20 feet

Sample Description: Silty SAND (SM), Tan-Orange, Fine to Medium, Trace Clay

Test Method:

**ASTM D 422** 

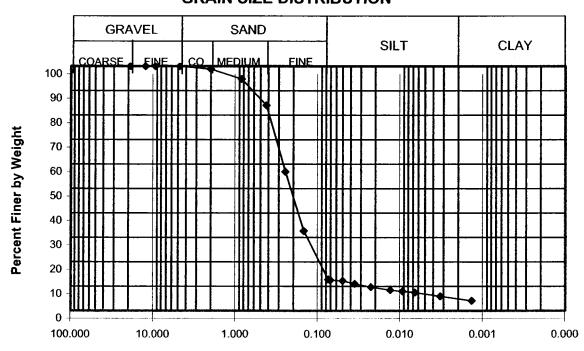
### Sieve Analysis Data

### **Hydrometer Analysis Data**

SIEVE NO.	PERCENT PASSING
3/4 Inch	100.0
1/2 Inch	100.0
3/8 Inch	100.0
4	100.0
10	98.8
20	94.8
40	84.0
60	57.0
100	32.7
200	12.8

DIAMETER (mm)	PERCENT FINER
0.0708	12.6
0.0501	12.3
0.0357	11.0
0.0227	9.7
0.0132	8.5
0.0094	8.0
0.0066	7.5
0.0033	5.9
0.0014	4.1

### **GRAIN SIZE DISTRIBUTION**



Grain Size (mm)

Appendix C PRG Calculations

# Appendix C Recommended Preliminary Remediation Goals Groundwater Residential Scenario Site 11, NAB Little Creek

Chemical	Recommended PRG (mg/L)	Basis
VOCs		
1,1-Dichloroethane	2.9E+00	Child, HQ = 1

Child scenario selected for noncarcinogenic PRGs since child scenario more conservative (lower PRGs). For constituents with basis of CR = 10<sup>-5</sup>, PRG for CR =10<sup>-5</sup> less than PRG for applicable HQ. Used CR of 10<sup>-5</sup> to keep overall carcinogenic risk below 10<sup>-4</sup>. Applicable HQ chosen to keep total HI for each target organ below 1.

filename: Appendix C PRC Calculation.XLS

worksheet: sumGWresPRG

Appendix D
Applicable or Relevant and Appropriate
Requirements

### **Acronyms and Abbreviations**

ARAR	Applicable or relevant and appropriate requirement	POTW	Publicly Owned Treatment Works
BTAG	Biological Technical Assistance Group	ppm	Parts per Million
CERCLA	Comprehensive Environmental Response, Compensation and Liability Act	RAO	Remedial Action Objective
CFC	Chlorofluorocarbon	RBC	Risk-Based Concentrations
CFR	Code of Federal Regulations	RCRA	Resource Conservation and Recovery Act
DCR	Virginia Department of Conservation and Recreation	SDWA	Safe Drinking Water Act
DNH	Division of Natural Heritage	SMCL	Secondary Maximum Contaminant Level
IDW	Investigation Derived Waste	TCLP	Toxicity Characteristic Leaching Procedure
MCL	Maximum Contaminant Level	TSCA	Toxic Substance Control Act
MCLG	Maximum Contaminant Level Goal	UIÇ	Underground Injection Control
NAAQS	National Ambient Air Quality Standards	USACE	US Army Corps of Engineers
NESHAPs	National Emission Standards for Hazardous Air Pollutants	USC	United States Code
NPDES	National Pollutant Discharge Elimination System	USEPA	United States Environmental Protection Agency
NSDWRs	National Secondary Drinking Water Regulations	UU/UE	Unlimited Use/Unrestricted Exposure
NSPS	New Source Performance Standards	VAC	Virginia Administrative Code
OSWER	Office of Solid Waste and Emergency Response	VMRC	Virginia Marine Resource Commission
PCB	Polychlorinated biphenyls	VPA	Virginia Pollutant Abatement
PMCL	Primary Maximum Contaminant Level	VPDES	Virginia Pollutant Discharge Elimination System

### References

Commonwealth of Virginia, 2004. Preliminary Identification, Applicable or Relevant and Appropriate Requirements.

USEPA, 1998. CERCLA Compliance with Other Laws Manual: Interim Final. Office of Emergency and Remedial Response. EPA/540/G-89/006.

USEPA, 1998. CERCLA Compliance with Other Laws Manual: Part II. Clean Air Act and Other Environmental Statutes. Office of Emergency and Remedial Response.

USEPA, 1998. RCRA, Superfund & EPCRA Hotline Training Manual. Introduction to Applicable or Relevant and Appropriate Requirements. EPA540-R-98-020.

## Table D-1 Federal Chemical-Specific ARARs Site 11 Feasibility Study NAB Little Creek, Virginia Beach, Virginia

Perceptible   Precipite   Citation   Alternative   ARAR   Comment   Commen	
Air NAAOS specify the maximum concentration of each criteria pollutant (carbon monoxide, lead, nthrogen dioxide, particulate matter, ozone, suffur dioxide) which is to be permitted in the ambient air, as averaged over a period of time. Requirements differ for new sources of air pollutant emissions and existing sources. Requirements delianment, non-ait animent, unclassified, or transport) (see Federal Location-Specific ARARs).  Air NESHAPS are point-source standards address both new and existing sources at the point of emission. Eight hazardous air pollutants of emission. Eight hazardous air pollutants (absostos, benzene, benjim, coke over emissions. Inorganic rasene), memory, addicuncities, and virty chloride) were initially designated. The 1980 amendments greatly expanded the field of hazardous air pollutants.  Emissions of hazardous air pollutants (absostos, benzene, benjim, pollutants and designating 174 source categories. Maximum Achievable Control Technology standards were developed for all source categories that emit hazardous air pollutants.  Emissions of hazardous air pollutants (absostos, benzene, benjim, pollutants and report of criteria pollutants and ring the pollutants and designating 174 source categories. Maximum Achievable Control Technology standards were developed for all source categories that emit hazardous air pollutants.  Emissions of hazardous air pollutants (absostos, benzene, benjim, pollutants and designating 174 source categories. Maximum Achievable Control Technology standards were developed for all source categories that emit hazardous air pollutants.  Emissions of riteria pollutants from a pollutants from a pollutants and designating 174 source categories. Maximum Achievable Control Technology standards were developed for all source categories in that emit hazardous air pollutants.  Emissions of riteria pollutants from a pollutants from a pollutants from a pollutants from a pollutants from a pollutants from a pollutants from a pollutants from a pollutants from a pollutants from a p	
Air NAAOS specify the maximum concentration of each criteria pollutant (carbon monoxide, lead, nitrogen dioxide, particulate matter, ozone, suffur dioxide) which is to be permitted in the ambient air, as averaged over a period of time. Requirements differ for new sources of air pollutant emissions and existing sources. Requirements else offer beased however, the application and on the sire should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the site should be suffered to the suffered to t	100
on the air quality designation of the site's location (i.e., attainment, uncleasafied, or transport) (see Federal Location-Specific ARARs).  Air NESHAPS are point-source standards for hazardous air poliutants. These standards address both new and existing sources at the point of emission. Eight hazardous air poliutants are point source, benytium, coke oven emissions, inorganic arsenic, mercury, radionuclides, and vinyt chloride) were initially designated. The 1990 amendments greatly expanded the field of hazardous air poliutants and designating 174 source categories. Maximum Achievable Control Technology standards were developed for all source categories that emit hazardous air poliutants.  Seie Drinking Water/School Special Control Technology standards were developed for all source categories of the standards and selegiating 174 source categories. Maximum Achievable Control Technology standards were developed for all source categories of the standards and selegiating 174 source categories. Maximum Achievable Control Technology standards were developed for all source categories of the standards serve to protect public water systems. Primary drinking water standards consist of federally enforceable MCLs. MCLs are the highest level of a contaminant that is allowed in drinking water.	
These standards address both new and existing sources at the point of emission. Eight hazardous air pollutaris (asbestos, benzene, benyilium, coke oven emissions, Inorganic arsenic, mercury, radionuclidas, and vinyl chloride) were initially designated. The 1990 amendments greatly expanded the list of hazardous air pollutarits, including 189 new pollutarits and designating 174 source categories. Maximum Achievable Control Technology standards were developed for all source categories that emit hazardous air pollutarits.  Sele Diminist Wateries  Groundwater  Sele Diminist Wateries  Solve standards serve to protect public water systems. Primary dirinking water standards serve to protect public water systems. Primary dirinking water standards consist of federally enforcesible MCLs. MCLs 15 service connections or serve at least 25 year— in the highest level of a contaminant that is allowed in dirinking water.	
pollutarité and designating 174 source catégories. Maximum Achievable Control Technology standards were developed for all source catégories that emit hazardous air pollutarit.  Sale Drinking Wall  Groundwater Groundwater SDWA standards serve to protect public water systems. Primary drinking water standards consist of federally enforceable MCLs. MCLs are the highest level of a contaminant that is allowed in drinking water.  In pact to public water systems that have at least 40 CFR 15 service connections or serve at least 25 year.  In pact to public water systems that have at least 40 CFR 16 service connections or serve at least 25 year.  In pact to public water systems that have at least 40 CFR 16 service connections or serve at least 25 year.  In 141.16 and for on-site ground or surface waters that are 141.61 to	
Groundwater SDWA standards serve to protect public water systems. Primary dirikking water standards consist of federally enforceable MCLs. MCLs are the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In the highest level of a contaminant that is allowed in drinking water. In	implementation of ERH will be treated.
drinking water standards consist of federally enforceable MCLs. MCLs are the highest level of a contaminant that is allowed in drinking water.  15 service connections or serve at least 25 year. 141.11 to MCLs. However, the aquifer is not current residents. May also be cleanup standards 141.16 and future to be used as a potable water supplied ron-site ground or surface waters that are 141.61 to MCLs. However, the aquifer is not current residents. May also be cleanup as the first of tuture to be used as a potable water supplied to ron-site ground or surface waters that are 141.61 to MCLs. However, the aquifer is not current residents.	
	ntly, nor reasonably anticipated in the
current or potential sources of drinking water. 141.66 3 - ERH & ERU Applicable This remedial acquirer is not current future to be used as a potable water supp	
Groundwater SDWA standards serve to protect public water systems. The MCLG is Impact to public water systems that have at least 40 CFR 2 - ERD TBC Although MCLGs are non-enforceable stated the level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety connections or serve at least 25 year. 141.50 to implemented with a target goal of achieving count residents. May also be cleanup standards 141.55	ving MCLs.
and are non-enforceable public health goals. for on-elte ground or surface waters that are current or potential sources of drinking water.  3 - ERH & ERD TBC Atthough MCLGs are non-enforceable stated current or potential sources of drinking water.	ving MCLs.
Groundwater National Secondary Drinking Water Regulations (NSDWRs or secondary Impact to public water systems that have at least 40 CFR 143 2 - ERD TBC Although secondary MCLs are non-enforce standards) are non-enforceable guidelines regulating contaminants that may cause cosmetic effects (such as skin or tooth discoloration) or cound residents. May also be cleanup standards is being implemented with a target goal or cound residents. May also be cleanup standards	of achieving MCLs.
aesthetic effects (such as taste, odor, or color) in drinking water.  for on-site ground or surface waters that are current or potential sources of drinking water.  3 - ERH & ERD TBC  Although secondary MCLs are non-enforce in being implemented with a target goal or surface.	
Water, air, fish Chemical concentrations corresponding to fixed levels of human health. Assessment of potential human health risks. USEPA 2 - ERD TBC The RAO is to reduce concentrations in g practicable with a goal of achieving UU/U practicable with a goal of achieving UU/U.  RBC Tables TBC Tables TBC Tables Region III practicable with a goal of achieving UU/U.	UE for the site. It is not anticipated that
3 - ERH & ERD TBC The FAO is to reduce concentrations in g practicable with a goal of achieving UU/U RBCs will be used to establish UU/UE.	
USEPA Region IN BITAG Screening Veture	Property of the Control of the Contr
Soll, sediment, Chemical concentrations corresponding to fixed levels of risks to surface weter (sold and or surface weter surface weter (sold and or surface).  Assessment of potential ecological risks.  USEPA 2 - ERD TBC There are no unacceptable ecological risk screening values do not apply.  BTAG BTAG	
Screening 3 - ERH & ERD TBC There are no unacceptable ecological risk Values	

### Table D-2 Virginia Chemical-Specific ARARs Site 11 Feasibility Study

	NAB Little Creek, Virginia Beach, Virginia								
Media	Requirement	Prerequisite	Citation		ARAR Determination	Comment			
Groundwater	Establishes groundwater quality standards to protect the public health or welfare and enhance the quality of water.	Standards are used when no MCL is available.	Groundwater Quality Standards , 9 VAC 25-280	2 - ERD 3 - ERH & ERD	Applicable	This remedial action is being completed to address concentrations in groundwater.  This remedial action is being completed to			
⊟avilentačniciš				3 - ERH & ERD	Applicable	address concentrations in groundwater.			
Groundwater	Ensures that all water supplies destined for public consumption be pure water. Cleanup levels for potential drinking water sources must be based on PMCLs. In the absence of PMCLs, other health-based standards or criteria, or best professional judgment based on risk assessment, may be employed. Where groundwater that is a potential drinking water source discharges to surface water, the cleanup level		Waterworks Regulations , 12 VAC 5-590-10 to 1280	2 - ERD	Relevant and Appropriate	This ARAR is not applicable because the aquifer is not currently, nor reasonably anticipated in the future to be used as a potable water supply. This remedial action is being implemented with a target goal of achieving MCLs.			
	at the discharge point would be the more stringent of either the PMCL or a discharge limit based on the Water Quality Standards.			3 - ERH & ERD	Relevant and Appropriate	This ARAR is not applicable because the aquifer is not currently, nor reasonably anticipated in the future to be used as a potable water supply. This remedial action is being implemented with a target goal of achieving MCLs.			
Groundwater	SMCLs are guidelines pertaining to aesthetic qualities of drinking water (i.e., color, odor, and taste).	Potential drinking water source.	Waterworks Regulations, 12 VAC 5-590-10 to 1280	2 - ERD	TBC	The aquifer is not currently, nor reasonably anticipated in the future to be used as a potable water supply. Therefore this criteria does not apply.			
				3 - ERH & ERD	ТВС	The aquifer is not currently, nor reasonably anticipated in the future to be used as a potable water supply. Therefore this criteria does not apply.			

### Table D-3 Federal Location-Specific ARARs Site 11 Feasibility Study NAB Little Creek, Virginia Beach, Virginia

TOTAL Entire O'CON, Triginia Seast, Triginia								
Location	Requirement	Prerequisite	Citation		ARAR Determination	Comment		
Embary CE			Marie de periode de la Serie					
(except for ozone) "best avai poliutant t associate allowable secondary increase o	New major stationary sources/major modifications shall apply "best available control technology" for each regulated pollutant having a potential to emit greater than the associated"significant emission rate." Demonstration that allowable emissions increases or reductions (including secondary emissions) will not cause a significant emissions increase over baseline emissions or contribute to a violation of the NAAQS.	Major stationary sources that emits, or has the potential to emit, 100 tons per year or more of any regulated pollutant; any other stationary source that emits, or has the potential to emit, 250 tons per year or more of any regulated pollutant.	40 CFR 52.21(j)	2 - ERD	Not Applicable	This remedial action does not involve a major new or modified source of regulated air pollutants.		
				3 - ERH & ERD	Relevant and Appropriate	This remedial action will include discharge from an air stripping system designed to treat vapors including regulated air pollutants. However, it is not anticipated that the system will be a major source.		
Non-attainment area (ozone)		Clean Air Act , Part D §173(1) to (3); 40 CFR 51.18(j)	2 - ERD	Not Applicable	This remedial action does not involve a major new or modified source of regulated air pollutants.			
				3 - ERH & ERD	Relevant and Appropriate	This remedial action will include discharge from an air stripping system designed to treat vapors including regulated air pollutants. However, it is not anticipated that the system will be a major source.		

### Table D-4

### Virginia Location-Specific ARARs Site 11 Feasibility Study

Location	Requirement	Prerequisite	Citation	Alternative	ARAR Determinatio	Comment
elindenpertie (sp) i	Záservetlőn égő szántszásána szántás a tillága talája tösété	Commence Commence	6.62 - 2.7 <b>6</b> 26.96			
and its tributaries	Criteria that provide for the protection of water quality of the Chesapeake Bay and its tributaries, that will also accommodate economic development in Tidewater Virginia. Under these requirements, certain locally designated tidal and nontidal wetlands, as well as other sensitive land areas, may be subject to limitations regarding land-disturbing activities, removal of vegetation, use of impervious cover, erosion and sediment control, stormwater management, and other aspects of land use that may have effects on water quality.	Location is within a Chesapeake Bay Preservation Area.	Chesapeake Bay Preservation Area Designation and Management Regulations . 9 VAC 10-20-10 to 260	2 - ERD 3 - ERH & ERD	Applicable	Site 11 is located within the Chesapeake Bay watershed. However, the remedy will not involve or effect tributaries of the Chesapeake Bay.  Site 11 is located within the Chesapeake Bay watershed. However, the remedy will not involve or effect tributaries of the Chesapeake Bay.

### Table D-5 Federal Action-Specific ARARs Site 11 Feasibility Study NAB Little Creek, Virginia Beach, Virginia

Action	Requirement	Duanaminita	C:4-4:	A 14 41	IADAD	0
Action	Requirement	Prerequisite	Citation	Alternative	JAKAK	Comment
					Determination	
Retalline Conselo		A CONTRACTOR OF STREET	Listing and the second			Committee of the Commit
		Off-site disposal of hazardous wastes.	40 CFR 240 to 282	2 - ERD	Relevant and	This remedy will involve offsite disposal of IDW
	for further management. Administrative RCRA standards				Appropriate	However, based on site history, the IDW is not
	include the obligation to obtain permits and keep various			_		anticipated to be hazardous. IDW generated
	records at all hazardous waste treatment, storage, and				1	during the implementation of this remedial action
	disposal facilities; and the requirement to include a hazardous					will be characterized prior to disposal.
	waste manifest when sending hazardous wastes off-site.					
			1	3 - ERH & ERD	Relevant and	This remedy will involve offsite disposal of IDW.
				i	Appropriate	However, based on site history, the IDW is not
					l .	anticipated to be hazardous. IDW generated
						during the implementation of this remedial action
						will be characterized prior to disposal.
ĺ				1		
		L		ĺ	1	

#### Table D-6 Virginia Action-Specific ARARs Site 11 Feasibillty Study NAB Little Creek, Virginia Beach, Virginia

	NAB Little Creek, Virginia Beach, Virginia								
Action	Requirement	Prerequisite	Citation	Alternative	ARAR Determination	Comment			
Air emissions from disturbance Standards for visible emissions, fugitive dust/emissions, of soil, freatment of soil or water, or other pollutant management activities		Source of visible emissions, fugitive dust/emissions, and/or a stationary source that emits or may emit any toxic pollutant.	Standards of Performance for Visible Emissions and Fugitive Dust/Emissions [Rule 5-1], 9 VAC 5-50-60 to 120; USEPA National Emission Standards for Hazardous Air Pollutants [Rule 6-1], 9 VAC 5-60-60 to 80;	2 - ERD	Not Applicable	This remedial action does not involve discharges to air.  This remedial action will include discharge from			
	Emission Standards for Toxic Pollutants		from New and Modified Sources [Rule 6-5], 9 VAC 5-50-60-300 to 370			an air stripping system.			
Handling, storage, treatment,	Provides for the control of all hazardous wastes that are generated within, or transported to, the Commonwealth for the purposes of storage, treatment, or disposal or for the purposes of resource conservation or recovery. Any disposal facility must be properly permitted and in compliance with all operational and monitoring requirements of the permit and regulations.	Management of wastes that meet the definition of hazardous waste.	Hazardous Waste Regulations, 9 VAC 20-60-12 to 1505; Regulations Governing the Transportation of Hazardous Materials, 9 VAC 20-110-10 to 130	2 - ERD	Relevant and Appropriate	This remedy will generate soil and water IDW which will be characterized for off site disposal. Based on site history, it is not anticipated that IDW will be characterized as hazardous waste.			
:				3 - ERH & ERD	Relevant and Appropriate	This remedy will generate soil and water IDW which will be characterized for off site disposal. Based on sile history, it is not anticipated that IDW will be characterized as hazardous waste.			
Handling, storage, treatment, disposal, and/or transportation of solid waste IDW		Management of wastes that meet the definition of solid waste.	Solid Waste Management Regulations, 9 VAC 20-80-10 to 790	2 - ERD	Applicable	This remedy will generate soil and water IDW which will be characterized for off site disposal.			
	requirements for undertaking corrective actions at solid waste management facilities. Any disposal facility must be properly permitted and in compliance with all operational and monitoring requirements of the permit and regulations.			3 - ERH & ERD	Applicable	This remedy will generate soil and water IDW which will be characterized for off site disposal.			

Appendix E Preliminary Cost Estimate

COMPARISON OF TOTAL COST OF REMEDIAL ALTERNATIVES								
Site: Location; Phase: Base Year: Date:		NAB Little Creek Site 11 Feasibility Study (-3 2006 January 20, 2005	30% to +50%;			DISCOUNT RATE		
Alternative		No A		989.50 <b>.60</b> 488.50 <b>.60</b>	2 RD. r	ERH	LERD	
Target	NA		A		-Source -Plume	ERH - Source ERD - Pollahing Source ERD - Plume		
Approach	·	N.	A	ERD treatment of (~18 - :	target depth inten 23 ft bgs)r	val ERH/ERD treatment of target de interval (~18 - 23 ft bgs)		
	Number of Years	Cost per Year	Total Cost	Cost per Year	Total Cost	Cost per Year	Total Cost	
Capital Cost YEAR 0	1	0	0	\$499,000	\$499,000	\$1,047,000	\$1,047,000	
Annual Cost YEAR 1-7	7	0	0	\$167,367	\$1,171,571	Aller 🕳 North		
Annual Cost YEAR 8-14	7	0	0	\$135,928	\$951,497	- 1		
Annual Cost YEAR 1-3	3	0	0	-	9.0	\$167,367	\$502,102	
Annual Cost YEAR 4-14	11	0	0	<del>73</del> (45)		\$135,928	\$1,495,209	
Long Term Cost YEAR 15-30	16	0	0	\$19,090	\$305,440 -	\$19,090	\$305,440	
(Long Term Monitoring) Periodic Cost YEAR 0-30	6	0	0	- \$6,900 T	\$41,400	\$6,900	\$41,400	
(5-Year Reviews) TOTAL COST			\$0		\$2,968,908		\$3,391,151	
PRESENT VALUE COST	-			1		1922		
Total Net Present Value (Discou	nt rate 3.1%)		0		\$2,399,000		\$2,841,000	

Disclalmer: This estimate is an Order of Magnitude cost estimate, suitable for use in project evaluation and planning. This estimate has been prepared without equipmer specifications, layout, design or engineering calculations. Expected level of accuracy is +50% / -30%. Actual construction costs will vary from this estimate due to market conditions, actual costs of purchased materials, quantity variations, regulatory requirements, final design details and other project-specific factors existing at the time of construction.

### Comparison of -30% +50%

Site:

NAB Little Creek

Location:

Site 11

Phase:

Feasibility Study (-30% to +50%)

Base Year:

2006

Date:

January 20, 2005

Alternative		1	2 100	3
		No Action	ERD	ERH & ERD
Capital Cost		\$0	\$499,000	\$1,047,000
Range of Estimate	-30%	\$0	\$349,300	\$732,900
	+50%	\$0	\$748,500	\$1,570,500
Total Net Present Valu	ue (Discount rate 3.1%)	\$0	\$2,399,000	\$2,841,000
Range of Estimate	-30%	\$0	\$1,679,300	\$1,988,700
	+50%	\$0	\$3,598,500	\$4,261,500

Alternative 2: Element:	Enhanced Reductive Dechloring Overall System Components	ation Using Lact	ate				
Site: Location: Phase: Base Year: Date:	NAB Little Creek Site 11 Feasibifity Study (-30% to +50%) 2006 January 20, 2005	ve.					
CAPITAL	COSTS DESCRIPTION	ory	UNIT	UNIT.	IOTAL:	NOTES	
	PRE-INJECTION ACTIVITIES YEAR 0 LACTATE INJECTION REPORTING (INJECTION & SAMPLING) SAMPLING LAND USE CONTROLS SUBTOTAL				\$137,925 \$142,888 \$24,869 \$107,800 \$20,000 \$433,482	Engineer's Estimate Recent Similar Project Engineer's Estimate Engineer's Estimate Recent Similar Project	
	CONTINGENCY	15%	of	\$433,482	\$65,022		
	TOTAL CAPITAL COST				\$499,000		
ÄNNUÄL	OPERATING COSTS YEAR 1.7	gry	UNIT	UNIT COST	TOTAL	NOTES	A
	LACTATE INJECTION REPORTING (INJECTION & SAMPLING) SAMPLING LAND USE CONTROLS SUBTOTAL			4 ·	\$81,768 \$24,869 \$33,900 \$5,000 \$145,537	Recent Similar Project Engineer's Estimate Engineer's Estimate Recent Similar Project	E S
	CONTINGENCY	15%	of	\$145,537	\$21,831	* * * * * * * * * * * * * * * * * * *	
	TOTAL ANNUAL OPERATING COST				\$167,367	•	:
ANNUAL	OPERATING COSTS YEAR 8-14	ату	UNIT	UNIT COST	TOTAL.	NOTES ***	
r.	LACTATE INJECTION REPORTING SAMPLING LAND USE CONTROLS SUBTOTAL				\$54,430 \$24,869 \$33,900 \$5,000 \$118,198	Recent Similar Project Engineer's Estimate Engineer's Estimate Recent Similar Project	
	CONTINGENCY	15%	of	\$118,198	\$17,730	Ŷ	¢.
	TOTAL ANNUAL OPERATING COST				\$135,928	4	
LONG TE	RM MONITORING YEAR 15-30 DESCRIPTION	ατγ	UNIT	UNIT *	TOTAL	NOTES	
	SAMPLING LAND USE CONTROLS SUBTOTAL	1	EA EA	\$11,600 \$5,000	\$11,600 \$5,000 \$16,600	Engineer's Estimate Recent Similar Project	
	CONTINGENCY	15%	of	\$16,600	\$2,490		ŝ
	TOTAL ANNUAL OPERATING COST			***	\$19,090		
PERIODIO	COST	ρ	UNIT	UNIT COST	TOTAL	NOTES	
	5 YEAR REVIEWS			· .	\$6,000	Recent Similar Project	
	CONTINGENCY	15%	of	\$6,000	\$900		: :
	TOTAL PERIODIC OPERATING COST			·	\$6,900	Karana ya karana ya karana ya karana ya karana ya karana ya karana ya karana ya karana ya karana ya karana ya	
PRESENT	VALUE ANALYSIS	·				Based on 3.1% D	iscount Rate

Note: This estimate is an Order of Magnitude cost estimate, suitable for use in project evaluation and planning. This estimate has been prepared without equipment specifications, layout, design or engineering calculations. Expected level of accuracy is +50% / -30%. Actual construction costs will vary from this estimate due to market conditions, actual costs of purchased materials, quantity variations, regulatory requirements, final design details and other project-specific factors existing at the time of construction.

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Alternative 3: Element:	Electrical Resistance Heating and Overall System Components	Enhanced Redu	ctive Deci	lorination Using	Lactate	
Site:	NAB Little Creek					
Location:	Site 11					
Phase: Base Year:	Feasibility Study (-30% to +50%) 2006					
Date:	January 20, 2005					•
GAPITAL (	OSTS					
4				UNIT		
	DESCRIPTION	QTY	UNIT	COST	TOTAL	NOTES: 100
	ERH SYSTEM & PRE-INJECTION ACTIVITIES YEAR 0 LACTATE INJECTION				\$603,225	Vendor & Engineer's Estimate
	REPORTING				\$135,086 \$24,869	Recent Similar Project Engineer's Estimate
	SAMPLING LAND USE CONTROLS				\$127,100 \$20,000	Engineer's Estimate Recent Similar Project
	SUBTOTAL				\$910,279	recoun comment reject
	CONTINGENCY	15%	of	\$910,279	\$136,542	
	TOTAL CAPITAL COST			<u> </u>	\$1,047,000	
					<del>• • • • • • • • • • • • • • • • • • • </del>	
ANNUAL C	PERATING COSTS YEAR 1-3				1	
	DESCRIPTION	QTY	UNIT	UNIT	TOTAL	NOTES
			W1411			
ļ	LACTATE INJECTION REPORTING	,			\$81,768 \$24,869	Recent Similar Project Engineer's Estimate
1	SAMPLING				\$33,900	Engineer's Estimate
}	LAND USE CONTROLS SUBTOTAL				\$5,000 \$145,537	Recent Similar Project
	CONTINGENCY	15%	of	\$145,537	\$21,831	
	TOTAL ANNUAL OPERATING COST	1370	O.	<b>4140,337</b>	\$167,367	
	TOTAL ANNUAL OF ERATING COST				¥161,301	
ANNUAL C	PERATING COSTS YEAR 4-14					
	DESCRIPTION	QTY	UNIT	UNIT	TOTAL	NOTES
	LACTATE INJECTION				\$54,430	Recent Similar Project
	REPORTING				\$24,869	Engineer's Estimate
ŀ	SAMPLING LAND USE CONTROLS				\$33,900 \$5,000	Engineer's Estimate Recent Similar Project
	SUBTOTAL			-	\$118,198	Noon on an indicate the control of t
	CONTINGENCY	15%	of	\$118,198	\$17,730	
	TOTAL ANNUAL OPERATING COST				\$135,928	
LONG TER	M MONITORING YEAR 15-30					
	DESCRIPTION	QTY	UNIT	COST	TOTAL	NOTES
	SAMPLING			,	\$11,600	Engineer's Estimate
	LAND USE CONTROLS				\$5,000	Recent Similar Project
	SUBTOTAL				\$16,600	:
	CONTINGENCY	15%	of	\$16,600	\$2,490	
	TOTAL ANNUAL OPERATING COST				\$19,090	
PERIODIC	COST		7-			-
FERIODIC				LINIT		
•	DESCRIPTION	OTY	UNIT	COST	TOTAL	NOTES
	5 YEAR REVIEWS				\$6,000	Recent Similar Project
	CONTINGENCY	15%	of	\$6,000	\$900	
	TOTAL PERIODIC OPERATING COST				\$6,900	
PRESENT	VALUE ANALYSIS				•	Based on 3.1% Discount Rate

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